

Differences in methane production, storage and transport  
among plant community types during a wet summer at Mer  
Bleue bog, Ottawa

by

Elizabeth Cowan

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## ABSTRACT

This study assessed indicators of methane (CH<sub>4</sub>) production, storage and transport among plant community types to investigate the spatial and temporal variability in CH<sub>4</sub> dynamics during a wet summer at the Mer Bleue bog. Community type was largely differentiated by sedge (*Eriophorum vaginatum*) presence, which increased as the average water table depth approached the surface. Stable isotope analysis identified CO<sub>2</sub> reduction as the dominant methanogenic pathway in all plots, but with more depleted source signatures as percent *Eriophorum* increased, possibly due to CH<sub>4</sub> transport via aerenchymatous tissues. Increasing peat temperature contributed to increasing CH<sub>4</sub> concentrations at 50 cm, near the long-term average water table, and dissolved organic carbon and total dissolved nitrogen at all depths, and a seasonal increase in chamber CH<sub>4</sub> emissions. Plant community did not relate to spatial differences in pore water characteristics, but increasing percent *Eriophorum* was associated with greater CH<sub>4</sub> emissions.

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## LIST OF SYMBOLS AND ABBREVIATIONS

Symbol	Units	Definition
$A$	$\text{m}^2$	chamber surface area
$Abs_{254}$	$\text{m}^{-1}$	absorbance at 254 nm
$c_a$	$\mu\text{mol mol}^{-1}$	ecosystem atmospheric $\text{CH}_4$ concentration
$c_b$	$\mu\text{mol mol}^{-1}$	background $\text{CH}_4$ concentration
$c_s$	$\mu\text{mol mol}^{-1}$	source $\text{CH}_4$ concentration
$C$		carbon
$C-A$		<i>Chamaedaphne</i> automatic chamber
$\text{CH}_4$		methane
$\text{CO}_2$		carbon dioxide
$C/N$		carbon to nitrogen ratio
$CRDS$		cavity ring-down spectroscopy
$\delta^{13}\text{C}_a$	‰	isotopic composition of the sampled $\text{CH}_4$
$\delta^{13}\text{C}_b$	‰	isotopic composition of the background $\text{CH}_4$
$\delta^{13}\text{C}_s$	‰	isotopic composition of the $\text{CH}_4$ source component
$\delta^{13}\text{C}-\text{CH}_4$	‰	ratio of $^{13}\text{CH}_4$ to $^{12}\text{CH}_4$ relative to V-PDB marine carbonate standard
$dC/dt$	$\mu\text{mol mol}^{-1}$ dry air $\text{s}^{-1}$	rate of change in $\text{CH}_4$ mixing ratio
$\text{DOC}$	$\text{mg L}^{-1}$	dissolved organic carbon
$\text{DOY}$		day of year
$E-M$		<i>Eriophorum</i> manual chamber
$ES$		early summer (DOY 155-189)
$EV$		<i>Eriophorum vaginatum</i>

$EV\%$	%	% LAI attributed to <i>Eriophorum vaginatum</i>
$F_{CH_4}$	$\text{mg m}^{-2} \text{ h}^{-1}$	$\text{CH}_4$ flux
$H_2$		hydrogen
$L-A$		<i>Ledum</i> automatic chamber
$LAI$	$\text{m}^2 \text{ m}^{-2}$	leaf area index
$LAI-2000$	$\text{m}^2 \text{ m}^{-2}$	total biomass LAI as measured with LAI-2000
$LS$		late summer (DOY 222-255)
$LSp$		late spring (DOY 132-154)
$MS$		mid-summer (DOY 190-221)
$N$		nitrogen
$NEE$	$\mu\text{mol m}^{-2} \text{ s}^{-1}$	net ecosystem exchange of $\text{CO}_2$
$O_2$		oxygen
$OM$		organic matter
$\rho$	Pa	air pressure
$PVC$		polyvinyl chloride
$PW_{CH_4}$	$\mu\text{mol L}^{-1}$	pore water $\text{CH}_4$ concentration
$R^*$	$8.314 \text{ J K}^{-1} \text{ mol}^{-1}$	ideal gas constant
$sccm$	sccm	standard cubic centimeter per minute
$S_{CH_4}$	$\text{mg m}^{-2} \text{ h}^{-1}$	storage change of pore water $\text{CH}_4$
$SUVA_{254}$	$\text{L m}^{-1} \text{ mg}^{-1} \text{ DOC}$	UV absorbance at 254 nm standardized to DOC
$T_{air}$	$^{\circ}\text{C}$	air temperature measured at tower site
$T_{tower}$	$^{\circ}\text{C}$	peat temperature measured at tower site
$T_{10cm}$	$^{\circ}\text{C}$	peat temperature at 10 cm depth (chamber-specific)
$T_{50cm}$	$^{\circ}\text{C}$	peat temperature at 50 cm depth at tower location
$TDN$	$\text{mg L}^{-1}$	total dissolved nitrogen

$Tg$	$10^{12}$ g	teragram
$V$	$m^3$	chamber volume
$WT$	cm	water table

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## 1.0 INTRODUCTION

Wetlands have been identified as particularly important ecosystems for global carbon (C) dynamics (Bridgham *et al.* 2006). In particular, northern peatland ecosystems have sequestered 76-96 Tg C a<sup>-1</sup> through the uptake of carbon dioxide (CO<sub>2</sub>) over the long term (Gorham 1991, Gorham 1995). However, they also emit approximately 38 Tg methane (CH<sub>4</sub>) a<sup>-1</sup>, which has a higher potential for climate forcing over the short term in comparison to CO<sub>2</sub> (Bartlett & Harriss 1993). Considering the variable atmospheric growth rate over the past 3 decades, including a recent increase in the growth rate of CH<sub>4</sub> concentrations since 2007 (Kirschke *et al.* 2013), there is a current need to examine the potential feedbacks and responses of peatland CH<sub>4</sub> emissions to climate change.

Atmospheric CH<sub>4</sub> depends on a number of sources and sinks, with wetland emissions of CH<sub>4</sub> representing the largest natural emitter. Increased understanding of the processes that lead to CH<sub>4</sub> loss in wetlands such as peatlands is crucial for the development of a global CH<sub>4</sub> budget that accurately captures CH<sub>4</sub> source and sink estimates (Mikaloff Fletcher *et al.* 2004a, Chanton *et al.* 2005, Kirschke *et al.* 2013).

Climate change projections indicate a multitude of possible responses by northern peatlands (Bridgham *et al.* 2013). For example, vascular plants (especially deciduous shrubs and graminoids) are expected to dominate over mosses and lichens as temperatures and growing season length increases (Walker *et al.* 2006, IPCC 2007, Gallego-Sala & Prentice 2013). The composition of the peatland plant community, which reflects environmental variables such as nutrient availability and long-term water table (WT) depths, is often used as an indicator of the potential magnitude of CH<sub>4</sub> emissions among and within peatlands. The dominant vegetative communities of northern peatlands



also directly impact CH<sub>4</sub> exchange by influencing substrate availability for methane-producing microorganisms and gas transport above and below the WT via aerenchymatous tissues (Couwenberg *et al.* 2011, Bridgham *et al.* 2013, Ward *et al.* 2013).

While vegetative community has been identified as an important ecosystem response indicator, there remains a need for increased understanding of the complex relation between plant community composition and CH<sub>4</sub> emission. This study seeks to couple the use of autochamber CH<sub>4</sub> flux measurements with pore water and stable carbon isotope analyses to achieve a better understanding of CH<sub>4</sub> production, storage and transport pathways for different vegetative communities along depth profiles at a temperate ombrotrophic bog. This will aid in determining the relative importance of dominant environmental controls, including vegetation type, on CH<sub>4</sub> emission dynamics.

Three key research hypotheses follow from this central research aim:

1. Spatial and temporal variability in CH<sub>4</sub> emissions will vary with plant community characteristics, where sedge-dominated plots are expected to have greater emissions than plots dominated by hummock species. This hypothesis follows findings by Lai *et al.* (2014a) and is based on the notion that CH<sub>4</sub> production, storage and transport will all be impacted by the abiotic and biotic variables that are associated with the different plant community types. These include, for example, the average WT depth and the presence of vegetation with aerenchymatous tissues.

2. As the growing season progresses, increasing peat temperature and plant growth will be associated with changes in belowground pore water characteristics that suggest increased substrate availability and increased CH<sub>4</sub> production rates, especially with reducing conditions in the saturated zone at a depth close to the long-term average WT. This hypothesis is based on previous laboratory studies that have identified these environmental factors as important controls on methanogenic metabolism (Dunfield *et al.* 1993, Moore & Dalva 1993, Yavitt & Seidman-Zager 2006).
3. The seasonal increase in CH<sub>4</sub> storage will be lowest at chamber plots that are associated with the highest seasonal CH<sub>4</sub> emissions, as these emissions represent a loss of stored CH<sub>4</sub> from peat. This follows previous evidence of a decoupling between CH<sub>4</sub> production and emission to the atmosphere due to transport limitations (i.e. diffusive transport of CH<sub>4</sub> dominates over plant-mediated transport) (Moore & Dalva 1993, Blodau 2002, Brown *et al.* 2014).

## 2.0 BACKGROUND

### 2.1 Northern peatlands and climate

According to a digital, high-resolution database of wetland sites developed by Matthews & Fung (1987), wetlands were historically estimated to cover a global area of approximately  $5.3 \times 10^{12} \text{ m}^2$  or roughly 3.6% of the global terrestrial surface. More recent wetland inventory analyses suggest this estimate may be up to 2 times greater (Bridgham *et al.* 2013). The total northern wetland area used in global  $\text{CH}_4$  emission studies varies widely from  $2.6 \times 10^{12}$  to  $9.0 \times 10^{12} \text{ m}^2$  (Petrescu *et al.* 2010) and is a major source of uncertainty when assessing the global  $\text{CH}_4$  budget (Kirschke *et al.* 2013). While the distribution of these wetlands tends to be scattered and the areas small, they represent an important component in the global C cycle (Matthews & Fung 1987, Bridgham *et al.* 2013). Among these wetlands are northern peatlands where organic matter (OM) has accumulated as peat to a depth of 40 cm or more (Gorham 1995). These northern peatland ecosystems cover an area of approximately  $3.5 \times 10^{12} \text{ m}^2$  and have a mean peat depth of ~2.3 m (Gorham 1991). They are estimated to contain between 273 and  $455 \times 10^{15} \text{ g}$  of C, 98.5% of which is in the form of peat (Gorham 1991, Turunen *et al.* 2002). This large range in C storage estimation is attributed to uncertainty surrounding global peat depths and area (Gorham 1991).

Northern peatlands are currently or have been net sinks for C because net primary production (i.e. photosynthetic uptake of carbon dioxide,  $\text{CO}_2$ ) occurs at a greater rate than the decomposition of OM, resulting in the accumulation of peat (Gorham 1991, Gorham 1995, Moore *et al.* 1998). However, the persistent oxygen ( $\text{O}_2$ )-depleted

conditions in wetlands are also favourable for the production of CH<sub>4</sub>, with natural wetlands representing the largest natural CH<sub>4</sub> source and contributing ~30% of global CH<sub>4</sub> emissions (Bridgman *et al.* 2013, Kirschke *et al.* 2013). Bartlett & Harriss (1993) estimated that wetlands north of 45°N release up to 38 Tg CH<sub>4</sub> a<sup>-1</sup> while globally, wetlands may emit 55 to 231 Tg CH<sub>4</sub> a<sup>-1</sup> (Houweling *et al.* 2000, Wuebbles & Hayhoe 2002, Neef *et al.* 2010). For northern peatlands, the global long-term rate of net atmospheric CO<sub>2</sub> sequestration ranges from 76-96 Tg C a<sup>-1</sup> over the past 4600 years (Gorham 1991, Gorham 1995). At individual site level, the mean 6-year CO<sub>2</sub> exchange at the Mer Bleue temperate ombrotrophic bog was  $-40.2 \pm 40.5$  g C-CO<sub>2</sub> m<sup>-2</sup> a<sup>-1</sup> (where negative values indicate a net uptake by the peatland), while CH<sub>4</sub> exchange and dissolved organic C (DOC) export were  $3.7 \pm 0.5$  g C-CH<sub>4</sub> m<sup>-2</sup> a<sup>-1</sup> and  $14.9 \pm 3.1$  g C-DOC m<sup>-2</sup> a<sup>-1</sup> respectively. While the CH<sub>4</sub> exchange component at Mer Bleue is small in comparison to DOC export, many peatlands have a seasonal CH<sub>4</sub> flux value in the same range as DOC loss (Roulet *et al.* 2007). However, while characterized by significant spatial and temporal variability, CO<sub>2</sub> uptake usually offsets CH<sub>4</sub> emissions in the long-term (i.e. several centuries) in wetlands (Petrescu *et al.* 2015). For example, at the Stordalen palsamire complex in subarctic Sweden, average annual CH<sub>4</sub> emissions were 18-22 g C-CH<sub>4</sub> m<sup>-2</sup> while annual CO<sub>2</sub> uptake was 46 g C-CO<sub>2</sub> m<sup>-2</sup> over 8 years (2001-2008) of micrometeorological monitoring (Christensen *et al.* 2012). At the Degerö Stormyr boreal minerogenic oligotrophic mire complex in northern Sweden, annual CH<sub>4</sub> emissions and CO<sub>2</sub> uptake were 9 g C-CH<sub>4</sub> m<sup>-2</sup> and 55 g C-CO<sub>2</sub> m<sup>-2</sup>, respectively, during 2004 (Nilsson *et al.* 2008). Methane emissions from the western peatland flux station northeast of Athabasca, Alberta ( $2.4$  g C-CH<sub>4</sub> m<sup>-2</sup>; Long *et al.* 2010) are much less than the average

annual uptake of CO<sub>2</sub> of 144 g C-CO<sub>2</sub> m<sup>-2</sup> (Syed *et al.* 2006). At a fen area with continuous permafrost in northeastern Greenland, CH<sub>4</sub> emissions were also less than CO<sub>2</sub> uptake (2.8 g C-CH<sub>4</sub> m<sup>-2</sup> vs. 48.7 g C-CO<sub>2</sub> m<sup>-2</sup>, over 87 and 62 days respectively; Friborg *et al.* 2000, Soegaard *et al.* 2000). Although CH<sub>4</sub> emissions may be much less than CO<sub>2</sub> sequestration and CH<sub>4</sub> is short-lived in the atmosphere, it is more effective at absorbing incoming infrared radiation and is thus a more potent greenhouse gas than CO<sub>2</sub> (Bartlett & Harriss 1993). According to the Intergovernmental Panel on Climate Change (2013), CH<sub>4</sub> has 86 times the relative cumulative forcing index as CO<sub>2</sub> on a mass basis over a 20-year period. Although characterized by large inter-annual variability, growth of CH<sub>4</sub> concentrations in the atmosphere slowed and stabilized during the mid-1990's, remaining relatively constant until 2006 (Bridgham *et al.* 2013). Since 2007, however, atmospheric CH<sub>4</sub> concentrations have been on the rise with an average atmospheric concentration of 1.80 ppm in 2010 (Kirschke *et al.* 2013). The variable atmospheric growth rate of CH<sub>4</sub> over recent decades signals a need for increased understanding of the factors influencing natural peatland CH<sub>4</sub> emissions. Yet, the controls over CH<sub>4</sub> production, consumption and transport are poorly represented in current CH<sub>4</sub> biogeochemistry models, largely due to a limited number of CH<sub>4</sub> flux datasets with which these models may be improved and validated (Bridgham *et al.* 2013, Kirschke *et al.* 2013).

Peatlands are often neglected in global climate change models despite their importance for global C cycling, as there exist large uncertainties surrounding wetland controls on C fluxes, such as wetland specific plant functional types (Moore *et al.* 1998, IPCC 2013). Dissolved organic carbon (DOC) export, CO<sub>2</sub> exchange, CH<sub>4</sub> exchange, and C storage constitute the major components of peatland C cycles and are primarily

controlled by air and substrate temperature, pH, plant community structure, substrate availability/quality and reduction/oxidation (i.e. redox) boundaries associated with WT position (Moore *et al.* 1998, Blodau 2001).

### **2.1.1 *CH<sub>4</sub>* production**

According to Klass (1984), biogenic CH<sub>4</sub> production requires modest temperatures, ambient pressures, fully saturated conditions and the absence of other free-energy electron acceptors (e.g. O<sub>2</sub>, nitrate, sulphate). The production of CH<sub>4</sub> is directly dependent on microbial activity, thus an organic C substrate and mixed population of CH<sub>4</sub>-producing microorganisms (i.e. methanogens) is also necessary (Klass 1984, Clark & Fritz 1997). This organic C substrate serves as a food source to support methanogens, which require small, simple dissolved organic molecules with molecular weights < 600 g mol<sup>-1</sup> (Fenchel *et al.* 1998, Blodau 2001). These molecules are produced when fermentive bacteria reduce complex organic molecules in the peat to simpler molecules like fatty acids, which are then transformed by acetogenic bacteria to acetate with CO<sub>2</sub> and hydrogen (H<sub>2</sub>) as byproducts (Equation 1). Organic-rich environments thus facilitate reactant supply (i.e. acetate, H<sub>2</sub> and CO<sub>2</sub>) for methanogenic pathways (Clark & Fritz 1997).



Recent <sup>14</sup>C-labelling studies suggest that root exudates (i.e. plant photosynthate) serve as organic C substrate for CH<sub>4</sub> production in the rhizosphere. These root exudates also indirectly increase CH<sub>4</sub> production through priming effects, where they trigger the

decomposition of less labile soil OM (Bridgham *et al.* 2013). Additionally, methanogens require mineral nutrients, a reducing agent and a source of nitrogen (i.e.  $\text{NH}_4^+$ ). As obligate anaerobes, methanogens cannot tolerate  $\text{O}_2$  and are most active under strongly reducing conditions of less than -200 mV as demonstrated in Table 2-1 (Zhi-Guang 1985, Mansfeldt 2003, Vorenhout *et al.* 2004).

Table 2-1: Redox pairs and redox values (Eh) at transformation at reference pH of 7.0 (Adapted from Vorenhout *et al.* 2004)

	<i>Oxidized form</i>	<i>Reduced form</i>	<i>Eh at transformation (mV)</i>
Oxygen	$\text{O}_2$	$\text{H}_2\text{O}$	+600 to +400
Nitrogen	$\text{NO}_3^-$	$\text{N}_2\text{O}$ , $\text{N}_2$ , $\text{NH}_4^+$	250
Manganese	$\text{Mn}^{4+}$	$\text{Mn}^{2+}$	225
Iron	$\text{Fe}^{3+}$	$\text{Fe}^{2+}$	+100 to -100
Sulfur	$\text{SO}_4^{2-}$	$\text{S}^{2-}$	-100 to -200
Carbon	$\text{CO}_2$	$\text{CH}_4$	< -200

An understanding of the  $\text{CH}_4$  production pathways remains limited (Galand *et al.* 2010). It has been established that the decomposition of OM by a form of anaerobic respiration in methanogenic *Archea* produces  $\text{CH}_4$  as an ultimate end product (Conrad 2005). Methane is primarily produced in the upper portion of saturated peat and/or sedimentary environments, where  $\text{O}_2$  is absent, yet methanogens still have access to sufficient organic substrate (Chanton *et al.* 2005). The two main  $\text{CH}_4$  production pathways in terrestrial freshwater systems include acetoclastic methanogenesis (Equation 2) and hydrogenotrophic methanogenesis (Equation 3) (Chanton *et al.* 2005).



The fermentation of acetate (acetoclastic methanogenesis, Equation 2) occurs when acetic acid ( $CH_3COOH$ ) acts as the terminal electron acceptor (i.e. food source) resulting in the production of  $CH_4$  and  $CO_2$  gas (Clark & Fritz 1997). While acetoclastic methanogenesis can play an important role in both ombrotrophic and minerotrophic peatlands, only a few species of methanogens can use acetate as a substrate (Avery *et al.* 1999, Kotsyurbenko *et al.* 2004, Kotsyurbenko *et al.* 2007, Galand *et al.* 2010). Thus,  $CO_2$  reduction with  $H_2$  gas to form  $CH_4$  (hydrogenotrophic methanogenesis, Equation 3) is typically viewed as the dominant pathway of peatland  $CH_4$  production. The methanogens required for this process are thought to be present in differing levels of diversity in all peatlands (Kotsyurbenko *et al.* 2007). While these two mechanisms have been identified as the main drivers of methanogenesis in most environments, identifying their relative contributions and how each pathway is controlled by environmental factors remains a challenge (Conrad 2005, Kotsyurbenko *et al.* 2007, Galand *et al.* 2010). For example, freshwater wetlands seem to be heavily influenced by vegetation: *Sphagnum*-moss-vegetated bogs produce  $CH_4$  mainly by  $CO_2$  reduction while fens dominated by *Carex* sedges rely on acetate fermentation (Chanton *et al.* 2005). Additionally, stable isotope measurements from pore water depth profiles obtained at two temperate wetlands show that the upper peat layers (rich in labile organic C) are favourable for acetoclastic methanogenesis, while  $CO_2$  reduction is increasingly important as peat becomes more recalcitrant with depth (Hornibrook *et al.* 1997).



### 2.1.2 *CH<sub>4</sub> consumption*

As the CH<sub>4</sub> produced in the anaerobic saturated zone diffuses up through the aerobic peat layer along a concentration gradient, CH<sub>4</sub>-consuming microbes (i.e. methanotrophs) reoxidize CH<sub>4</sub> to methanol (CH<sub>3</sub>OH), formaldehyde (CH<sub>2</sub>O), formate (CHOO<sup>-</sup>) then CO<sub>2</sub> (Whalen 2005). The main CH<sub>4</sub> consumption types include low and high affinity oxidation, which occur at high (>40 ppm) and close to atmospheric (<12 ppm) CH<sub>4</sub> concentrations respectively (Segers 1998, Le Mer & Roger 2001). In northern peatlands, low affinity oxidation occurring at high CH<sub>4</sub> concentrations is the dominant CH<sub>4</sub> consumption process (Bender & Conrad 1995). The CH<sub>4</sub> is used as a C and energy source by methanotrophs (Le Mer & Roger 2001). For CH<sub>4</sub> oxidation to proceed, the presence of O<sub>2</sub> is key as it acts as an electron acceptor for the formation of CO<sub>2</sub>. Oxidizing conditions >400 mV indicate the predominance of O<sub>2</sub> in a substrate, and are thus associated with maximum methanotrophic activity (Zhi-Guang 1985, Mansfeldt 2003).

As sufficient O<sub>2</sub> is present in the aerobic zone (i.e. above WT) and maximum CH<sub>4</sub> substrate occurs in the anaerobic zone (i.e. below WT), there is some evidence that CH<sub>4</sub> consumption peaks within ~25 cm of the aerobic/anaerobic interface or average WT position before CH<sub>4</sub> concentrations are too low to support methanotrophs (Segers 1998). Methanotrophs could potentially form hotspots of CH<sub>4</sub> oxidation below the average WT position if sufficient O<sub>2</sub> reaches these depths through plant transport. For example, C deprivation experiments have demonstrated that methanotrophic bacteria persist under anaerobic conditions, suggesting that methanotrophs can enter an anaerobic dormant state and significantly attenuate internal metabolism (Roslev & King 1994, Roslev & King

1995). Thus, when aerenchymatous tissues transport sufficient O<sub>2</sub> below the WT, methanotrophic activity can proceed through the oxidation of these high CH<sub>4</sub> concentrations in the saturated zone.

### **2.1.3 CH<sub>4</sub> transport**

Molecular diffusion and ebullition (i.e. sudden release of CH<sub>4</sub> bubbles) through the peat and diffusion through plant aerenchyma (i.e. soft plant tissue containing air spaces) comprise the three main pathways of CH<sub>4</sub> emission to the atmosphere (Figure 2-1). The diffusion pathway is driven by a CH<sub>4</sub> concentration gradient where CH<sub>4</sub> concentrations decrease traveling up the peat profile from the zone of CH<sub>4</sub> production (i.e. deeper anaerobic peat layers) to the atmosphere (Blodau *et al.* 2007, Dorodnikov *et al.* 2013). While diffusive transport is the slowest of the three pathways (especially in the saturated anaerobic peat layers), it serves the important role of transporting CH<sub>4</sub> through the aerobic upper peat layer where microbial CH<sub>4</sub> consumption through oxidative processes can occur (Whalen 2005).

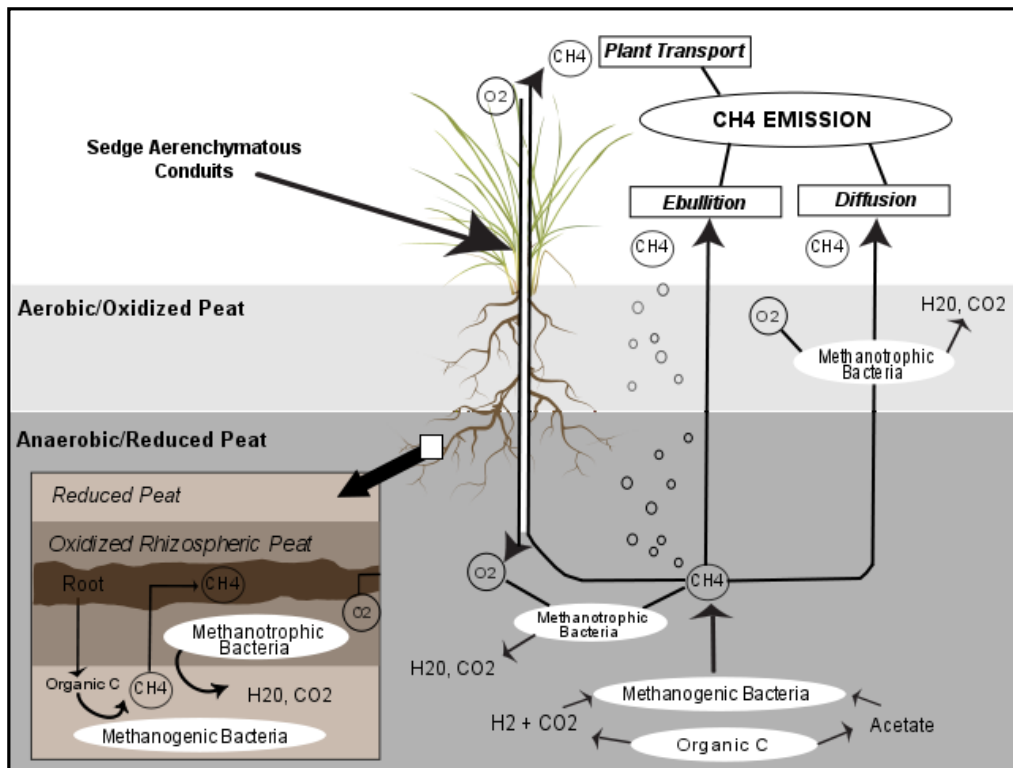


Figure 2-1: Methane production, consumption and transport pathways in peat layers (Adapted from Le Mer & Roger 2001).

The atmospheric release of free-phase  $\text{CH}_4$  in gas bubbles, known as ebullition, can be episodic in nature where it is initiated by pore water  $\text{CH}_4$  supersaturation at depth (i.e. in anaerobic zone of  $\text{CH}_4$  production). Bubbles form when the hydrostatic pressure in peat is exceeded by the partial pressure of all dissolved gases in solution (Chanton & Whiting 1995). These  $\text{CH}_4$  bubbles are not immediately released to the atmosphere; rather, the gas bubbles accumulate in peat pores until overpressure zones develop in peat and the bubbles move towards the peat surface (Kellner *et al.* 2004). Slow ebullition, where bubbles are released at a steady rate, typically results in a large proportion of bubble  $\text{CH}_4$  being consumed through oxidation in the peat column above the WT (Rosenberry *et al.* 2006). Episodic ebullition events, however, involve much higher rates of degassing. These sudden bubbling events can overwhelm  $\text{CH}_4$  oxidation potential by

methanotrophs in the aerobic peat layer and release much greater volumes of CH<sub>4</sub> than slow ebullition (Rosenberry *et al.* 2006). The rupture of a peat fabric/layer may create a preferential flow path thereby enabling such large volumes of gas release through episodic ebullition. For example, ebullition has been shown to release 40-53 g CH<sub>4</sub> m<sup>-2</sup> per event at the Red Lake peatland, northern Minnesota (Glaser *et al.* 2004). Falling atmospheric pressure has also been associated with episodic ebullition events, with results by Tokida *et al.* (2007) indicating an increase in CH<sub>4</sub> flux by 2 orders of magnitude over short time scales (<2 hours). These large releases of CH<sub>4</sub> significantly contribute to total CH<sub>4</sub> flux in some northern peatlands (Tokida *et al.* 2007).

The third transport pathway is facilitated by the aerenchymatous tissues of some vascular peatland plants such as sedges, which function as conduits for CH<sub>4</sub> gas transport from the roots to the atmosphere. Methane travels up through the aerenchyma as a result of molecular diffusion along the concentration gradient and bulk flow due to pressure differences between internal plant air spaces and the atmosphere (Lai 2009). While transport capabilities are species specific, this relatively fast pathway reduces the residence time of CH<sub>4</sub> in the aerobic zone and thus limits CH<sub>4</sub> oxidation potential in upper peat layers (Blodau 2001, Strack *et al.* 2006). Sedges in particular often have roots that reach below the WT depth, with sedge root biomass peaking close to 50 cm below peat surface and extending as deep as 1 m in a temperate bog where the average WT is ~42 cm below the surface (Moore *et al.* 2002, Murphy & Moore 2010). These roots add labile C and other substrates to the anaerobic zone (Strack *et al.* 2006). In the opposite direction, aerenchyma provide a conduit for the diffusion of atmospheric O<sub>2</sub> to the rhizosphere below the aerobic/anaerobic boundary (i.e. radial oxygen loss), creating

aerated volumes or oxic “pockets” where CH<sub>4</sub> oxidation could occur at depth (Vasander & Kettunen 2006). Thus, a larger sedge root biomass may cause increased CH<sub>4</sub> consumption by enabling O<sub>2</sub> transport and associated reoxidation of CH<sub>4</sub> in the rhizosphere below the WT (Blodau 2001, Strack *et al.* 2006). However, there is some evidence suggesting that not all aerenchymatous plant species provide the opportunity for CH<sub>4</sub> oxidation at depth. For example, Frenzel and Rudolph (1998) could not determine any significant CH<sub>4</sub> oxidation at depth for *Eriophorum* species. They concluded that this species was associated with low CH<sub>4</sub> oxidation rates and did not reduce CH<sub>4</sub> emissions. This could result from plant influence on biogeochemistry at depth, where different species produce root exudates of varying type and quality (Frenzel & Rudolph 1998).

## **2.2 Environmental controls on peatland CH<sub>4</sub> emissions**

### **2.2.1 Redox boundary associated with WT position**

Methane emissions can vary spatially and temporally at fine scales as numerous interactive environmental factors affect biologically mediated CH<sub>4</sub> emissions from soils (Bartlett & Harriss 1993, Blodau 2001, Chanton *et al.* 2005). The microtopography of peatlands is typically divided into horizontal microstructure levels that have characteristic WT positions and vegetative communities (Strack *et al.* 2008, Rydin & Jeglum 2013). For example, hummock-hollow microtopography is common in ombrotrophic bogs. Hummocks are characterized by dwarf shrubs and tend to be 20-50 cm above the lowest surface level (Rydin & Jeglum 2013). Hollows refer to the depressions between hummocks, which can be a combination of lawns, carpets and mud-bottoms where graminoids and/or mosses dominate and the WT is closer to the peat surface (Rydin &

Jeglum 2013). The microtopographic differences in vegetative community and associated depth to WT influence CH<sub>4</sub> dynamics and contribute to its spatial variability. These processes are also governed by the vertical structuring of peatlands, which is intricately linked with WT position. Aerobic conditions (presence of O<sub>2</sub>) occur above the WT in the *acrotelm* or unsaturated upper peat layer (top ~5-40 cm). Beneath this layer, the *catotelm* is characterized by saturated, anaerobic conditions (depletion of O<sub>2</sub>) (Rydin & Jeglum 2013).

The boundary between aerobic and anaerobic conditions shifts according to the position of the WT, inducing both biotic and abiotic changes (Limpens *et al.* 2008, Rydin & Jeglum 2013). Such changes could include alterations in dominant vegetation types. For example, Bubier *et al.* (2006) found that WT, in conjunction with water chemistry, was an important control on vegetation distribution at Mer Bleue, a temperate bog. Their results showed that most dominant ericaceous shrubs, including *Vaccinium myrtilloides*, *Ledum groenlandicum* and *Chamaedaphne calyculata*, favoured drier microtopographic positions while *Kalmia angustifolia* and *Eriophorum vaginatum* had higher biomass with WT closer to the peat surface (Bubier *et al.* 2006). Shifts in WT position could also induce variations in decomposition rates and processes (Rydin & Jeglum 2013). This hydrological control (i.e. WT position) defines the boundary between reduction and oxidation zones in the soil profile and is thus identified as a critical factor for CH<sub>4</sub> production in many CH<sub>4</sub> flux studies within peatlands (e.g. Turetsky *et al.* 2008, Lai *et al.* 2014a, Lai *et al.* 2014b) and among peatlands (e.g. Turetsky *et al.* 2014). Theoretically, a WT closer to the peat surface would result in larger CH<sub>4</sub> emissions as the saturated CH<sub>4</sub> production zone increases while the unsaturated CH<sub>4</sub> consumption zone shrinks. For

example, Turetsky *et al.* (2008) manipulated WT position at an ecosystem-scale in an Alaskan peatland and found that a raised WT resulted in the largest CH<sub>4</sub> fluxes. However, a study at a poor fen site demonstrated that a lowered WT led to increased biomass in hollows and CH<sub>4</sub> fluxes remained relatively high despite the thicker aerobic peat layer (Strack *et al.* 2008). This suggests that increased biomass could offset lowered WT effects through substrate addition and enhanced plant-mediated CH<sub>4</sub> transport at peatland hollows over a time scale of 2-5 years, highlighting the importance of climate change feedbacks in relation to vegetation and WT position (Strack *et al.* 2008).

At short time scales (i.e. weeks to months), the interaction between CH<sub>4</sub> production, consumption and transport processes may not be at equilibrium with varying environmental controls, such as WT position (Blodau 2002). It has been suggested that rates of CH<sub>4</sub> production in peat are decoupled from atmospheric fluxes to some extent (Blodau 2002). Changes in storage of CH<sub>4</sub> within the peat pore water would result. For example, Moore *et al.* (1990) observed large negative changes in stored CH<sub>4</sub> in peat pore water (260 to <90  $\mu\text{mol L}^{-1}$ ) corresponding with increased CH<sub>4</sub> fluxes in subarctic/northern boreal fens. The largest mean daily CH<sub>4</sub> flux ( $262 \pm 205 \text{ mg m}^{-2} \text{ d}^{-1}$ ) on August 12, 1989, released 3-4 g CH<sub>4</sub> m<sup>-2</sup> from storage (Moore *et al.* 1990). Also, CH<sub>4</sub> production itself may not respond immediately to driving variables. Results from peat incubations have provided evidence for a time lag between the onset of saturated conditions and the initiation of CH<sub>4</sub> production processes (Öquist & Sundh 1998). Kettunen *et al.* (1999) found that a 1-week period of aerobic conditions followed by re-saturation of the peat profile produced low CH<sub>4</sub> fluxes due to a time lag between rising WT (i.e. re-saturation) and reactivation of CH<sub>4</sub> production processes. Similar results were

previously demonstrated through laboratory peat column experiments, where CH<sub>4</sub> emission rates did not reach significant levels until up to 10 days after the WT was raised to the surface of the column (Moore & Dalva 1993). Knorr *et al.* (2008) observed that draining (i.e. experimental drought) followed by rewetting delayed CH<sub>4</sub> production in temperate peat by weeks to months. Recent results indicate that dropping WT levels can reoxidize terminal electron acceptors in peat, thus suppressing CH<sub>4</sub> production processes after WT is raised until more favourable terminal electron acceptors have been consumed (Deppe *et al.* 2010b). The result is hysteresis in the relationship between CH<sub>4</sub> emissions and WT (Blodau & Moore 2003). The zone of WT fluctuation in particular exhibits a memory for CH<sub>4</sub> production rates based on past conditions. The history at a particular depth location thus affects CH<sub>4</sub> production rates when saturated anaerobic conditions commence (Blodau 2002).

### **2.2.2 Organic substrate supply**

Once the peat is saturated, CH<sub>4</sub> production becomes limited by the availability of organic substrate (Segers 1998). The compounds that comprise DOC (e.g. acetate, cell residues, humic/fulvic macromolecules) have a variety of sources and functions in peatlands (Blodau 2001). As previously detailed, annual DOC export can be in the same range as annual CH<sub>4</sub> emissions (Roulet *et al.* 2007). The production and supply of this DOC impacts the activity of methanogens. Simple compounds such as acetate, ethanol and glucose have been identified as the dominant substrates used by methanogens for CH<sub>4</sub> production, with ethanol being preferentially consumed in the first 12 hours after laboratory incubation of rice paddy sediment (Chawanakul *et al.* 2009, Kirstine & Galbally 2012). For fen peat samples, Coles and Yavitt (2002) demonstrated that



available substrates influence methanogenic activity. After adding acetate, ethanol or glucose, their results showed that net methanogenesis increased immediately following acetate addition, and within 24 hours after addition of ethanol and glucose, which must be degraded by fermentation into smaller compounds for uptake by methanogens (Coles & Yavitt 2002). As measurements were made immediately following substrate addition, the increase in CH<sub>4</sub> production cannot be attributed to population growth response, which requires more time, and thus provides evidence for substrate limitation of methanogenesis. The quantity of organic substrate was also highlighted as a limitation on CH<sub>4</sub> production through in vitro assays, where CH<sub>4</sub> production was enhanced by the addition of substrates to saturated peat originating from bog hollows (Yavitt & Seidman-Zager 2006).

The quality of the DOC supply can also influence CH<sub>4</sub> production rates. The quality of DOC is influenced by major environmental controls including WT position, peat temperature and surface vegetation (Lai 2009). Recent studies found that DOC for peatlands with a raised WT was more labile and less humified in comparison to lowered WT conditions. WT drawdown resulted in increased pore water DOC concentrations with higher aromaticity, possibly resulting from DOC leaching from increased plant production under lowered WT conditions (Moore *et al.* 1998, Hribljan 2012, Hribljan *et al.* 2014). A peat warming experiment in a poor fen in Michigan showed that pore water DOC increased in both lability and concentration for warmed plots (Kane *et al.* 2014). This is expected as the rate of temperature-dependent decomposition processes increase with warming (Kane *et al.* 2014). Furthermore, the quality of peatland DOC can be influenced by surface vegetation characteristics, as acid-insoluble OM is typical of shrub-

dominated sites and has been shown to be negatively correlated with CH<sub>4</sub> production in peat from Appalachian wetland sites (Yavitt & Lang 1990). Similarly, in a substrate manipulation experiment with incubated peat samples collected from five different sites at Storåmyren, Sweden, (minerotrophic lawn, ombrotrophic lawn, ombrotrophic wet carpet, mud-bottom, ombrotrophic hummock), Bergman *et al.* (2000) suggested that easily degradable substrate availability, as well as temperature, influenced CH<sub>4</sub> production rates within sites. Chanton *et al.* (2008) examined peat from a variety of northern peatland sites (Minnesota, Northern Alberta, and Alaska) and their results suggest that DOC from sedge-dominated peatlands is more labile than DOC from peatlands dominated by *Sphagnum* or other woody plants. DOC was relatively young in comparison to bulk peat for these peatlands regardless of surface vegetation, indicating that fresh and more recent plant C is important for DOC dynamics (Chanton *et al.* 2008). As DOC fuels CH<sub>4</sub> producing microbes in saturated peat layers, the quantity and quality of DOC substrate supply represent important environmental controls on peatland CH<sub>4</sub> emissions.

### **2.2.3 Peat temperature and acidity**

Northern peatland CH<sub>4</sub> production and emission rates increase with increasing peat temperature. Moore and Dalva (1993) demonstrated this through a laboratory column study of subarctic and temperate peat in which averaged CH<sub>4</sub> emissions at 23°C were 6.6 times greater than at 10°C. Temperature relations are commonly expressed using Q<sub>10</sub> values, which represent the change in reaction rate per 10°C change in temperature. For example, CH<sub>4</sub> emissions from peat monoliths (originating from Scottish Ellergower moss hollows) suggested methanogenesis Q<sub>10</sub> values ranging from 2.1-4.0

(Thomas *et al.* 1996). In an incubation analysis of peat slurries from temperate and subarctic peatlands, CH<sub>4</sub> production also demonstrated strong temperature dependence with Q<sub>10</sub> values ranging from 5.3 to 16 (Dunfield *et al.* 1993). Both CH<sub>4</sub> production and consumption processes were optimized at about 25°C in this study. However, consumptive CH<sub>4</sub> oxidation processes were not as sensitive to temperature, as indicated by lower Q<sub>10</sub> values of 1.4 to 2.1 (Dunfield *et al.* 1993).

Methane transport mechanisms can also be influenced by temperature. Ebullition CH<sub>4</sub> flux in particular can be moderated by temperature, where peat cooling events cause the contraction of ebullition bubbles and dissolution of CH<sub>4</sub> which limit ebullition over the short-term (Fechner-Levy & Hemond 1996). Increases in peat temperature could potentially increase CH<sub>4</sub> transport through ebullition as bubble volumes increase and CH<sub>4</sub> gas accumulates (Fechner-Levy & Hemond 1996). Plant CH<sub>4</sub> transport can also be enhanced with warmer temperatures as pressurized ventilation in the aerenchyma increases and subsequently allows for greater flow of CH<sub>4</sub> to the atmosphere (Große 1996). As such, numerous studies have reported the importance of peat temperature near the average position of the WT as an important control on CH<sub>4</sub> emissions (Bubier *et al.* 1995b, Moore *et al.* 2011, Lai *et al.* 2014a, Brown *et al.* 2014, Turetsky *et al.* 2014).

Peat acidity is also a control on CH<sub>4</sub> processes as the activities of methanogenic and methanotrophic organisms peak under specific optimum conditions. Methanogens have been shown to grow optimally at pH ranging from 6 to 8 (Garcia *et al.* 2000). Methanotrophs favour more acidic peat ranging from pH 4.3 to 5.9 (Kamal & Varma 2008). After incubating peat slurries, Dunfield *et al.* (1993) found that native peat pH was about 2 pH units lower than optimum pH ranges for CH<sub>4</sub> production and consumption in

temperate and subarctic peat (5.5-7.0 and 5.0-6.5 respectively). As *Sphagnum*-dominated ombrotrophic bogs such as Mer Bleue are rain-fed peatlands, they do not receive significant quantities of mineral bases derived from groundwater. Consequently, the acidic decomposition products remain unneutralized and the surface waters of bogs are characterized by an acidic pH of about 4 (Shotyk 1988). As the pH of ombrotrophic bogs lies outside the optimum ranges for CH<sub>4</sub> production and consumption, the acidity of these peatlands may limit the activity of both methanogens and methanotrophs.

#### **2.2.4 Vegetation community structure**

As northern latitude peatlands are expected to experience higher temperatures and longer growing seasons, predictions suggest vascular plants will out-compete bryophytes and lichens (Walker *et al.* 2006, IPCC 2007, Gallego-Sala & Prentice 2013). Warming experiments have demonstrated the potential for deciduous shrubs and graminoids to increase in height and cover, while mosses and lichens decrease in cover due to shade intolerance (Walker *et al.* 2006).

Previous studies have demonstrated that vegetative community composition is a good indicator of CH<sub>4</sub> exchange within and among peatlands (Bubier 1995, Bubier *et al.* 1995a, Dias *et al.* 2010, Couwenberg *et al.* 2011, Levy *et al.* 2012, Gallego-Sala & Prentice 2013, Ward *et al.* 2013). The composition of the plant community not only reflects environmental variables such as nutrient availability and long-term WT depths, but also directly impacts CH<sub>4</sub> exchange by supplying root exudates, increasing available substrates to methanogens, regulating peat moisture and influencing transport through the CH<sub>4</sub> consumption zone (Couwenberg *et al.* 2011, Bridgham *et al.* 2013, Ward *et al.*

2013). For instance, Ward *et al.* (2013) found that vegetative community composition, particularly graminoid presence, was more important than a 1°C increase in air temperature for regulating CH<sub>4</sub> flux within an ombrotrophic blanket bog in northern England. Similarly, a CH<sub>4</sub> flux model for data from 21 sites in the United Kingdom showed that the highest explanatory power stemmed from plant composition data (Levy *et al.* 2012). As previously detailed, plant communities dominated by sedges are associated with enhanced CH<sub>4</sub> emission presumably because sedges allow CH<sub>4</sub> produced at depth to bypass the consumption zone through aerenchymatous tissues (Couwenberg *et al.* 2011, Ward *et al.* 2013). Sedges also provide methanogens with an increased supply of acetate and other organic substrates (Ward *et al.* 2013). However, Strack *et al.* (2008) suggest that at sites with dense sedge cover (i.e. drained lawns), O<sub>2</sub> transported below the WT may enhance oxidation in the catotelm over the short term (2-5 years) and could reduce net CH<sub>4</sub> emissions.

## **2.3 Measurement techniques and approaches**

### **2.3.1 Autochamber CH<sub>4</sub> flux measurements**

Methane gas exchange between the surface and the atmosphere may be measured using a number of techniques including chambers and eddy covariance. The manual static non-steady state chamber method uses the increase in CH<sub>4</sub> concentration in the chamber headspace over time to calculate the CH<sub>4</sub> flux at the surface (Pihlatie *et al.* 2013). This method is ‘static’ in that there is no air flow circulating between the chamber and an analyzer. Instead, discrete air samples are collected over time manually. While chambers are portable and inexpensive, the CH<sub>4</sub> flux data from these chambers are limited both

spatially and temporally (e.g. fewer plots sampled at weekly to monthly intervals) due to practical constraints of manual sample collection (Lai *et al.* 2014a). In the eddy covariance technique, CH<sub>4</sub> fluxes are measured as the covariance of high frequency vertical wind velocity fluctuations and CH<sub>4</sub> mixing ratio fluctuations over a period of time such as 30 min with the appropriate air density and water vapour corrections applied (e.g. Brown *et al.* 2014). Eddy covariance CH<sub>4</sub> flux measurements represent the flux from an integrated area of roughly 1 ha and are thus difficult to interpret in relation to dominant vegetation types and common environmental variables that vary spatially (Lai *et al.* 2014a). These limitations highlight the value of automated chamber measurements, which achieve a higher temporal resolution than manual chamber measurements (e.g. one measured flux for each chamber every half hour) and increase the likelihood of observing elevated but transient CH<sub>4</sub> emission phenomena in specific plots representing the plant communities.

### **2.3.2 *Stable isotope analysis***

In conjunction with autochamber CH<sub>4</sub> flux measurements, stable C isotope analysis has the potential to reveal more detail into peatland CH<sub>4</sub> emission processes. Carbon exists as two stable isotopes in natural environments, <sup>12</sup>C and <sup>13</sup>C, with relative abundances of 98.9% and 1.1% respectively (Farquhar *et al.* 1989). Stable isotope fractionation occurs because heavier isotopes (i.e. <sup>13</sup>C) are cycled and transformed differently than lighter isotopes (i.e. <sup>12</sup>C) through biogeochemical processes including photosynthesis, respiration, and physical processes such as diffusion (Dawson & Siegwolf 2007). These processes alter the <sup>13</sup>C/<sup>12</sup>C composition of C-containing matter

and gases in the soil, biosphere and atmosphere, creating isotopic signatures that can be used to trace C transformation processes (Farquhar *et al.* 1989, Werner *et al.* 2012).

Stable C isotope analysis adopted the use of delta notation to express any sample's abundance of  $^{13}\text{C}$  in relation to  $^{12}\text{C}$  (Werner *et al.* 2012). According to Quay *et al.* (1991),  $\delta^{13}\text{C}$  is the ratio of  $^{13}\text{C}$  to  $^{12}\text{C}$  in parts per thousand, per mil (‰), relative to a marine carbonate standard, PeeDee Belemnite:

$$(4) \quad \delta^{13}\text{C} = \left( \frac{{}^{13}\text{C}/{}^{12}\text{C}_{SAM}}{{}^{13}\text{C}/{}^{12}\text{C}_{V-PDB}} - 1 \right) \times 1000$$

The PeeDee Belemnite international standard was based on a marine fossil found in South Carolina's Pee Dee formation, which had an unusually high ratio of  $^{13}\text{C}$  relative to  $^{12}\text{C}$  (Werner *et al.* 2012). PeeDee Belemnite was established as  $\delta^{13}\text{C}=0$  ‰, thus most natural samples have a negative  $\delta^{13}\text{C}$  value because they are more depleted in  $^{13}\text{C}$  than the standard. With increasing isotope studies and greater demand for use of the standard, the original PeeDee Belemnite had to be replaced with Vienna PeeDee Belemnite (V-PDB) (Werner *et al.* 2012).

The three main  $\text{CH}_4$  source categories have distinct isotopic signatures, with microbially produced  $\text{CH}_4$  having more depleted average values ( $\delta^{13}\text{C}-\text{CH}_4 = -60$  ‰) than fossil fuel  $\text{CH}_4$  ( $\delta^{13}\text{C}-\text{CH}_4 = -40$  ‰) or biomass burning  $\text{CH}_4$  ( $\delta^{13}\text{C}-\text{CH}_4 = -25$  ‰) (Quay *et al.* 1991, Miller 2005). The measured steady state atmospheric  $\delta^{13}\text{C}-\text{CH}_4$  value of  $-47$ ‰ is offset by  $6$ ‰ from the signature of atmospheric  $\text{CH}_4$  sources ( $\delta^{13}\text{C}-\text{CH}_4 = -53$ ‰) due to isotopic fractionation associated with global  $\text{CH}_4$  sinks (i.e. tropospheric OH, soils, and stratosphere) (Whiticar 2000, Miller 2005). Peatland  $\text{CH}_4$  emissions are dominated by

microbially produced CH<sub>4</sub> with  $\delta^{13}\text{C}\text{-CH}_4$  values ranging from -85 to -40‰ due to the influence of production, consumption, and transport processes (Hornibrook *et al.* 2000a). Peatland isotopic signatures also vary depending on the time of year. For example, Avery *et al.* (1999) found that  $\delta^{13}\text{C}\text{-CH}_4$  of pore water could shift by up to 11‰ annually at a Michigan peatland. This temporal variation was attributed to the shift of dominant CH<sub>4</sub> production pathway from CO<sub>2</sub> reduction (~-71‰) in the winter to acetate fermentation (-44‰) in the spring. The CH<sub>4</sub> production pathways can be distinguished by their  $\delta^{13}\text{C}\text{-CH}_4$  values as acetate fermentation yields a relatively <sup>13</sup>C-enriched signature (-65 to -50 ‰) compared to the more depleted CO<sub>2</sub> reduction signature (-110 to -60 ‰) (Whiticar *et al.* 1986). This is due to stronger methanogenic discrimination against <sup>13</sup>C during CO<sub>2</sub> reduction relative to acetate fermentation (Dorodnikov *et al.* 2013). Parent OM (i.e. C<sub>3</sub> vs. C<sub>4</sub> vegetation) will also influence  $\delta^{13}\text{C}\text{-CH}_4$  values (Chanton *et al.* 2005). An increased understanding of these CH<sub>4</sub> production pathways reveals information about decomposition patterns (i.e. fresh vs. old C) and the fate of C pools (i.e. pools with rapid turnover time vs. pools with longer turnover time) (Dorodnikov *et al.* 2013).

Methane consumption processes driven by methanotrophs cause shifts in <sup>13</sup>C isotopic signatures as well. Coleman *et al.* (1981) found that residual CH<sub>4</sub> from bacterial CH<sub>4</sub> oxidation became enriched in <sup>13</sup>C. Thus, the microbes preferentially consume lighter <sup>12</sup>C-CH<sub>4</sub>, leaving residual CH<sub>4</sub> enriched with the heavier <sup>13</sup>C isotope in the upper peat layer (Dorodnikov *et al.* 2013). This C isotope fractionation factor was found to range from 5 to 30 ‰ in various culture experiments (Whiticar 1999).

Transport processes also affect CH<sub>4</sub> isotopic signatures. The CH<sub>4</sub> diffusion distance through the aerated zone influences the  $\delta^{13}\text{C}\text{-CH}_4$  value since methanotrophs



fractionate C as discussed above while at the same time, diffusion and plant-mediated transport processes preferentially remove lighter  $^{12}\text{C}$ -CH<sub>4</sub>. Methane fractionation may then be dependent on WT position (i.e. thickness of aerobic zone), transport mechanism and day of year/time of day (Dorodnikov *et al.* 2013). As increased sedge cover would allow for increased plant-mediated CH<sub>4</sub> transport, the vegetation community structure may influence the dominant CH<sub>4</sub> transport pathway and  $\delta^{13}\text{C}$ -CH<sub>4</sub> values accordingly.

### 2.3.3 Keeling Plot Approach

The  $\delta^{13}\text{C}$ -CH<sub>4</sub> signatures of CH<sub>4</sub> sources can be determined using the Keeling Plot approach. First proposed by Charles Keeling (1958, 1960), the “Keeling Plot” two end-member mixing model analyzes the relationship between the concentration and isotopic abundance of trace gases, such as CH<sub>4</sub>, in the atmosphere (Miller & Tans 2003, Pataki *et al.* 2003). This approach relies on the conservation of mass, where the sampled ecosystem atmospheric CH<sub>4</sub> concentration ( $c_a$ ) is a combination of the background CH<sub>4</sub> concentration ( $c_b$ ) and source CH<sub>4</sub> concentration ( $c_s$ ) (Pataki *et al.* 2003):

$$(5) \quad c_a = c_b + c_s$$

This method assumes that the only two gas components mixing are the source and background CH<sub>4</sub> and that their isotopic ratio stays constant over the observation period (Pataki *et al.* 2003). As such, each CH<sub>4</sub> component's C isotope ratio ( $\delta^{13}\text{C}$ ) follows the equation:

$$(6) \quad \delta^{13}C_a c_a = \delta^{13}C_b c_b + \delta^{13}C_s c_s$$

By substituting equation (5) into equation (6),

$$(7) \quad \delta^{13}C_a c_a = \delta^{13}C_b c_b + \delta^{13}C_s (c_a - c_b)$$

$$(8) \quad \delta^{13}C_a c_a = \delta^{13}C_b c_b + \delta^{13}C_s c_a - \delta^{13}C_s c_b$$

$$(9) \quad \delta^{13}C_a c_a = c_b (\delta^{13}C_b - \delta^{13}C_s) + \delta^{13}C_s c_a$$

$$(10) \quad \delta^{13}C_a = (c_b / c_a) (\delta^{13}C_b - \delta^{13}C_s) + \delta^{13}C_s$$

where  $\delta^{13}C_a$  is the isotopic composition of the sampled  $CH_4$ ,  $\delta^{13}C_b$  is the isotopic composition of the background  $CH_4$ , and  $\delta^{13}C_s$  is the isotopic composition of the  $CH_4$  source component. As Equation 10 is in  $y=(m/x)+b$  linear form, the source  $\delta^{13}C_s$  value is the y-intercept  $b$  in this linear regression approach (Miller & Tans 2003). Figure 2-2 demonstrates a sample Keeling Plot, where the inverse  $CH_4$  concentration is plotted against  $\delta^{13}C-CH_4$  to obtain a y-intercept that represents the isotopic signature of the  $CH_4$  source (Miller & Tans 2003, Pataki *et al.* 2003). In this example, the relatively depleted source  $\delta^{13}C-CH_4$  value of -71.7 ‰ indicates the dominance of the  $CO_2$  reduction pathway for  $CH_4$  production.

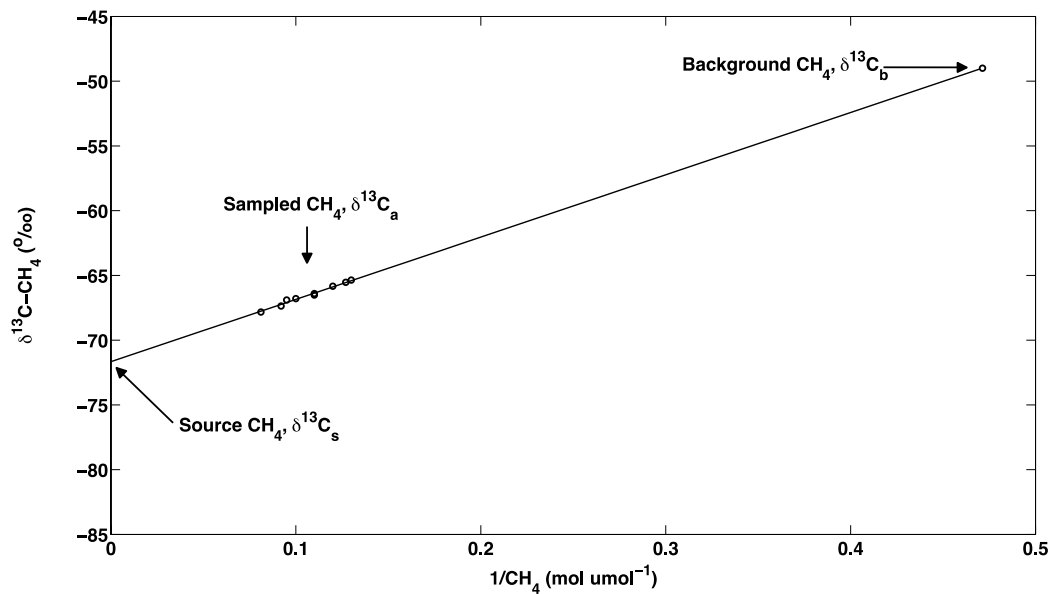


Figure 2-2: The Keeling Plot method as described by Equation 10. Inverse  $\text{CH}_4$  concentrations (mol  $\mu\text{mol}^{-1}$ ) are plotted vs. isotope  $^{13}\text{C}/^{12}\text{C}$  ratios (‰). The isotopic signatures of sampled  $\text{CH}_4$  ( $\delta^{13}\text{C}_a$ ) and background  $\text{CH}_4$  ( $\delta^{13}\text{C}_b$ ) are shown with open circles. The y-intercept (-71.7 ‰) is interpreted as the isotopic signature of the  $\text{CH}_4$  source,  $\delta^{13}\text{C}_s$ .

### 2.3.4 Redox potential measurements

As CH<sub>4</sub> production and consumption processes are microbially mediated, they require specific reduction-oxidation conditions for maximum activity levels. For example, strongly reducing conditions (<-200 mV) are favourable for the activity of methanogens, which are obligate anaerobes (Table 2-1; Zhi-Guang 1985, Mansfeldt 2003, Vorenhout *et al.* 2004). Methanotrophs, however, require O<sub>2</sub> as an electron acceptor and thus peak in activity under oxidizing conditions (>400 mV). Accordingly, the redox status of the peat substrate is an important parameter for investigating the spatial and temporal variability of CH<sub>4</sub> production potential. Measurements of the voltage difference between the peat substrate and a standard reference electrode represent the peat redox potential (Eh, mV) (Vorenhout *et al.* 2004). Redox potential indicates the ability of the peat substrate to gain or lose electrons and can range from +800 to -600 mV. Recent technological advances provide the opportunity for automated and continuous redox potential measurements from multiple locations in peatlands (Vorenhout *et al.* 2004, Vorenhout *et al.* 2011). With a field deployable multichannel datalogger and probe system, Eh profiles and temperature can be continuously monitored (e.g. 10 min frequency) and help inform the spatial and temporal variability of potential CH<sub>4</sub> production as a function of variation in temperature, microbial activity, WT position, and other environmental factors.

### 2.3.5 *Pore water collection and analysis*

To elucidate peat processes and associated turnover patterns at depth, vertical concentration profiles of chemical species in peatland pore water are commonly used (Blodau 2001, Blodau & Moore 2002). These pore water profiles allow for the quantification of processes along specific depth intervals, where the mass balance between these depths enhances process understanding of peat layers (Blodau 2001). Peatland pore water has historically been collected in situ by using equilibrium diffusion chambers (“peepers”) or suction samplers. As introduced by Hesslein (1976), the peeper technique relies on diffusive equilibration principles for in situ pore water collection without sample degassing or oxidation (Steinmann & Shotyk 1996, Gao *et al.* 2012). The peeper itself is typically constructed of Plexiglas or some sort of acrylic body that is equipped with a continuous series of dialysis chambers at certain depth intervals and dialysis membranes that separate the chambers from the surrounding peat. The chambers, which are filled with deionized water before installation, equilibrate with the surrounding pore water through diffusion of dissolved constituents (Steinmann & Shotyk 1996, Gao *et al.* 2012). While the peeper technique produces higher resolution concentration profiles, it requires long equilibration times of 27-270 hours at minimum (Gao *et al.* 2012). These peepers can also disturb the peat plot into which they are installed, and they come at a high cost. This hinders the technique’s utility for investigating temporal variations in pore water composition at multiple plots simultaneously and at the same plot multiple times.

The suction method of pore water extraction was thus developed to enable repeated sample collection from the same research plot. Pore water sampling via suction requires the installation of tubular samplers into the site peat, followed by the application

of weakly negative pressure to the tube for sample collection (Gao *et al.* 2012). Blodau and Moore (2002) found that retrieving peat pore water through suction could result in a mix of pore water from various depths and pore sizes. When compared to pore water collected from a peeper, the pore water collected via suction had higher dissolved inorganic carbon (DIC) concentrations with larger variability in values and more gradual concentration gradients (Blodau & Moore 2002). Thus, peat pore water extracted through suction techniques could produce concentration patterns that are partially confounded by macroporosity effects including preferential flow patterns. However, these sampling devices do not cause disruption to the peat layers to the same extent as larger peepers. In addition, pore water from suction samplers can be collected without device removal and does not require long equilibration times thus allowing for higher frequency measurements and more constrained temporal investigations. For deeper pore water, simple stand-pipe piezometers can be used, as they are less expensive and easier to install than peepers (Waddington *et al.* 2009, Gao *et al.* 2012). In a study investigating whether the presence of piezometers impacts below-surface pore water CH<sub>4</sub> concentrations at two peatlands, Waddington *et al.* (2009) concluded that pore water obtained from open piezometers had significantly lower CH<sub>4</sub> concentrations than pore water from sealed samplers. The open piezometers created preferential flow paths in the peat, allowing CH<sub>4</sub> to vent to the atmosphere (Waddington *et al.* 2009). Although piezometers can alter peatland gas dynamics, the disturbance is minimized with sealed piezometers. Therefore, to enhance understanding of both spatial and temporal variability of pore water constituents and their role in peatland CH<sub>4</sub> dynamics, the suction sampling technique and

use of sealed piezometers are viable alternatives to the classic equilibrium diffusion chambers.

The present study focuses in particular on the differences in CH<sub>4</sub> cycling dynamics among plant community types during a wet summer at the Mer Bleue ombrotrophic bog in Ottawa, Ontario, Canada. It aims to provide a more detailed examination of indicators of CH<sub>4</sub> production, storage and transport processes for three dominant vegetative communities at Mer Bleue through the analysis of CH<sub>4</sub> fluxes in conjunction with measurements of emitted stable isotope signatures, pore water depth profiles and redox potential. Previous studies at this site have examined the spatial and temporal variability of CH<sub>4</sub> emissions (Lai *et al.* 2014a, Moore *et al.* 2011), ecosystem-scale CH<sub>4</sub> emissions (Brown *et al.* 2014), and belowground CH<sub>4</sub> dynamics (Blodau 2001, Blodau & Moore 2003a, Blodau & Moore 2003b, Blodau *et al.* 2007), yet none have simultaneously investigated the high frequency temporal variability of CH<sub>4</sub> emissions with pore water characteristics in conjunction with isotopic techniques across the main vegetative communities. The current study hopes to provide further understanding of the production, storage and transport processes that control CH<sub>4</sub> emissions at Mer Bleue.

### 3.0 METHODS

#### 3.1 Site description

The Mer Bleue peatland complex is located 10 km east of Ottawa, Ontario (45.41°N, 75.48°W) covering a 28 km<sup>2</sup> area (Figure 3-1) (Roulet *et al* 2007, Moore *et al.* 2011, Lai *et al.* 2012). The region is characterized by a cool continental climate with a mean annual air temperature of  $6.4 \pm 0.8^{\circ}\text{C}$  and mean annual precipitation of 943 mm, 47% of which falls during the growing season between May and September (1981-2010, Environment Canada 2014). Westerly winds blow most frequently, with an average annual velocity of 13 km h<sup>-1</sup>. Each year, the first fall frost arrives ~October 7 at Mer Bleue and the last spring frost occurs ~April 30, with ~210 days where the air temperature is above freezing (Environment Canada 2014). While winter snow cover conditions vary widely between years, peak snow packs tend to range between 60 and 80 cm with snow cover present from December through March (Lafleur *et al.* 2003).

The progression of events leading to the formation of Mer Bleue bog is described in detail by Roulet *et al.* (2007). While the oldest lake sediments of the Mer Bleue peatland basin are ~9000 years old, the conditions for the peatland's formation began prior to 13,200 years ago when regional ice retreat enabled the invasion of Glacial Lake Iroquois. By 13,100 years ago, the Champlain Sea invaded this area and laid down 40-50 m of silty clay marine sediments. Freshwaters were dominant again in this basin around 10,600 years ago. Between 12,000 and 9500 years ago, outbursts from the Ottawa River eroded the postglacial channel system (in which Mer Bleue now lies) into the former floor of the Champlain Sea basin. The Mer Bleue peatland's formation spanned the last 8400 years,



with initial development as a fen, which replaced the existing lake. This was followed by bog phase initiation between 7100 and 6800 years ago (Roulet *et al.* 2007, Moore *et al.* 2011).

The northwestern arm of the Mer Bleue peatland complex is dominated by a dome-shaped ombrotrophic bog (Figure 3-1), which has hosted a long-term C cycling research program for 17 years. This area has an undulating hummock-lawn-hollow microtopography with a 0.25 m mean difference in elevation between hummock and hollow (Lafleur *et al.* 2003). The peat depth varies by ~5 m from the center (5-6 m) to the margins (<0.3 m) where a band of beaver ponds encircles the bog (Roulet *et al.* 2007). The bog surface is covered by *Sphagnum* moss (primarily *Sphagnum capillifolium*, *Sphagnum magellanicum* and *Sphagnum angustifolium*) with some *Polytrichum strictum* present. At an average height of 0.2-0.3 m, the dominant evergreen and deciduous shrubs include: *Chamaedaphne calyculata*, *Ledum groenlandicum* (= *Rhododendron groenlandicum*), *Kalmia angustifolia* and *Vaccinium myrtilloides*. Sedges (*Eriophorum vaginatum*) and select small trees on hummocks (*Larix laricina* and *Betula populifolia*) add sparse cover to the bog surface (Moore *et al.* 2011). Mer Bleue is considered to be a relatively dry bog, as indicated by the 1998-2014 mean growing season WT depth of 42 cm below hummock top at the tower site. Over this 16-year period, the mean growing season WT position ranged from 33 to 50 cm below peat surface, with observed daily WT minimum and maximum of 78 and 23 cm below the peat surface respectively.



Figure 3-1: Location of the tower site, gas analyzers and logging equipment, and chamber area at the Mer Bleue research site. The star, representing the tower site, is located at  $45^{\circ}24'33''\text{N}$ ,  $75^{\circ}31'07''\text{W}$ . Map data: Google, Digital Globe (Imagery Date: 9/24/2013).

### 3.1.1 Sampling Area

The research sampling area was composed of 6 automated chamber and 3 manual chamber plots located approximately 50 m south of the eddy covariance tower (Figure 3-1). The autochambers were chosen to be representative of the dominant vascular plant communities including *Chamaedaphne*- and *Ledum*-dominated communities (plots C-A 1, 2, and 3 and plots L-A 1, 2 and 3, respectively), while manual chambers were chosen to represent the *Eriophorum*-dominated community (plots E-M 1, 2, and 3). While the number of sampling chambers was limited by the existing infrastructure and physical arrangement of dominant plant communities, the nine selected chambers also captured this peatland's range in microtopography and WT. The autochamber and manual chamber plots were accessed via wooden boardwalks to minimize peat damage and vegetation disturbance. Each chamber was located < 15 m from the gas analyzers and logging equipment (Figure 3-1). The characteristics of each chamber plot, including dominance of vascular and nonvascular species and leaf area index (LAI), are presented in Table 3-1. Leaf area index was measured weekly at each chamber as m<sup>2</sup> leaf area per m<sup>2</sup> ground area (m<sup>2</sup> m<sup>-2</sup>) using two methods. First, biomass LAI was measured using a Plant Canopy Analyzer (LAI-2000, LI-COR Inc., Lincoln, NE). This instrument calculates the ratio of foliage area to ground area based on radiation measurements made above (2 measurements) and below (3 measurements) the standing plant canopy but does not include the LAI associated with creeping species or moss. It also cannot distinguish between living and dead foliage and simply gives a measure of canopy light interception. Green LAI was determined using the point intercept method. The point intercept frame (60 x 60 cm) had a fishing line grid of 5 x 5 cm squares for a total of 121 grid points.

The frame was leveled and centered above each sampling collar and a metal rod (8 mm diameter) was dropped at alternating grid points such that it was perpendicular to and touching the moss carpet (Adkinson 2009). The number of times the rod came in contact with a living vascular plant organ (leaf, stem, flower, shoot) for each species was recorded over 35 and 9 grid points for automated and static chamber collars, respectively (Larmola *et al.* 2013). The % cover and species composition of the vascular and nonvascular canopy was also determined (Table 3-1). All chambers plots had 100% or near 100% moss cover for the duration of the growing season.

Table 3-1: Vegetation characteristics at the 9 chamber plots including dominance (%) of vascular and nonvascular plant species, growing season maximum biomass leaf area index (LAI, m<sup>2</sup> m<sup>-2</sup>), total and *Eriophorum vaginatum* (EV) green LAI, and maximum percent of EV cover.

<b>Chamber</b>	<b>Vascular plant species (% dominance)<sup>a</sup></b>	<b>Nonvascular plant species (% dominance)<sup>a</sup></b>	<b>Max. biomass LAI (m<sup>2</sup> m<sup>-2</sup>)<sup>b</sup></b>	<b>Max. total green LAI (m<sup>2</sup> m<sup>-2</sup>)<sup>b</sup></b>	<b>Max. EV green LAI (m<sup>2</sup> m<sup>-2</sup>)<sup>b</sup></b>	<b>Max. % EV cover</b>
<b><i>Chamaedaphne</i></b>						
<i>C-A 1</i>	CC 53%, EV 27%, LG 10%, VO 5%, VM 3%, MT 2%	SP 63%, P 37%	2.83	6.03	2.03	34
<i>C-A 2</i>	CC 84%, LG 8%, VO 5%, MT 2%, VM 1%	P 60%, SP 40%	3.61	6.83	0	0
<i>C-A 3</i>	EV 50%, CC 42%, LG 6%, VM 2%	P 57%, SP 11%	7.24	10.54	6.34	60
<b><i>Ledum</i></b>						
<i>L-A 1</i>	LG 40%, EV 40%, MT 10%, CC 9%, VO 1%	SP 96%, P 4%	6.56	5.32	2.66	53
<i>L-A 2</i>	EV 74%, LG 18%, KA 3%, VO 3% MT 2%	SP 87%, P 8%	5.19	10.74	8.71	82
<i>L-A 3</i>	EV 68%, LG 22%, MT 7%, VO 3%	SP 67%, P 31%	4.22	8.09	5.83	74
<b><i>Eriophorum</i></b>						
<i>E-M 1</i>	EV 97%, LG 1%, CC 1%, MT 1%	SP 70%, P 6%	5.33	21	20.44	99
<i>E-M 2</i>	EV 83%, CC 9%, VO 5%, MT 3%	SP 64%, P 22%	4.18	12.11	11.11	93
<i>E-M 3</i>	EV 89%, VO 6%, KA 2%, CC 2%, LG 1%	SP 94%, P 0%	4.14	12.78	12.11	96

<sup>a</sup>CC: *Chamaedaphne calyculata*, EV: *Eriophorum vaginatum*, KA: *Kalmia angustifolia*, LG: *Ledum (rhododendron) groenlandicum*, MT: *Maianthemum trifolium*, VM: *Vaccinium myrtilloides*, VO: *Vaccinium oxycoccos*, AP: *Andromeda polifolia*, P: *Polytrichum* moss, SP: *Sphagnum* moss.

<sup>b</sup>From DOY 133 to DOY 239.

## 3.2 Data collection

### 3.2.1 Autochamber $\text{CH}_4$ flux measurements

The existing autochamber system (Figure 3-2) was used to collect chamber flux measurements  $24 \text{ h day}^{-1}$  between May 1 and October 1, 2014. Each autochamber (see Appendix A Section 7.0) was constructed using a PVC collar (height=0.385 m) inserted into the peat and attached to a clear Plexiglas<sup>®</sup> dome by an aluminum frame (see Lai *et al.* 2012 for full description). The collars enclosed a surface area of  $0.21 \text{ m}^2$  (internal diameter=0.52 m) with collars inserted to a depth of 0.16 to 0.31 m (Figure 3-4), which minimized air leakage through the peat. The autochamber domes were 0.205 m high and were sealed when closed by a partially-deflated bicycle tube and foam gasket. The chamber headspace air was mixed by a small brushless fan fixed to the inside of the dome. Pressure equilibration between external and internal environments was achieved during autochamber flux measurements through a 0.50 m long vent tube that was coiled, open at both ends and emerged from the dome top.

The autochamber system was controlled by a datalogger (CR23X, Campbell Scientific) and was programmed to close the chamber domes sequentially every half hour for 2.5 minutes during the day (09:00-21:00 EST) to minimize plant stress, and 15 minutes at night (21:00-09:00 EST) to minimize bias introduced by atmospheric turbulence (Lai *et al.* 2012). This produced a measured flux value for all chambers each half hour during the day and every 4 hours at night. An oil-free air compressor and pneumatic cylinder system powered the opening and closing of chambers. Sampling tubes drew the gases from the appropriate chamber to a sampling manifold where  $\text{CH}_4$

concentrations were measured using a fast CH<sub>4</sub> analyzer using off-axis integrated cavity output spectroscopy (model DLT-100, Los Gatos Inc., Mountain View, CA), and CO<sub>2</sub> concentrations were measured by a closed-path infrared gas analyzer (LI-6262, LI-COR Inc.) in a temperature controlled housing. Sampled once per second and averaged every 5 seconds, gas analyzer data were automatically sent to a computer in an enclosed hut at the nearby tower site (Figure 3-1). The gas analyzers and autochambers were maintained regularly, particularly at the beginning of the season (e.g. replaced tubing, patched leaks, repaired fans). The system's equipment (gas analyzers, data logger, air compressor and manifold and pump system) was located under a shelter at the junction of the primary and secondary boardwalks (Figure 3-5).



Figure 3-2: Photos of a) the autochamber system at Mer Bleue used to measure CH<sub>4</sub> fluxes at b) the plant community scale with c) a Plexiglas<sup>®</sup> dome fitted to a PVC collar.

Flux data were processed in Matlab 8.1.0.604 (The Math Works, Inc., 2013).

Only nighttime data were used as the autochambers closed for 15 minute periods, which reduced the flux underestimations caused when winds flush the near surface peat and there is an associated decrease in headspace turbulence during deployment (Lai *et al.* 2012). When deployment periods are too short (< 13 minutes), these turbulence conditions influence headspace CH<sub>4</sub> concentrations after the chamber closes. As a result, we used only the last 250 seconds of data to calculate nighttime CH<sub>4</sub> fluxes. CH<sub>4</sub> fluxes ( $F_{CH_4}$ , nmol m<sup>-2</sup> s<sup>-1</sup>) were calculated as:

$$(11) \quad F_{CH_4} = \frac{\rho V}{R^* T_{air} A} \frac{dC}{dt}$$

where  $\rho$  is air pressure (Pa),  $V$  is chamber volume (m<sup>3</sup>),  $R^*$  is the ideal gas constant (8.314 J K<sup>-1</sup> mol<sup>-1</sup>),  $T_{air}$  is mean air temperature (°K),  $A$  is chamber surface area (m<sup>2</sup>), and  $dC/dt$  is the rate of change in CH<sub>4</sub> mixing ratio in the chamber over the measurement period (nmol mol<sup>-1</sup> dry air s<sup>-1</sup>). The mixing ratio of CH<sub>4</sub> takes into consideration the effects of water vapour dilution and was calculated using the headspace mole fraction of CH<sub>4</sub> and the water vapour mole fraction measured by the LI-6262 gas analyzer. A negative CH<sub>4</sub> flux indicates uptake by the peatland while a positive flux indicates CH<sub>4</sub> loss to the atmosphere.



### 3.2.2 Static chamber $CH_4$ flux measurements

As there were no existing autochamber plots representative of the *Eriophorum*-dominated community, 3 *Eriophorum* plants for manual  $CH_4$  flux measurement were selected in close proximity to the autochamber system (Figure 3-3).

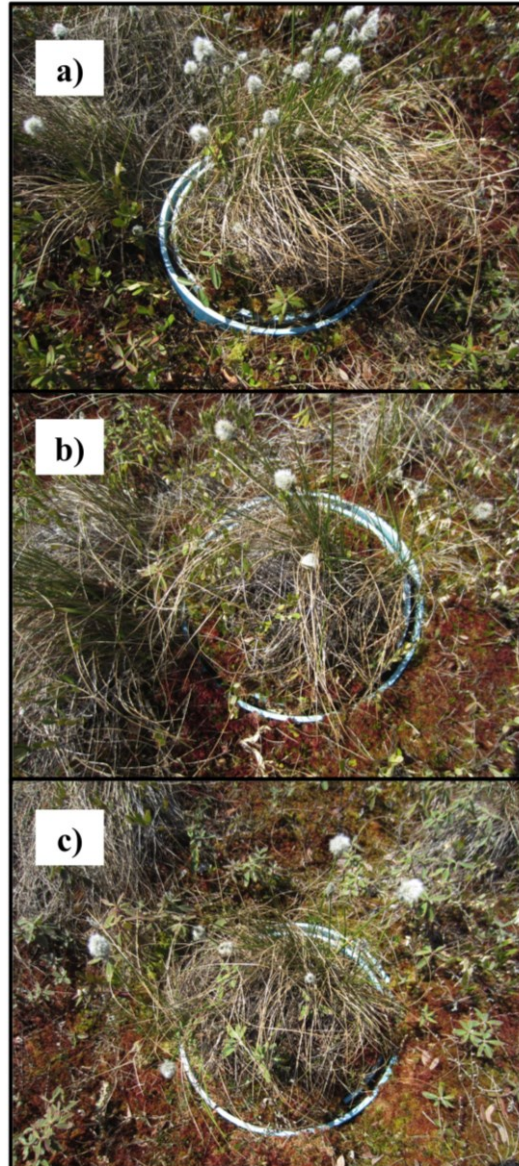


Figure 3-3: Photos of manual chamber collars inserted into the peat at a) E-M 1, b) E-M 2 and c) E-M 3 *Eriophorum* plants.

Grooved collars of 0.24 m internal diameter (surface area= 0.05 m<sup>2</sup>) were inserted ~10 cm into the peat to be compatible with existing 18 L opaque chambers at the site. All manual chambers were covered with aluminum foil tape to minimize temperature increase during measurement (Bubier *et al.* 1995a). Flux measurements were made at least twice weekly (2-3 rounds per sampling day) between 07:00 and 15:00 EST. A seal was created between collar and chamber during the measurement period by filling the collar grooves with water. The air in these static chambers was sampled at 5 min intervals for a total of 20 min starting at time 0. First, 60 cc of air was pumped between a sealed syringe and the chamber via tubing 3 times to mix the headspace air. Next 24 cc of air was extracted and injected via syringe and needle to an evacuated 12 cc vial with a small amount of magnesium perchlorate as a desiccant. Vials were analyzed for CH<sub>4</sub> mixing ratio on a Varian CP-3800 gas chromatograph at Carleton University, with methods described in detail by Wilson & Humphreys (2010). The gas chromatograph used flame-ionization detection and methanizer technology at 300°C and 350°C, respectively, with a He carrier gas at an injection rate of 30 ml min<sup>-1</sup>. Quality control was maintained during gas sampling using three replicates of five standards (439.2-15212.6 ppm CO<sub>2</sub>, 1.08-19.4 ppm CH<sub>4</sub>) and He blanks. The chamber was removed for at least 2 minutes between successive rounds of flux measurements.

The slope of the regression between CH<sub>4</sub> concentration and time during the 20-minute sampling period was used to calculate CH<sub>4</sub> fluxes (Roulet *et al.* 2007) following Equation 11. Samples were rejected when CO<sub>2</sub> and CH<sub>4</sub> concentrations remained close to ambient concentrations for the duration of the sampling period when a substantial flux was expected given the measurements from that chamber during a particular time period,

indicating contamination of the sample with ambient air through a possible chamber or vial leak. Samples were also rejected when there was a large initial spike in CO<sub>2</sub> and CH<sub>4</sub> concentrations, as this suggested that the placement of the chamber on the collar disturbed the peat and caused a rapid degassing of trace gases to the chamber headspace. Approximately 17% of samples were rejected.

Although there were some inherent differences among the chamber sampling methods (static vs. circulating flow, collar area, chamber headspace, slope determinations), we were confident that the two methods were comparable as the magnitude and temporal variability of fluxes from the *Eriophorum* plots were very similar to those presented by Lai *et al.* (2014a) when the autochambers had been located on *Eriophorum* plants near the static chamber plots used in this study.

### **3.2.3 Pore water collection**

Weekly pore water collection occurred between 07:00 and 15:00 EST starting on June 4 and ending on September 12, 2014. In order to minimize vegetation and subsurface peat disturbance within sampling collars, pore water was collected at least 20 cm from the collar edge in an area with similar moss and vegetation cover. The order of pore water collection for each sampling plot was randomized week-to-week. A portable pore water sipper (Figure 3-5a) was used to collect pore water directly below the WT on each sampling day. The sipper was inserted into the peat at the required depth and a 60 ml syringe was attached to the top of the sipper with a 3-way stopcock. The syringe was used to pull up and flush out 30 ml of pore water. After flushing, 30 ml of pore water was collected and the stopcock was closed to the atmosphere.

Pore water at 50, 65 and 80 cm below the peat surface was collected using permanent pore water samplers (Figure 3-4 , Figure 3-5b). Pore water “seepers” were constructed from a section of PVC piping (length=15 cm, internal diameter=2.54 cm) with rows of holes at 1 cm intervals drilled 10 cm from the sealed end of the seeper and covered in fine mesh. Rubber stoppers were inserted at both ends of the seeper and two small holes were drilled through the top stopper so that two Bev-A-Line tubes could be installed for sample extraction and nitrogen pressure equalization. A 3-way stopcock was added at the end of each tube to enable the use of a syringe above the peat surface. Pore water seepers were installed at 50, 65 and 80 cm on June 4, 2014 at each plot. Pore water was collected from seepers using a similar method to the sipper procedure. A 60 ml syringe was attached to the stopcock of the sample extraction tube emerging from the peat. A Tedlar bag (volume=1L) filled with nitrogen was attached to the second tube’s stopcock for pressure equilibration. After opening the stopcocks to enable nitrogen flow to depth and pore water flow to surface, a syringe was used to pull up and flush out 30 ml of pore water. The 30 ml pore water sample was then collected and both stopcocks were closed.

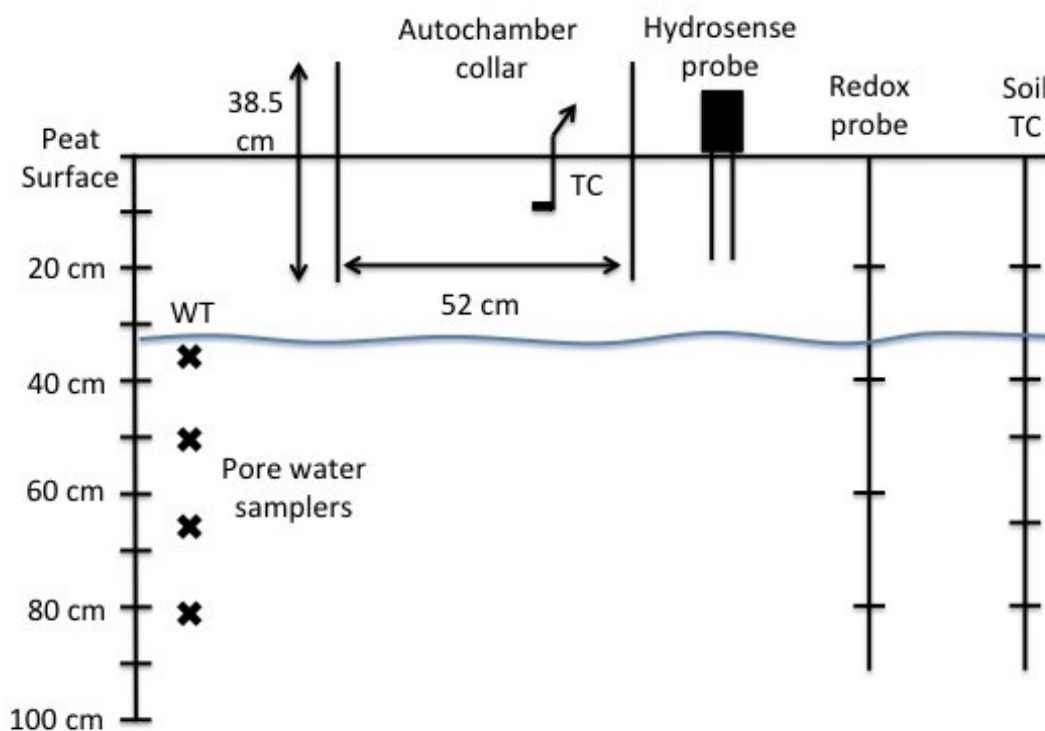


Figure 3-4: Example setup of instrumentation at an autochamber collar showing sampling depths for pore water samplers, Hydrosense probe, redox probe and soil thermocouples (TC) inside and outside the chamber collar. Note that only a single soil thermocouple profile was installed and was located near the flux tower ~ 50 m from the autochambers. All other sensor placements and depths are to scale.

A secondary pore water sampler was designed and installed on August 6, 2014 at 50, 65 and 80 cm for each plot. This fat sipper design consisted of a piezometer for each sampling depth (internal diameter=1.91 cm) with holes drilled at the appropriate depth, a cap sealing the bottom and a cap and septa at the top (Figure 3-5c). Before the first sampling, piezometers were flushed with nitrogen via tubing and a 22-gauge (0.7 mm x 25 mm) needle piercing the septa in the top cap (a second needle was inserted into the septa to allow for pressure equilibration). Samples were obtained by removing the top septa and cap, inserting a sipper and collecting pore water as per the sipper and syringe

method described above. After pore water extraction, the piezometers were sealed and flushed with nitrogen to limit O<sub>2</sub> entry to the subsurface.

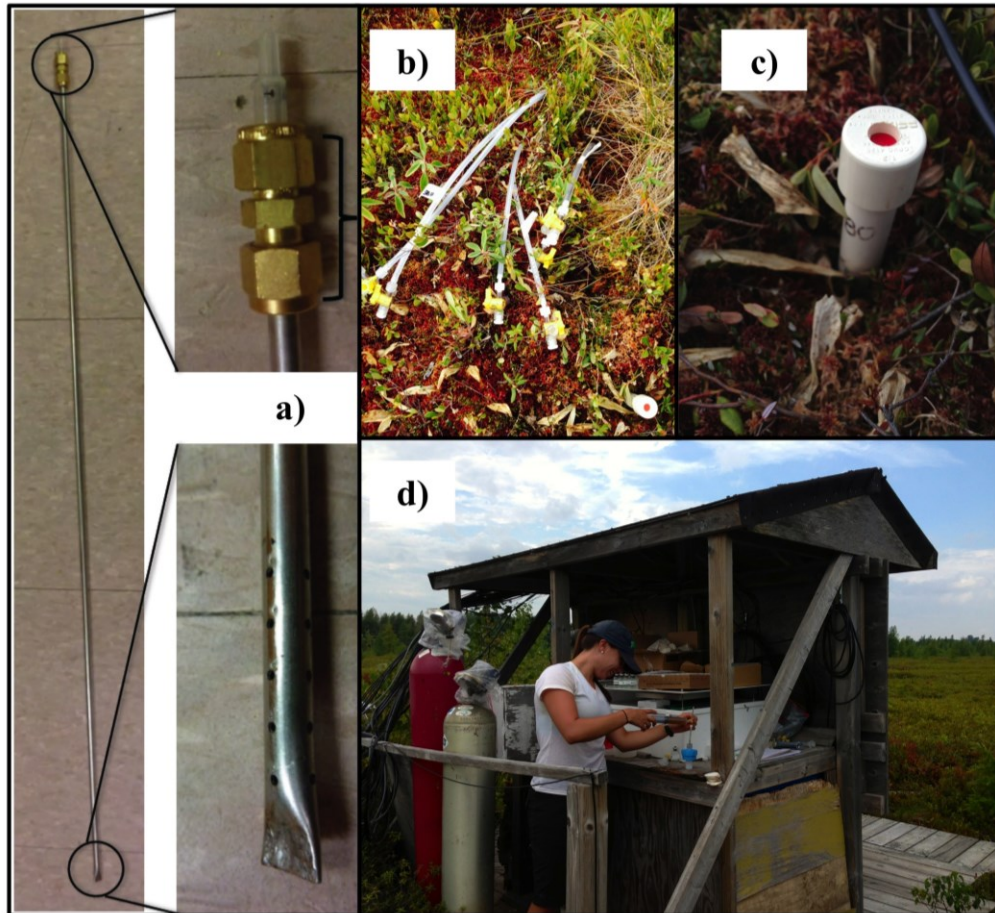


Figure 3-5: Photos of pore water sampling instrumentation where a) a pore water sipper was used to collect pore water from depths < 50 cm below peat surface and permanent b) pore water seepers and c) piezometers were used to collect pore water at 50, 65 and 80 cm. After samples were collected, d) pore water filtration and equilibration for isotope analysis was carried out in the field at a small equipment hut.

Once a 30 ml pore water sample was obtained in a 60 ml syringe using the above methods, the remaining 30 ml of the syringe was filled with ambient air. The stopcock was then closed to the atmosphere and the syringe was shaken for 2 minutes to equilibrate liquid and gas CH<sub>4</sub> concentrations. The syringe headspace was pushed through a 22-gauge needle to displace saltwater in a 30 ml vial (sealed and crimped) until

15 ml of headspace remained in the syringe. The remaining 15 ml of headspace sample air was used to overpressurize the saltwater solution. 2.5 ml of gas in the headspace was analyzed for CH<sub>4</sub> on the Varian CP-3800 as described above. The water remaining in the syringe was then filtered through a 0.40µm glass fiber filter (diameter=47 mm) and stored in a 20 ml scintillation vial in the fridge for DOC analysis. Pore water CH<sub>4</sub> concentrations (µmol L<sup>-1</sup>) were calculated according to the equations found in Appendix B Section 7.0.

Pore water samples were analyzed on a Shimadzu TOC/TN-V<sub>CSN</sub> analyzer at McGill University for DOC (mg L<sup>-1</sup>) and total dissolved nitrogen, TDN (mg L<sup>-1</sup>) using the combustion catalytic oxidation method at 680°C. The carbon to nitrogen ratio (C/N) was calculated as the ratio between DOC and TDN. UV absorbance (m<sup>-1</sup>) was determined at 250, 254 and 365 nm on an Agilent Cary 60 UV-Vis spectrophotometer with a fiber optic dip probe in dual beam mode at a scan rate of 600 nm/min. The aromaticity of DOC was approximated by calculating specific ultraviolet absorbance at 254 nm, SUVA<sub>254</sub> (L m<sup>-1</sup> mg<sup>-1</sup> DOC), as:

$$(12) \quad SUVA_{254} = \frac{UV \text{ Absorbance at } 254 \text{ nm (m}^{-1}\text{)}}{DOC \text{ (mg L}^{-1}\text{)}}$$

where the UV absorbance at 254 nm (m<sup>-1</sup>) is divided by the DOC concentration (mg L<sup>-1</sup>).

### 3.2.4 *Storage Change*

Each pore water CH<sub>4</sub> measurement was representative of a peat layer based on sampling depth (20 cm, below WT, 50 cm, 65 cm, 80 cm). For example, the pore air sample taken at 20 cm represented the aerated zone from the peat surface to the average WT position at each individual chamber. The average WT from DOY 155-255 was determined for each chamber and used to calculate the unique thickness of peat layers for each sampling depth at each chamber. Methane concentration values were weighted for each specific peat layer thickness.

These depth-weighted CH<sub>4</sub> concentration values were used for storage change calculations at specific depths (i.e. 50 cm sampling depth had associated peat layer thickness of 15 cm), or after summing over a total depth of 1 m. Units were converted from  $\mu\text{mol cm}^{-2}$  to  $\text{mg cm}^{-2}$ , and a plot of depth-weighted CH<sub>4</sub> concentrations ( $\text{mg cm}^{-2}$ ) over time (DOY) was constructed for each chamber. Linear regressions were then applied to obtain chamber-specific slope values, where slope represented the change in CH<sub>4</sub> storage over time for that chamber ( $\text{mg cm}^{-2} \text{d}^{-1}$ ). These values were converted to units of  $\text{mg m}^{-2} \text{h}^{-1}$  to obtain final storage change values ( $S_{\text{CH}_4}$ ,  $\text{mg m}^{-2} \text{h}^{-1}$ ). The average CH<sub>4</sub> flux ( $F_{\text{CH}_4}$ ,  $\text{mg m}^{-2} \text{h}^{-1}$ ) was then calculated for each chamber over the pore water sampling period using manual sampling day data.



### **3.2.5 Headspace sample collection**

Weekly headspace sample collection occurred on pore water sampling days between July 31 and September 12, 2014. At the end of autochamber and static chamber measurements, 30 ml headspace samples were obtained using the septum-stopcock extraction method with a 22-gauge needle and syringe. These 30 ml gas samples were injected into evacuated, sealed and crimped 30 ml vials.

Headspace gas samples were analyzed at Arizona State University on a Picarro G2201-*i* CRDS (Cavity Ring-down Spectroscopy) Analyzer in combined mode for isotopic C in CO<sub>2</sub> and CH<sub>4</sub>. This instrument had a simultaneous mode precision of <0.16‰ for  $\delta^{13}\text{C}$ -CO<sub>2</sub> and <1.15‰ for  $\delta^{13}\text{C}$ -CH<sub>4</sub>, and maximum peak-to-peak drift of <0.6‰ for  $\delta^{13}\text{C}$ -CO<sub>2</sub> and <1.5‰ for  $\delta^{13}\text{C}$ -CH<sub>4</sub> (Picarro, Inc. 2015). The sample flow rate was <50 sccm (typical ~25 sccm) at 101.3 kPa. The isotopic composition and concentration of sample and background CH<sub>4</sub> were used to calculate  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signatures at each chamber through the Keeling Plot approach (Equation 10).

### **3.2.6 Redox potential measurements**

This study used a HYPNOS 3 datalogger for redox measurements. The HYPNOS logger measured the difference in potential between the soil matrix and an Ag/AgCl reference probe (QM710X from Q-I-S) with 3 M KCl filling solution. Nine fiberglass probes with platinum sensor tips and temperature sensors (DS18B20,  $\pm 5^\circ\text{C}$  accuracy from 10 to  $85^\circ\text{C}$ ) at 20, 40, 60 and 80 cm (Figure 3-4) were installed adjacent to each chamber on July 18, 2014. The reference probe was installed close to the junction of the main and secondary boardwalks, near the equipment hut (Figure 3-1). The data were

stored on an SD card (Vorenhout *et al.* 2011). Readings were recorded every 10 minutes and data were downloaded from the SD card every ~2 weeks.

Standardized redox potential values ( $E_h$ , mV) were calculated according to:

$$(13) \quad E_h = E_m + E_{ref}$$

where  $E_h$  is the standardized redox potential,  $E_m$  is the measured redox potential, and  $E_{ref}$  is the potential of the reference probe (Vorenhout *et al.* 2011). As the Ag/AgCl reference probe had a 3M KCl filling solution, the linear relation between  $E_{ref}$  and temperature (Langmuir 1971) was used to determine temperature-specific  $E_{ref}$  values for the calculation of  $E_h$  values. These standardized redox potentials were then adjusted for pH according to:

$$(14) \quad E_{h-corrected} = E_h + 59(pH - 7)$$

where the pH was assumed to be stable at 4.

### **3.2.7 Environmental variables**

Ancillary field measurements included environmental variables presented in Table 3-2. These variables were measured at varying frequency, as indicated in the table, and employed both manual and automated methods. They served to characterize the microclimate and vegetation for each sampling chamber and to quantify differences between vegetative communities.

Wells were established at each sampling plot for the manual measurement of WT position. Depth to WT was measured manually at each plot using a “bubbler” each day

that manual flux measurements were obtained. WT was also measured at 30 min intervals at each plot throughout the season using automatic capacitive water level loggers installed in wells adjacent to each chamber. Rainfall was measured at 2 rain gauges (NE and S of main tower site hut) every day that fieldwork was conducted. An existing tipping bucket rain gauge near the eddy covariance tower also measured rainfall at the site, and was calibrated for under-catching using a relationship developed with manual measurements.

Peat temperature was measured continuously at 10 cm depth using chromel-constantan thermocouples installed in each collar as of May 29, 2014. An existing thermocouple profile at the tower site measured peat temperature at 20, 40, 50, 65 and 80 cm below the peat surface.

A datalogger (CR7X, Campbell Scientific) and existing instrumentation at the eddy covariance tower site recorded additional climate variables including air temperature, atmospheric pressure, and photosynthetic photon flux density. Lafleur *et al.* (2003) provide a description of instruments; measurements were recorded every 5 s and averaged over 30 min.

Air temperature (°C) and rainfall (mm) data collected at Mer Bleue were compared to 30-yr Ottawa climate normals (1981-2010, Environment Canada 2014) measured at the Ottawa International Airport, located approximately 20 km southwest of the research site. Each month's daily average temperature and standard deviation were analyzed along with total monthly rainfall from May through September.

Table 3-2: Frequency and method of manual and automated measurements for environmental variables.

<i><b>Variable</b></i>	<i><b>Measurement frequency</b></i>	<i><b>Measurement Method</b></i>
WT depth	2-3 times per week 30 min	1 well per chamber <sup>§</sup> Odyssey capacitive water level logger
Rainfall	3-4 times per week Variable	2 rain gauges <sup>§</sup> Tipping bucket rain gauge*
Peat temperature (10, 20, 40, 50, 65, 80 cm)	5s (averaged every 30 min)	Chromel-constantan thermocouples
Air temperature (0.5 m height)	5s (averaged every 30 min)	Chromel-constantan thermocouples*
Atmospheric pressure	5s (averaged every 30 min)	Barometer*
Photosynthetic photon flux density	5s (averaged every 30 min)	Quantum sensor*
Biomass LAI	1 time per week	LAI-2000 Plant canopy analyzer <sup>§</sup>
Green LAI	1 time per week	Point intercept method <sup>§</sup>
Moss moisture	2 times per week	Peat tabs <sup>§</sup>
Moss live sampling	1 time per week	Manual collection outside chamber <sup>§</sup>
Moss volumetric water content	2 times per week	HydroSense soil water sensor <sup>§</sup>

Note: LAI=leaf area index.

<sup>§</sup>manual measurement

\*existing instrumentation at eddy covariance tower site.

### 3.2.8 *Statistical analyses*

All statistical analyses were completed using JMP 12.0.1 (SAS Institute Inc.). For seasonal analyses, the data were separated into late spring (LSp, DOY 132-154), early summer (ES, DOY 155-189), mid-summer (MS, DOY 190-221), and late summer (LS, DOY 222-255) measurement periods. Fluxes were available for the late spring period, while pore water data collection started with the early summer period. Redox potential measurements were only active during the late summer period. All data were tested prior to analyses for assumptions of normality and equal variances. Data that did not meet the assumption of normality were transformed. While this did not fully correct the distribution in all cases, the restricted maximum likelihood (REML) method used in mixed model analysis is robust to deviations from normality (Fung & Xu 2010). Statistical analyses were tested at the significance level of 0.05, unless otherwise stated.

A simple linear regression analysis was applied to examine the relationship between mean seasonal depth to WT and log CH<sub>4</sub> flux. Averages were determined based on manual static chamber sampling day data to ensure completeness of flux data at *Eriophorum* plots. The relationship between mean 50 cm CH<sub>4</sub> storage change and mean CH<sub>4</sub> flux was examined using a Pearson's correlation analysis and nonlinear regression for manual sampling days during the pore water collection period. The relationships among indicators of CH<sub>4</sub> production, storage and transport were examined when all variables were available between DOY 212-255. Analyses of calculated averages of these variables at the 50 cm depth included a principal components analysis (PCA), and multiple correlation analysis using Pearson correlations.

Repeated measures analyses using mixed models were applied to investigate CH<sub>4</sub> fluxes throughout the growing season, pore water CH<sub>4</sub> concentrations during the pore water sampling period (DOY 155-255), and the  $\delta^{13}\text{C}$ -CH<sub>4</sub> isotopic signatures for headspace sampling days. The model was fit using restricted maximum likelihood (REML) method in the standard least squares personality to view the variation in intercepts and slopes. The REML method is the commonly accepted method for fitting mixed models, as it provides accurate results regardless of whether the data is balanced or not (JMP 7 Statistics and Graphics Guide 2007). The “chamber” variable was designated as a random effect and nested within vegetative community since the chamber plots were randomly selected from the vegetation populations. The fixed effects were analyzed to a factorial degree of 2 in order to examine interaction effects (JMP 7 Statistics and Graphics Guide 2007).

## **4.0 RESULTS**

### **4.1 Environmental conditions**

The environmental conditions for the 2014 season at Mer Bleue were analyzed in reference to the 1981-2010 climate normals (Environment Canada 2014) (Figure 4-1).

The daily average air temperatures were comparable with Ottawa climate normals for the early summer 2014 period, while the mid- to late-summer periods were slightly cooler than the long-term average. Mer Bleue experienced a wet summer compared to the last 30 years, with particularly heavy rainfall events in June contributing to a total rainfall of 569.0 mm from May through September in comparison to the climate normal of 440.4 mm for the same period (Figure 4-1). The WT recorded near the tower (where the long term record has been maintained since 1998) reflected these wet conditions and was high relative to the long-term average (Figure 4-2).

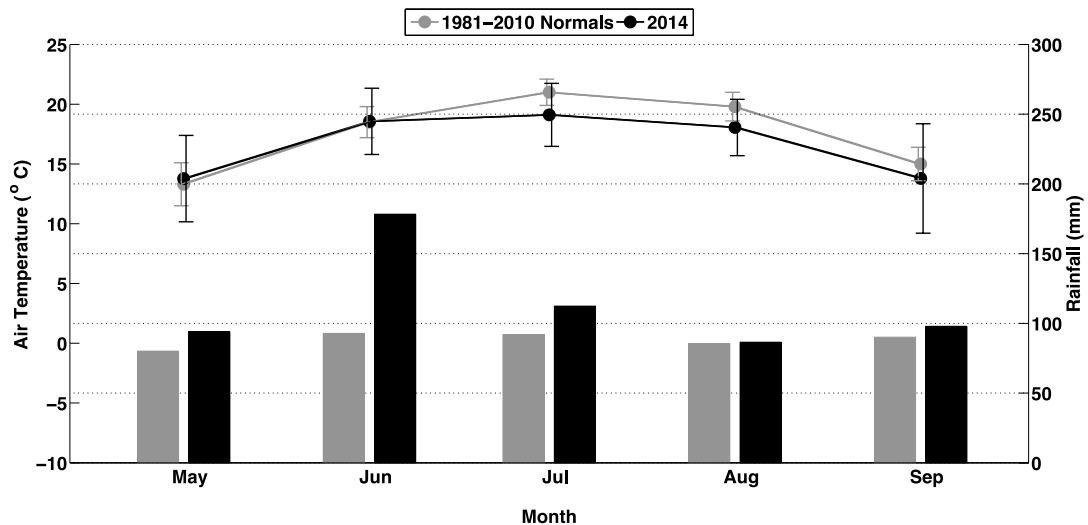


Figure 4-1: Mean ( $\pm 1$  SD) daily average air temperature (lines) and total monthly rainfall (bars) from May through September 2014 at Mer Bleue peatland in reference to 1981-2010 climate normals measured at the Ottawa Macdonald-Cartier International Airport (Environment Canada 2014).

The 2014 WT at a given location varied by only 10 cm from the mean and was thus restricted to a narrow fluctuation range compared to previous years. During 2014, the WT never reached mid-summer depths typical at Mer Bleue. Differences in WT among plant communities were a result of microtopography: *Chamaedaphne* plants were located on hummock tops while *Ledum* and *Eriophorum* were in hollows or depressions between hummocks (Figure 4-2). As a result, the *Chamaedaphne* community's WT was on average 10 cm lower than *Ledum* and *Eriophorum* communities. Note that the *Chamaedaphne* community and the tower site had very similar WT depth patterns because of their similar microtopographic position, while *Ledum* and *Eriophorum* communities both had WT closer to the surface due to their relatively low position. The



*Eriophorum* community had the shortest WT measurement period due to the timing of instrument installation and removal.

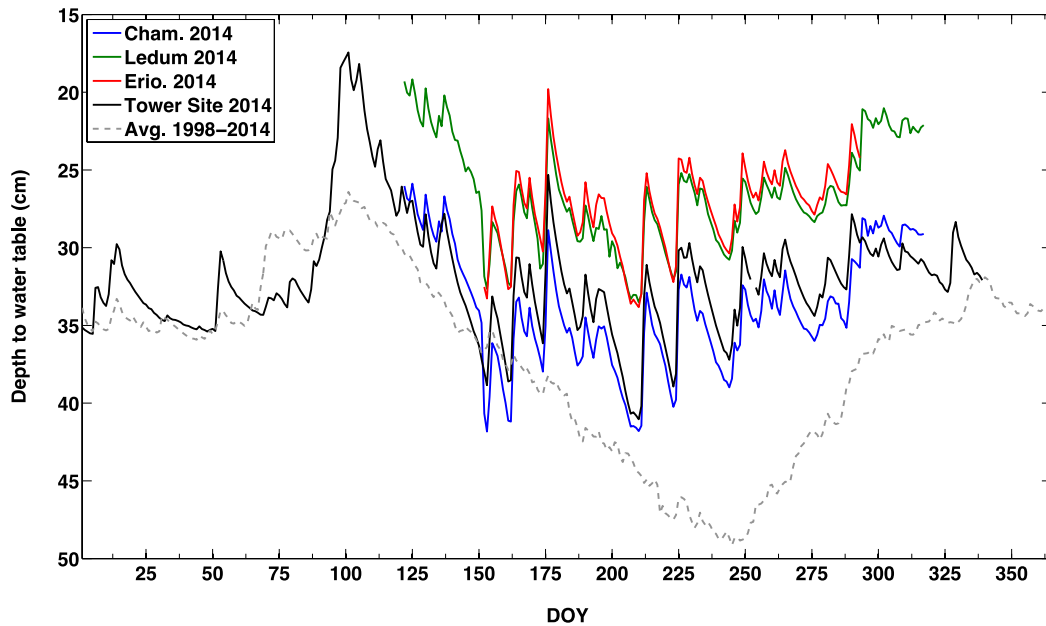


Figure 4-2: Mean daily depth to water table from peat surface (cm) measured at *Chamaedaphne*, *Ledum* and *Eriophorum*-dominated communities and tower site at Mer Bleue, 2014. The long-term mean water table position recorded at the tower site between 1998 and 2014 is shown for comparison.

The 10 cm peat temperatures followed daily mean air temperature patterns with slightly dampened day-to-day variations at this depth (Figure 4-3). With increased depth from the peat surface, temperatures were cooler with fewer extreme values as the dampening of day-to-day variations in peat temperature increased with depth (Figure 4-4). At 20 cm and below, peat temperature consistently increased over the 2014 measurement period to DOY 255.

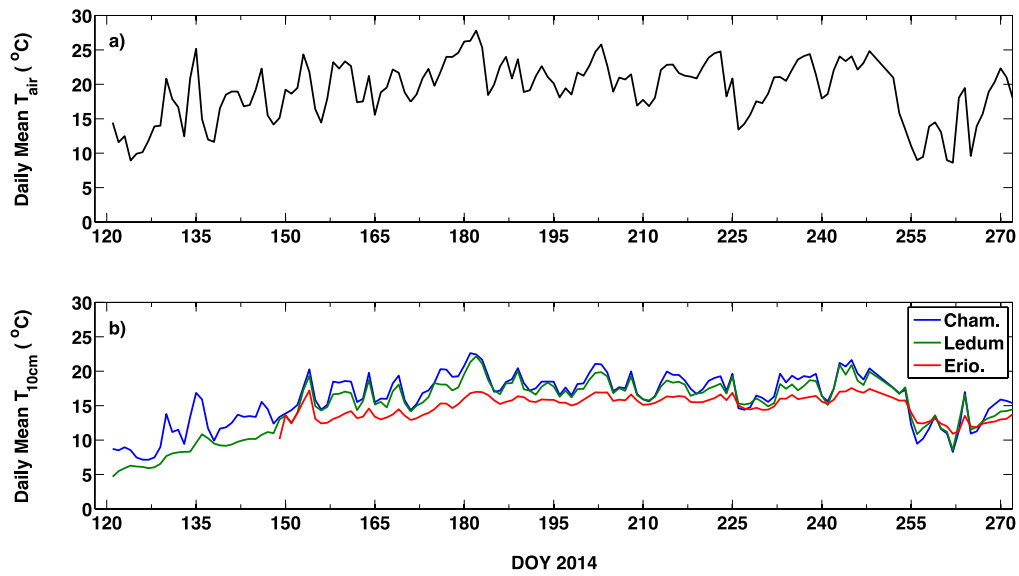


Figure 4-3: Daily mean a) air temperature ( $T_{air}$ , °C) and b) 10 cm peat temperature ( $T_{10cm}$ , °C) for the *Chamaedaphne*, *Ledum* and *Eriophorum*-dominated communities during the 2014 measurement period.

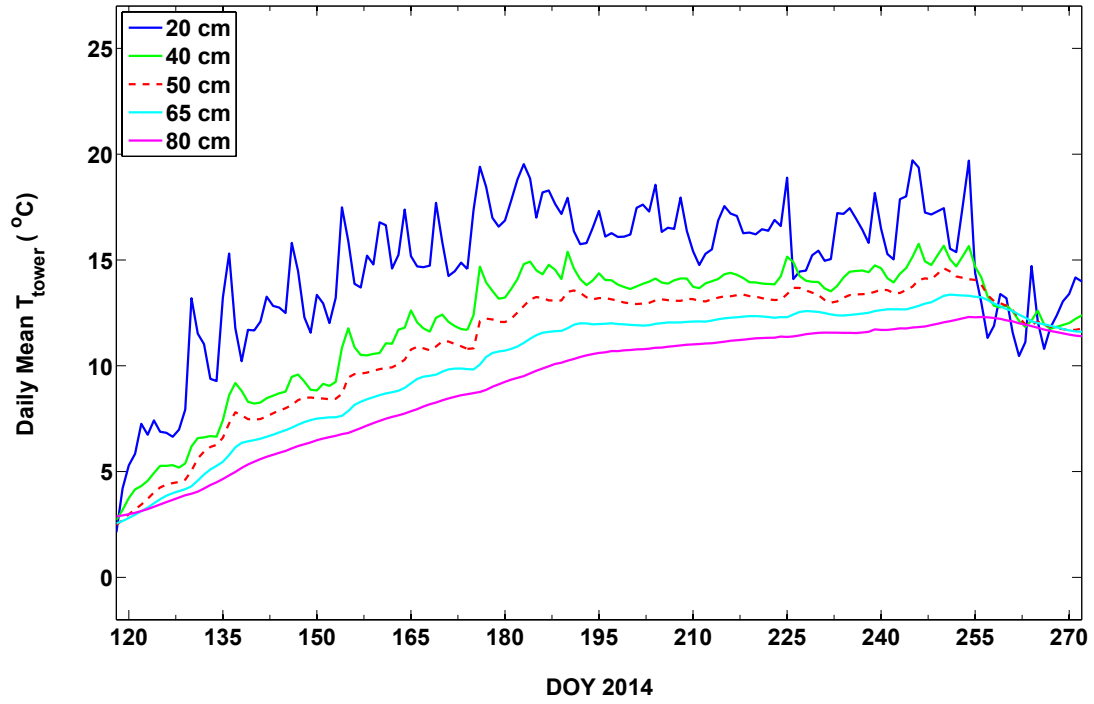


Figure 4-4: Daily mean peat temperature at the tower site ( $T_{\text{tower}}$ , °C) measured at 20, 40, 50, 65, and 80 cm below peat surface during the 2014 measurement period.

## 4.2 Methane fluxes

The *Chamaedaphne*-dominated autochamber community had the lowest mean daily CH<sub>4</sub> emissions compared to the *Ledum*- and *Eriophorum*-dominated communities (Figure 4-5). The *Eriophorum* community had the most temporally variable fluxes and by late July (~DOY 210) this community represented a CH<sub>4</sub> emission hot spot. The *Ledum* and *Eriophorum* communities' emissions demonstrated a clear seasonal increase to roughly mid-July (~ DOY 195) after which emission rates remained high until mid-September (~DOY 255). In contrast, the *Chamaedaphne* community's emissions increased more modestly with fluxes remaining relatively small. The manual static chamber sampling days, indicated by the open black circles in Figure 4-5, were representative of the overall trends for the *Chamaedaphne* and *Ledum* communities.

By splitting the season into late spring, early summer, mid-summer and late summer periods, differences in CH<sub>4</sub> fluxes between communities were apparent (Table 4-1). In particular, by mid- to late-summer, the *Ledum* and *Eriophorum* communities had mean daily CH<sub>4</sub> flux values that were at least 3 times greater than *Chamaedaphne* fluxes.

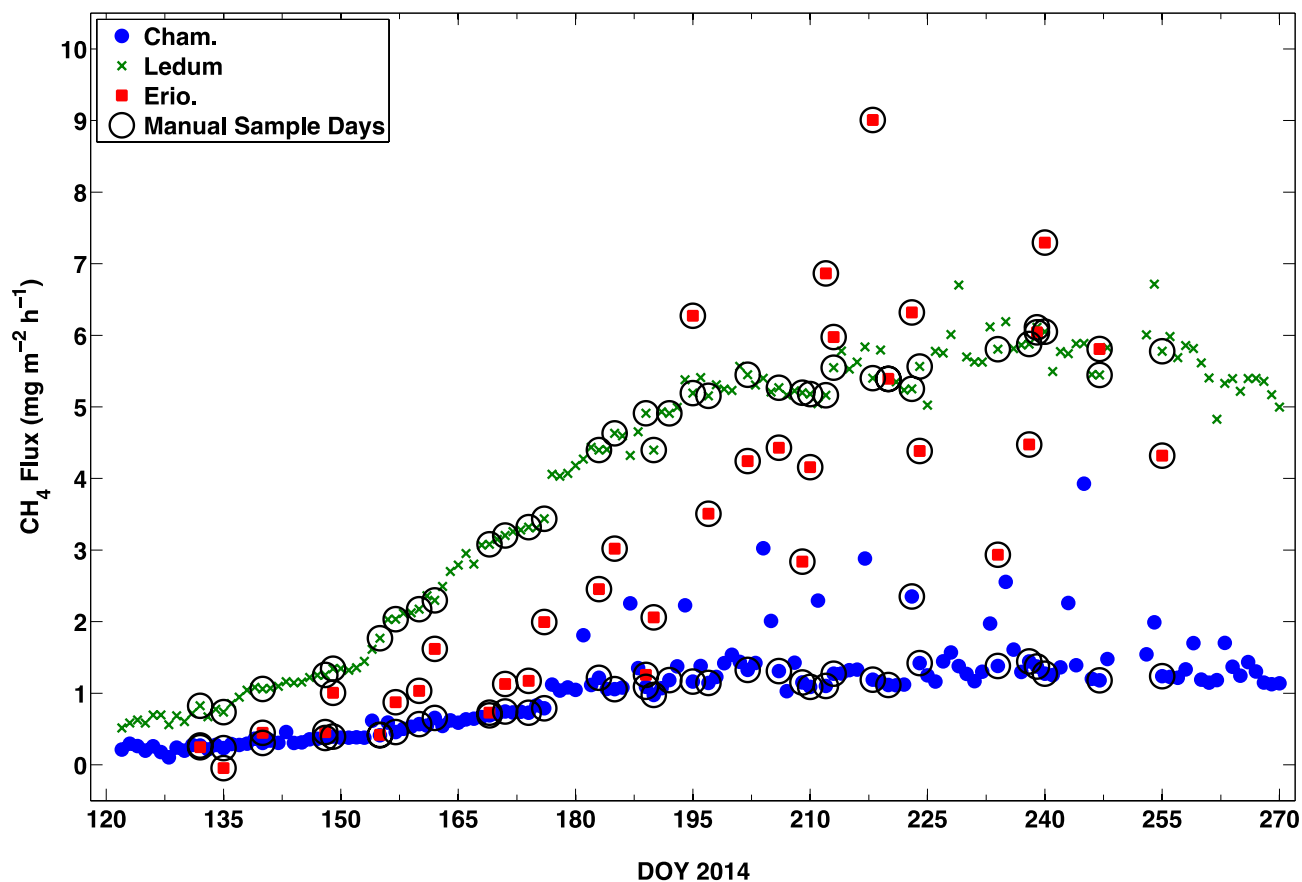


Figure 4-5: Mean daily CH<sub>4</sub> flux (mg m<sup>-2</sup> h<sup>-1</sup>) for *Chamaedaphne*, *Ledum* and *Eriophorum*-dominated communities at Mer Bleue during the 2014 measurement period (DOY 120-270). Open circles indicate days when manual static chamber flux sampling measurements were collected at the *Eriophorum* community chambers.

Table 4-1: Comparison of mean daily CH<sub>4</sub> flux, water table (WT) and 50 cm peat temperature at the tower site (T<sub>50cm</sub>) for *Chamaedaphne*, *Ledum* and *Eriophorum*-dominated communities at Mer Bleue, 2014. Standard deviation is given in parentheses. CH<sub>4</sub> fluxes with different superscripts across a row are significantly different at the  $\alpha=0.10$  level when accounting for the chamber as a random effect (or repeated measure) in a mixed model and using Student's *t*-test post-hoc analysis. Similarly, WT depths with different superscripts across a row are significantly different at the  $\alpha=0.10$  level when accounting for the chamber as a random effect (or repeated measure) in a mixed model and using Student's *t*-test post-hoc analysis.

<u>Chamaedaphne</u>		<u>Ledum</u>		<u>Eriophorum</u>		T <sub>50cm</sub> (°C)
CH <sub>4</sub> Flux (mg CH <sub>4</sub> m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	CH <sub>4</sub> Flux (mg CH <sub>4</sub> m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	CH <sub>4</sub> Flux (mg CH <sub>4</sub> m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	
Whole Period (DOY 132-255, n=891)						
0.99 <sup>b</sup> (0.82)	36.5 <sup>a</sup> (6.7)	4.13 <sup>a</sup> (2.13)	28.8 <sup>b</sup> (3.7)	2.95 <sup>ab</sup> (3.57)	28.5 <sup>b</sup> (3.6)	11.1 (3.0)
Late Spring (DOY 132-154, n=129)						
0.32 <sup>a</sup> (0.30)	35.2 <sup>a</sup> (7.0)	1.04 <sup>b</sup> (0.39)	27.3 <sup>a</sup> (3.6)	0.52 <sup>NA</sup> (0.51)	N/A	6.6 (1.8)
Early Summer (DOY 155-189, n=281)						
0.77 <sup>b</sup> (0.54)	36.0 <sup>a</sup> (7.2)	3.21 <sup>a</sup> (1.26)	28.3 <sup>ab</sup> (4.0)	1.39 <sup>b</sup> (1.45)	27.5 <sup>b</sup> (3.7)	11.3 (1.2)
Mid Summer (DOY 190-221, n=223)						
1.17 <sup>b</sup> (0.52)	37.0 <sup>a</sup> (6.5)	5.19 <sup>a</sup> (1.53)	29.7 <sup>b</sup> (3.3)	4.46 <sup>ab</sup> (4.15)	29.2 <sup>b</sup> (3.7)	13.2 (0.2)
Late Summer (DOY 222-255, n=185)						
1.46 <sup>b</sup> (1.24)	37.1 <sup>a</sup> (6.2)	5.74 <sup>a</sup> (1.82)	29.2 <sup>b</sup> (3.4)	5.12 <sup>a</sup> (4.11)	29.2 <sup>b</sup> (2.8)	13.6 (0.4)

From a chamber-specific perspective, CH<sub>4</sub> emissions for the *Ledum* were greatest for autochambers L-A 2 and 3 (Figure 4-6). While classified as part of the *Ledum* community, L-A 2 and 3 had *Eriophorum* plants within the chamber collar that constituted 74% and 68% of total LAI, respectively. In contrast, autochamber L-A 1 had a tightly constrained CH<sub>4</sub> emission pattern with moderate fluxes at about half the emission rates of the L-A 2 and 3 autochambers. E-M 1 was the driver of the highest daily *Eriophorum* fluxes, while the other two *Eriophorum* plots were characterized by less frequently elevated flux values. Autochambers C-A 1, 2, and 3, predominantly composed of *Chamaedaphne* shrubs, had the lowest CH<sub>4</sub> emissions despite sporadic higher flux events from C-A 1. While these nine chambers were selected from a relatively small sampling area within the peatland, the spatial variability of summer CH<sub>4</sub> fluxes was slightly larger than the temporal variability (Figure 4-6). For example, for any given manual flux sampling date, the coefficient of variation ranged from 52-102% (as a measure of spatial variation), while the coefficient of variation ranged from 28-87% for a given chamber across all 36 sampling dates (as a measure of temporal variation). Greatest temporal coefficients of variation were associated with the *Eriophorum* community plots (73 – 87%) compared to the *Ledum* (28 – 49%) and *Chamaedaphne* (45– 66%) plots.

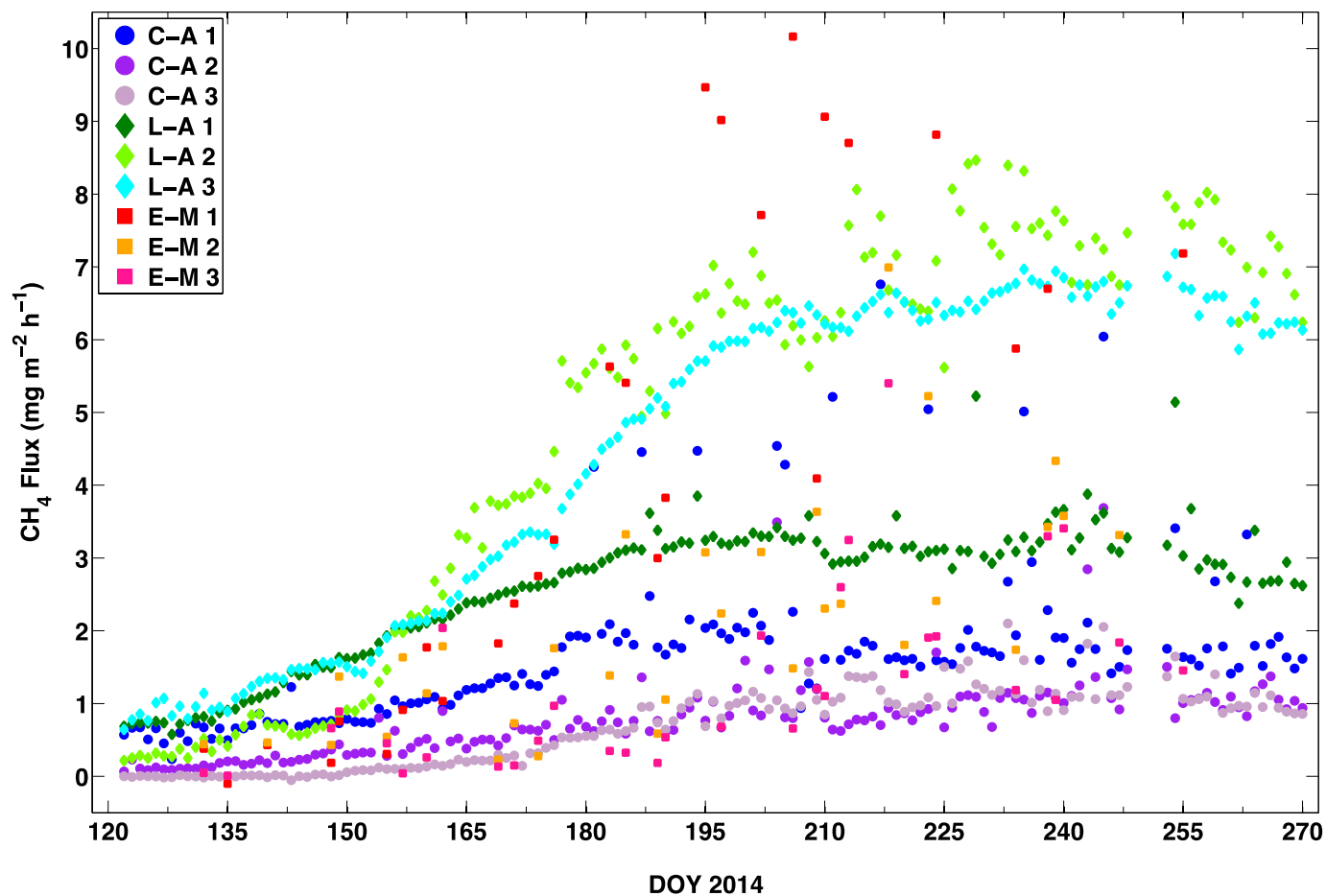


Figure 4-6: Mean daily CH<sub>4</sub> flux (mg m<sup>-2</sup> h<sup>-1</sup>) emitted from 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3) (circles), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) (diamonds) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) (squares) at Mer Bleue during the 2014 measurement period (DOY 120-270).



To better understand the factors influencing spatial and temporal variations in CH<sub>4</sub> emissions, the log transform of CH<sub>4</sub> fluxes for manually sampled days were analyzed using a mixed model with community as a fixed effect, chamber as a random effect (to account for repeated measures on the same location over time) and with the following variables as covariates: barometric pressure, 10 cm peat temperature and WT from each chamber, and 50 cm peat temperature from the tower location. There was no statistically significant effect of 10 cm peat temperature as a covariate of CH<sub>4</sub> fluxes ( $F(1,794)=2.6170, p=0.1061$ ). For example, the *Chamaedaphne* community plots' 10 cm peat temperatures varied greatly while there was a relatively restricted range of low CH<sub>4</sub> fluxes (Figure 4-7). There was a smaller temperature range and wider range of measured CH<sub>4</sub> fluxes for the *Ledum* community. The *Eriophorum* community demonstrated greatest variability among chambers in both temperature and CH<sub>4</sub> fluxes with greatest fluxes for E-M 1 despite little temperature variation at 10 cm while E-M 2 reached the highest temperatures but did not produce the greatest CH<sub>4</sub> fluxes. The day-to-day variation in 10 cm peat temperature obscures the relationship with CH<sub>4</sub> flux, such that peat temperature at 10 cm depth is not useful for analyzing CH<sub>4</sub> flux patterns (Figure 4-7). However, deeper peat temperatures 40-50 cm below the surface have been successfully applied in an examination of controls on CH<sub>4</sub> emissions (Moore *et al.* 2011, Lai *et al.* 2014a). There was a statistically significant effect of 50 cm peat temperature ( $F(1,791.8)=523.8658, p<0.0001$ ) with a significant interaction effect of vegetation community and 50 cm peat temperature ( $F(2,791.1)=62.7842, p<0.0001$ ). There was a positive relationship between 50 cm peat temperature and CH<sub>4</sub> flux (Figure 4-8) suggesting that peat temperature near the WT may be an important environmental control

on CH<sub>4</sub> emissions although the relationship between CH<sub>4</sub> flux and 50 cm peat temp varies among the vegetative communities.

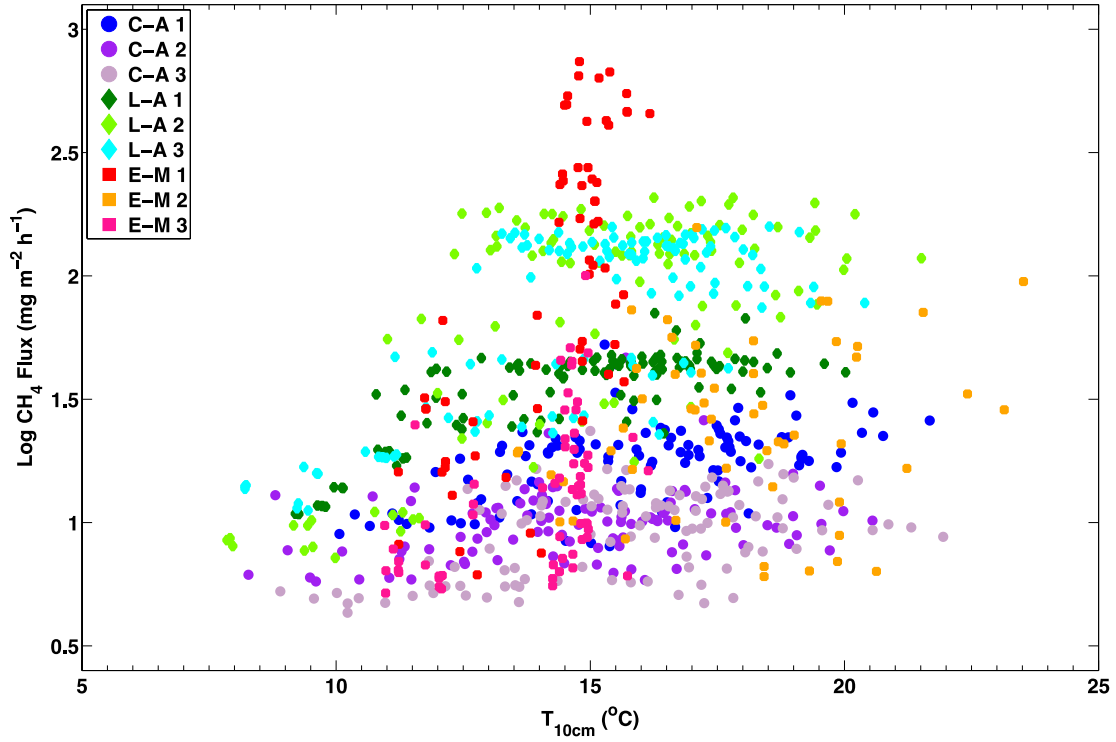


Figure 4-7: The relationship between 10 cm peat temperature ( $T_{10\text{cm}}$ , °C) and log CH<sub>4</sub> flux ( $\text{mg m}^{-2} \text{h}^{-1}$ ) emitted from 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3) (circles), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) (diamonds) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) (squares) at Mer Bleue for manual sampling day data (DOY 132-255).

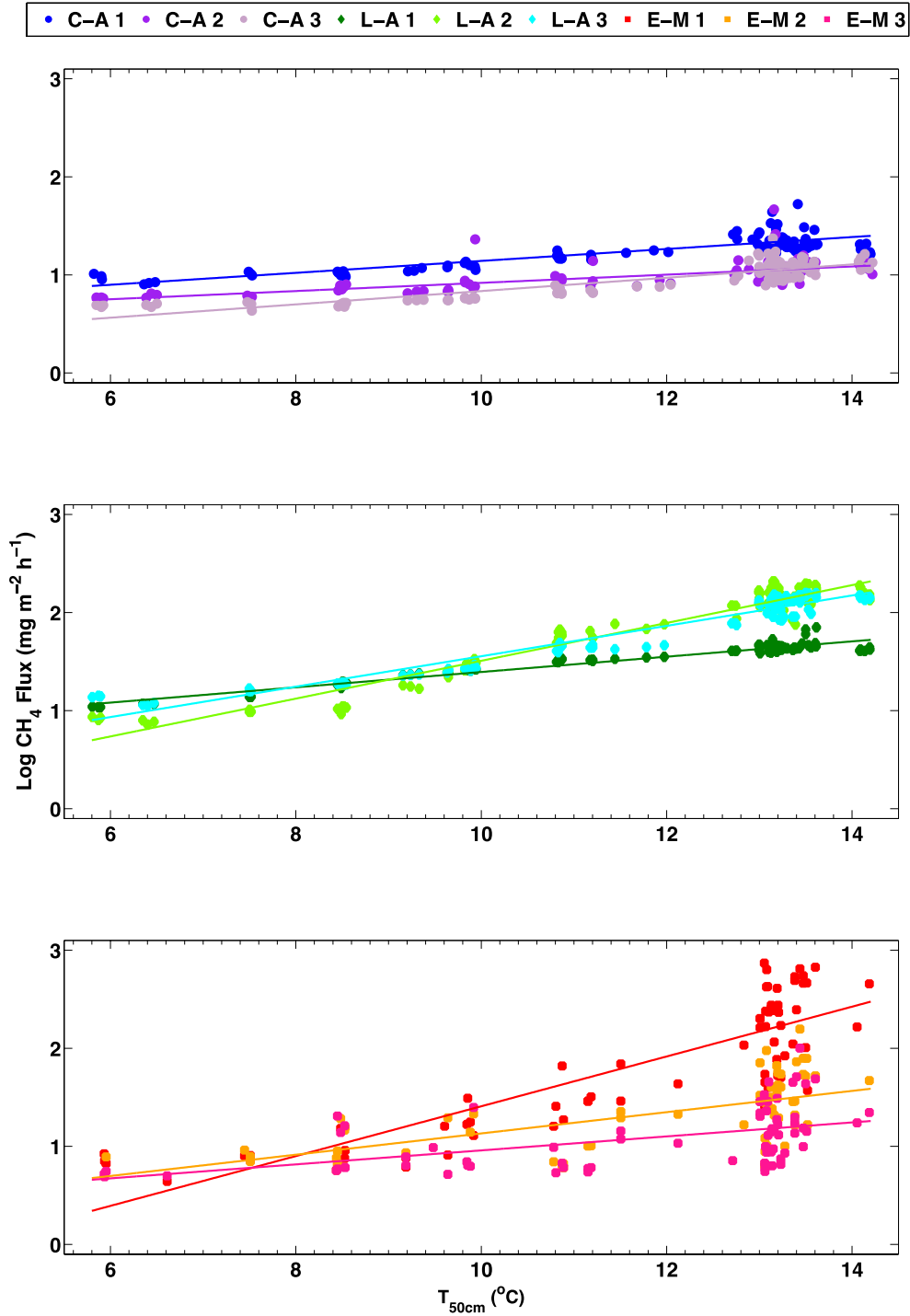


Figure 4-8: The relationship between 50 cm peat temperature ( $T_{50\text{cm}}$ , °C) and log  $\text{CH}_4$  flux ( $\text{mg m}^{-2} \text{h}^{-1}$ ) emitted from 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3) (circles), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) (diamonds) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) (squares) at Mer Bleue for manual sampling day data (DOY 132-255).

Overall, as WT rose, CH<sub>4</sub> flux increased when examining all daily chamber flux values (Figure 4-9) and all seasonal averages (Figure 4-10). As a covariate, WT had a significant relationship with CH<sub>4</sub> fluxes for manual sampling day data ( $F(1,793.8)=5.7958, p=0.0163$ ). For the seasonal averages, the significant relationship between log CH<sub>4</sub> flux and chamber-specific WT is  $2.435 \text{ mg m}^{-2} \text{ h}^{-1} + 0.03162 (\text{WT}) \text{ mg m}^{-2} \text{ h}^{-1} \text{ cm}^{-1}$  ( $F(1,816) = 169.77, p < 0.001, R^2 = 0.172$ ).

However, for a given chamber, there was little relationship with WT, which was expected given the small range in WT variation through the growing season. As a result, microtopography effects on WT drove the spatial variations observed in Figure 4-9 and Figure 4-10. For example, C-A 2 (a *Chamaedaphne* shrub hummock) had the lowest seasonal WT and lowest CH<sub>4</sub> fluxes, while L-A 3 and E-M 1 had WT closer to the surface and produced some of the largest CH<sub>4</sub> fluxes (Figure 4-10).

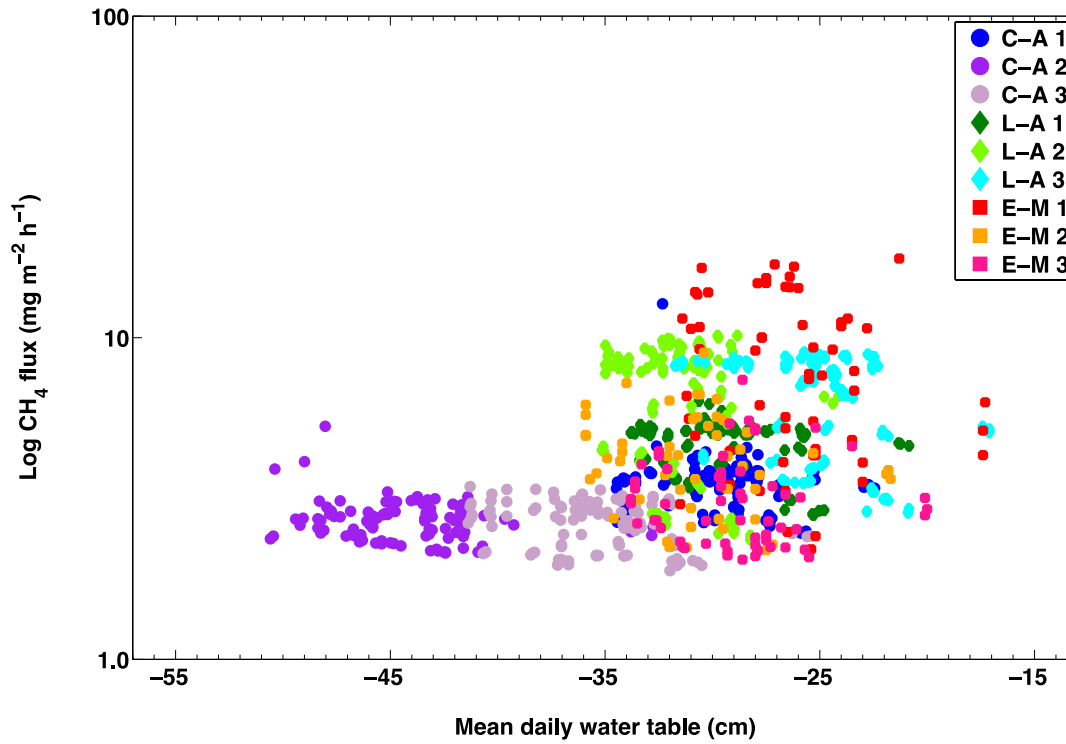


Figure 4-9: Relationship between mean depth to water table (cm) and log CH<sub>4</sub> flux (mg m<sup>-2</sup> h<sup>-1</sup>) from 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3) (circles), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) (diamonds) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) (squares) at Mer Bleue for manual static chamber sampling days (DOY 132-255) during the 2014 measurement period.

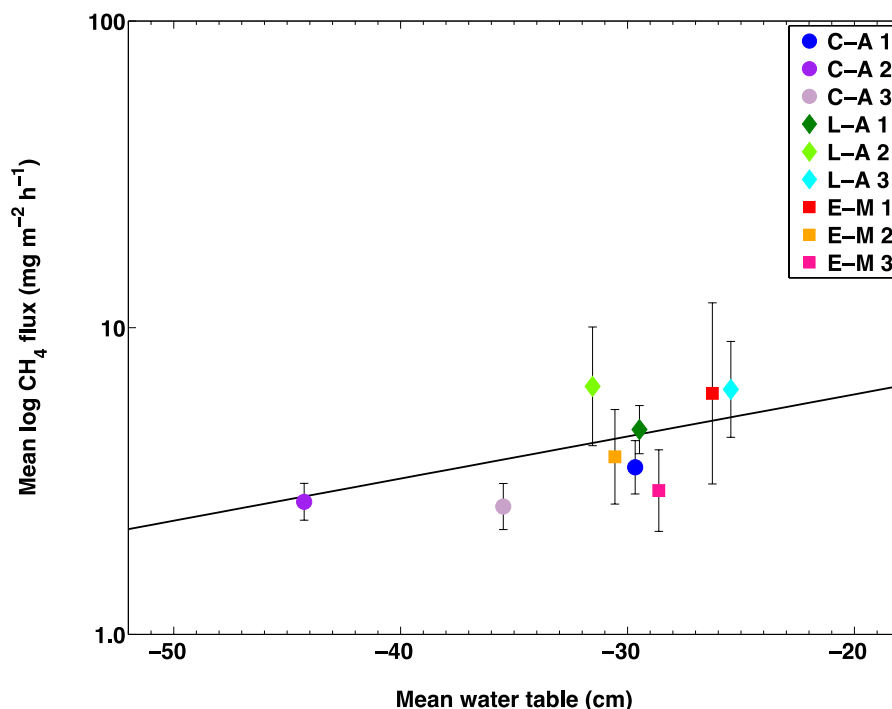


Figure 4-10: Relationship between mean seasonal depth to water table (cm) and mean ( $\pm$ SD) seasonal log CH<sub>4</sub> flux (mg m<sup>-2</sup> h<sup>-1</sup>) from 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3) (circles), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) (diamonds) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) (squares) at Mer Bleue for manual static chamber sampling days (DOY 132-255) 2014. The solid black line shows the linear regression calculated to predict log CH<sub>4</sub> flux based on mean daily water table, where predicted log CH<sub>4</sub> flux is equal to  $2.435 + 0.03162$  (WT) mg m<sup>-2</sup> h<sup>-1</sup> when WT is measured in cm from the peat surface at each chamber.

### 4.3 Redox potential measurements

Examples of redox potential measurements recorded at E-M 1 and E-M 2 are presented in Figure 4-11 and Figure 4-12 respectively. As the boundary between reduction and oxidation conditions is controlled by the position of the WT, the mean daily depth to WT is important to consider. Both chambers had WT below 20 cm for the late summer–fall sampling period. This is reflected by the average standardized redox

potential value of ~600 mV at 20 cm, which indicates aerated conditions and the predominance of O<sub>2</sub> (Table 2-1). In both of these cases, redox potential reached the most negative values at a depth of 40 cm – the depth closest to but still below the WT. At E-M 1, the 40 cm Eh value of -200 mV suggests the activity of methanogens for the reduction of CO<sub>2</sub> to CH<sub>4</sub>. At E-M 2, the redox potential at 40 cm is similar to that at 60 and 80 cm (-100 to 0 mV), suggesting a greater proportion of alternate terminal electron acceptors higher up the cascade of redox pairs in the soil, such as sulfate (SO<sub>4</sub><sup>2-</sup>) and ferric iron (Fe<sup>3+</sup>) (Table 2-1).

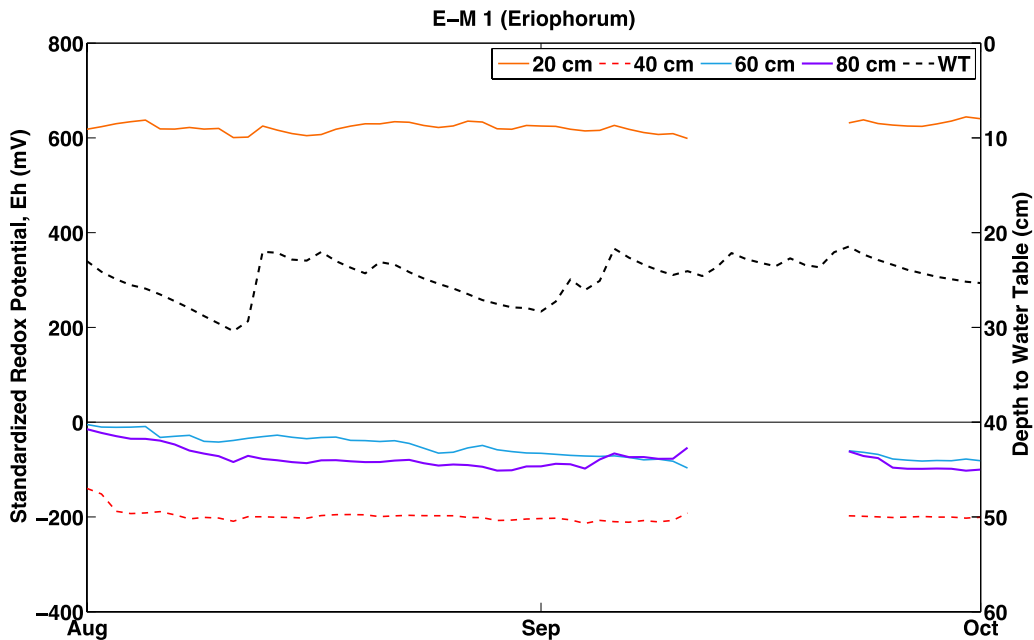


Figure 4-11: Mean daily redox potential (mV) referenced to Ag/AgCl reference electrode with 3 M KCl filling solution and corrected for pH of 4. Sampled at chamber E-M 1 (*Eriophorum*) for peat depths of 20cm, 40 cm, 60 cm, and 80 cm at Mer Bleue, 2014. Mean daily depth to water table (cm) is shown for reference (dotted black line) on the secondary y-axis.

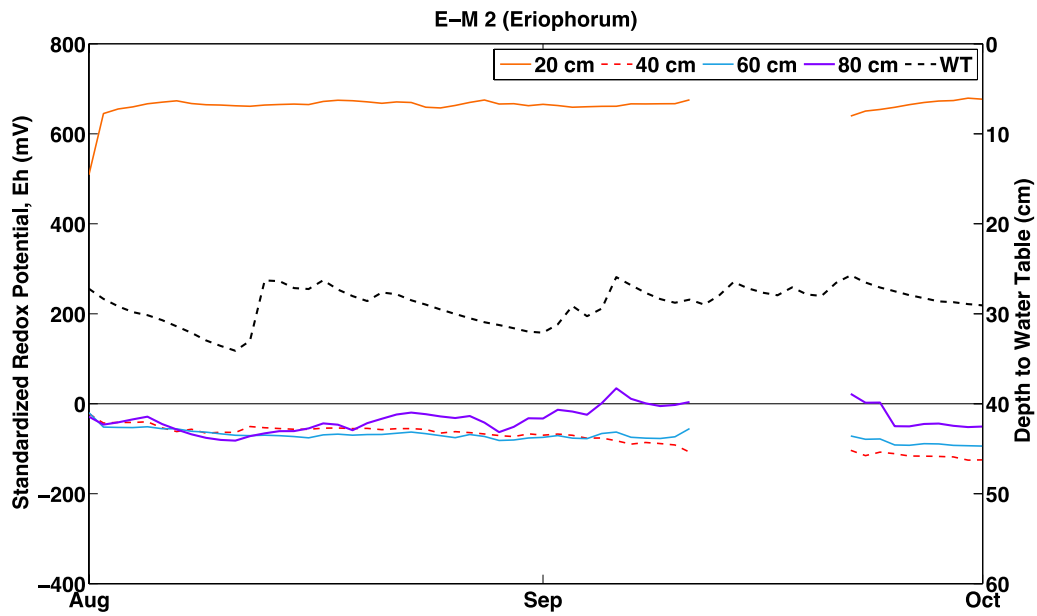


Figure 4-12: Mean daily redox potential (mV) referenced to Ag/AgCl reference electrode with 3 M KCl filling solution and corrected for pH of 4. Sampled at chamber E-M 2 (*Eriophorum*) for peat depths of 20cm, 40 cm, 60 cm, and 80 cm at Mer Bleue, 2014. Mean daily depth to water table (cm) is shown for reference (dotted black line) on the secondary y-axis.



#### 4.4 Pore water analyses

For temporal pore water analyses, DOC ( $\text{mg L}^{-1}$ ), TDN ( $\text{mg L}^{-1}$ ),  $\text{SUVA}_{254}$  ( $\text{L m}^{-1} \text{mg}^{-1} \text{DOC}$ ) and C/N ratio were averaged across all chambers for four sampling depths and presented in Figure 4-13. There was an increasing seasonal concentration of DOC and TDN observed in pore water (Figure 4-13). When DOC concentrations were lowest at the beginning of the summer,  $\text{SUVA}_{254}$  was at its highest.

The pore water C/N ratio did not have as much seasonal variation as both DOC and TDN increased throughout the summer, yet it did appear to mimic the variation in peat C/N ratio, which decreases with depth (Hornibrook *et al.* 2000b, Malmer & Wallén 2004, Rydin & Jeglum 2013, Wang *et al.* 2015). By constructing the same figure using data from *Eriophorum* plots only (Figure 4-14) the patterns have much greater variation over the season and with depth. In addition, the trend for these variables in the *Eriophorum* plots at 50 cm did not clearly increase or decrease over the season.

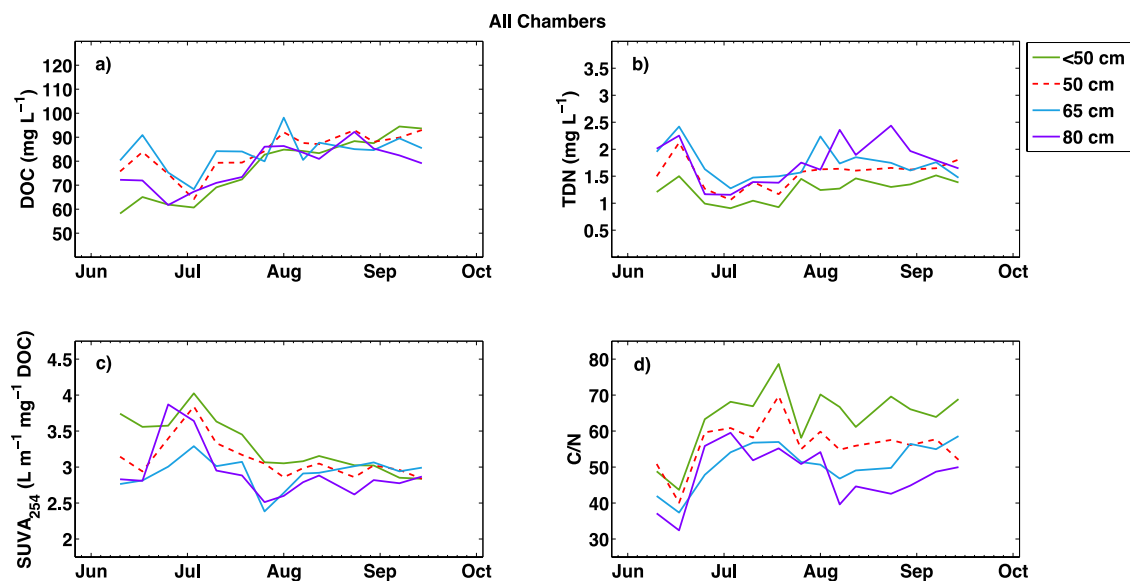


Figure 4-13: Mean daily (a) dissolved organic carbon (DOC, mg L<sup>-1</sup>), (b) total dissolved nitrogen (TDN, mg L<sup>-1</sup>), (c) specific ultraviolet absorbance at 254 nm (SUVA<sub>254</sub>, L m<sup>-1</sup> mg<sup>-1</sup> DOC), and (d) carbon to nitrogen ratio (C/N) for all chambers at various depths during the 2014 measurement period.

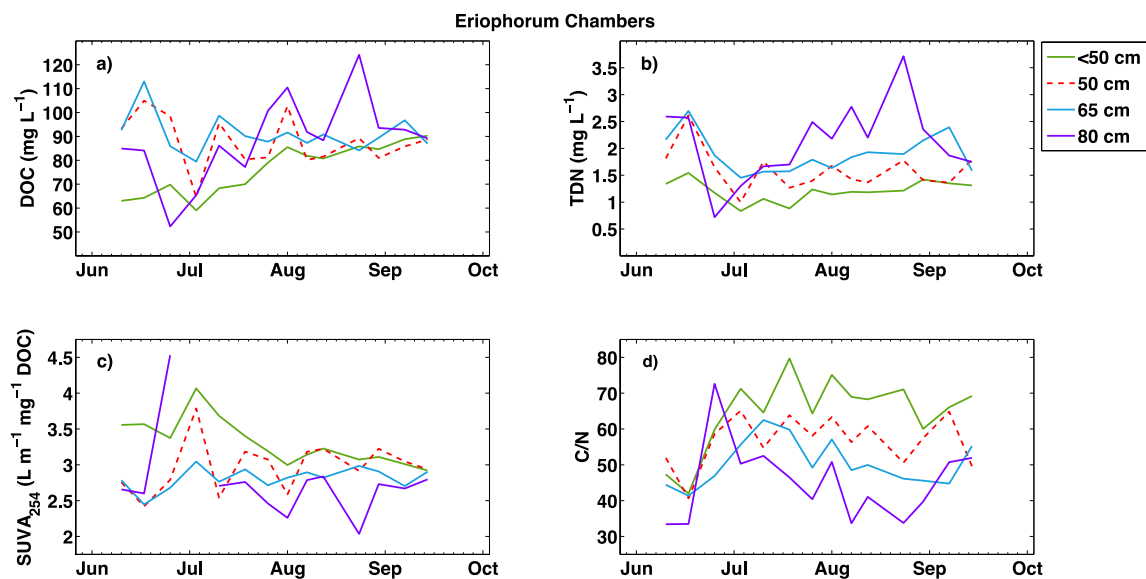


Figure 4-14: Mean daily (a) dissolved organic carbon (DOC, mg L<sup>-1</sup>), (b) total dissolved nitrogen (TDN, mg L<sup>-1</sup>), (c) specific ultraviolet absorbance at 254 nm (SUVA<sub>254</sub>, L m<sup>-1</sup> mg<sup>-1</sup> DOC), and (d) carbon to nitrogen ratio (C/N) for *Eriophorum*-dominated chambers at various depths during the 2014 measurement period.

Pore water CH<sub>4</sub> concentrations generally increased with depth to a peak at 50 cm during the late summer season. Samples obtained 20 cm below peat surface, which was in the aerobic zone above the WT, had small, near-ambient CH<sub>4</sub> concentrations. However, a few higher CH<sub>4</sub> concentrations (in the range of 100-230  $\mu\text{mol L}^{-1}$ ) were observed at 20 cm, particularly for the *Chamaedaphne* community in late summer. This could provide evidence for ebullition events (Figure 4-15). Pore water CH<sub>4</sub> was analyzed using a mixed model with chamber as a random effect and DOC, TDN, depth-specific peat temperature and WT as covariates, and depth and community as fixed effects. There was no statistically significant effect of community ( $F(2, 8.53)=1.4098, p=0.2958$ ) or sampling depth ( $F(1, 445.2)=0.4707, p=0.4930$ ) on pore water CH<sub>4</sub> concentrations over the entire sampling period (DOY 155-255). However, when only later summer sampling days were considered (DOY 212-255), there was a statistically significant effect of community ( $F(2, 9.669)=4.7589, p=0.0364$ ) where *Chamaedaphne* had significantly higher pore water CH<sub>4</sub> concentrations than the *Ledum* community.

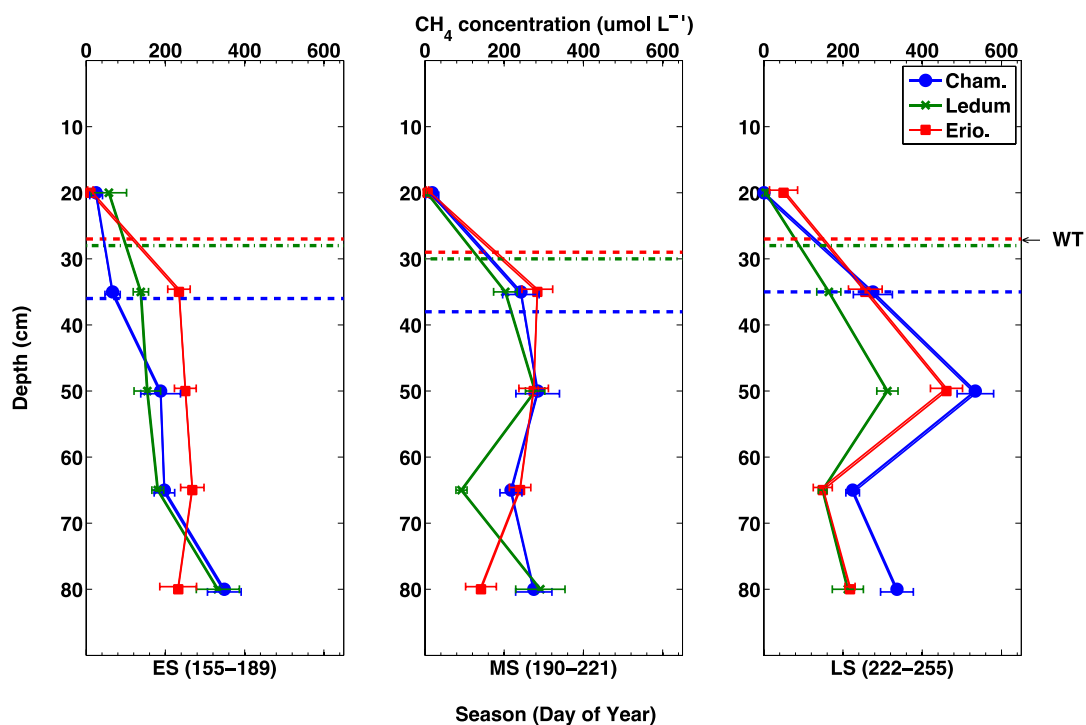


Figure 4-15: Pore water profiles of mean ( $\pm$  SE) dissolved  $\text{CH}_4$  concentrations ( $\mu\text{mol L}^{-1}$ ) for three vegetation communities at Mer Bleue. Depth was measured from peat surface (cm). Horizontal lines indicate mean depth to water table (cm) for each community. Note: ES, early summer (DOY 155-189); MS, mid-summer (DOY 190-221); LS, late summer (DOY 222-255) for the 2014 measurement period.

Only at 50 cm was there a tendency for pore water CH<sub>4</sub> concentrations to increase through the growing season (Figure 4-16). This trend seemed to be largely driven by several high pore water CH<sub>4</sub> concentrations sampled in late August and early September. To further investigate the importance of the 50 cm depth, CH<sub>4</sub> concentrations (μmol L<sup>-1</sup>) at 50 cm were analyzed on a plot basis (Figure 4-17). Methane concentrations appeared to increase over the summer, but there was large variability among chambers. In particular, C-A 3 fluctuated substantially but reached the highest 50 cm CH<sub>4</sub> concentrations of all chambers by late summer. The least variability among the *Ledum* chambers was L-A 3. E-M 1 was variable but produced the largest CH<sub>4</sub> concentrations among the *Eriophorum* chambers for most of the summer. Collectively, the largest increase in CH<sub>4</sub> concentration at 50 cm over time was in the *Chamaedaphne* plots.

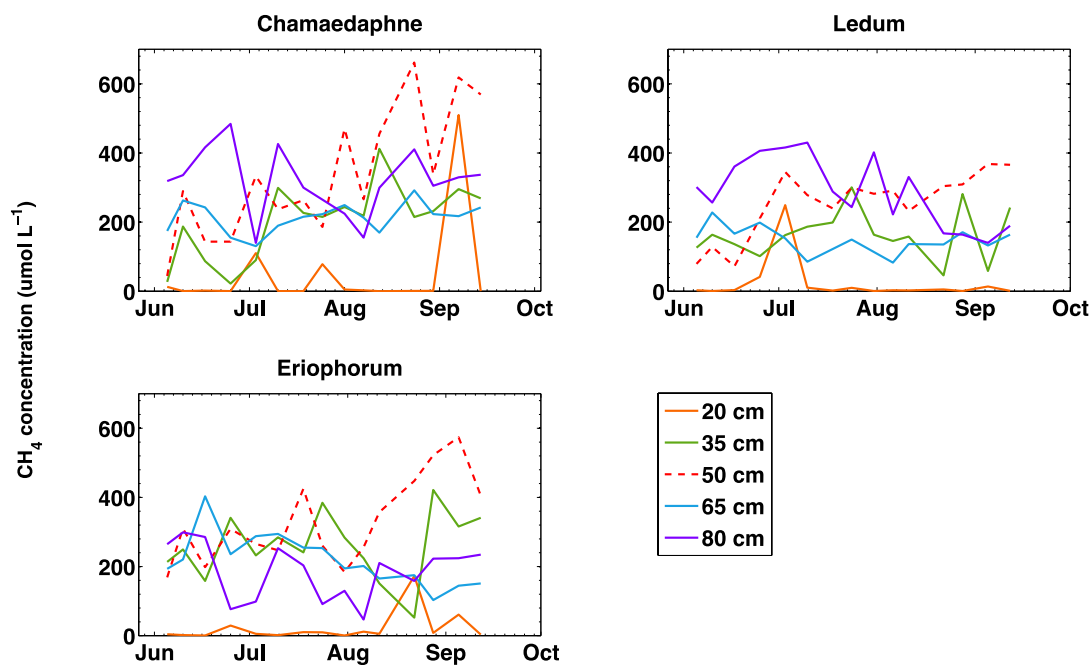


Figure 4-16: Seasonal dissolved pore water  $\text{CH}_4$  concentrations ( $\mu\text{mol L}^{-1}$ ) at 20, 35, 50, 65 and 80 cm for *Chamaedaphne*, *Ledum* and *Eriophorum*-dominated communities at Mer Bleue during 2014.

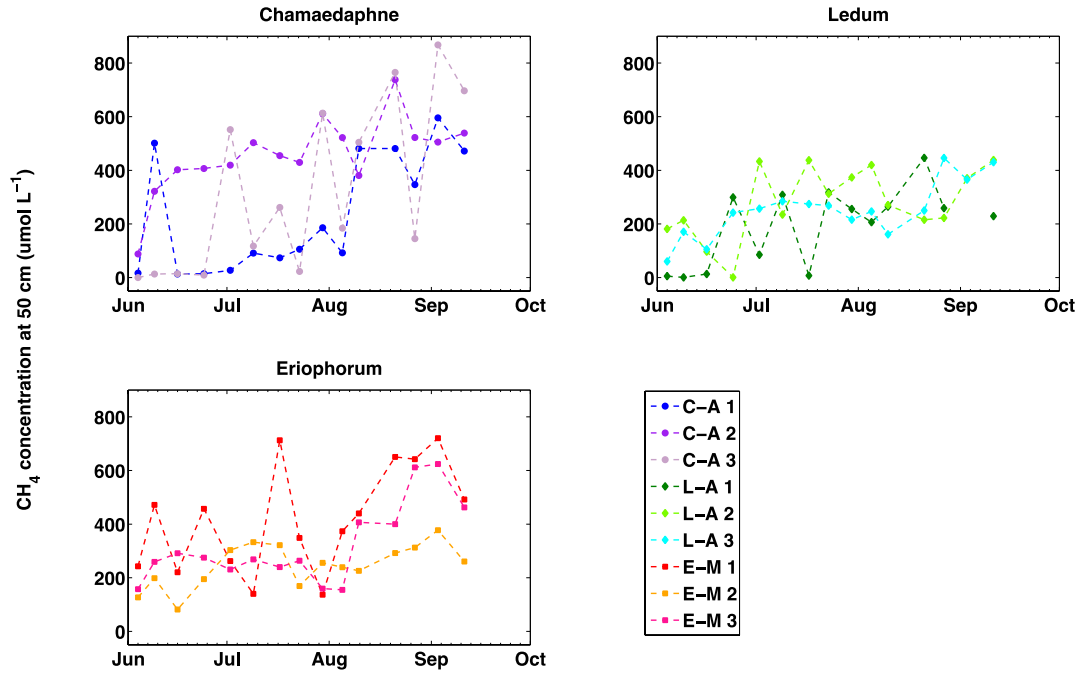


Figure 4-17: Seasonal dissolved pore water CH<sub>4</sub> concentrations (μmol L<sup>-1</sup>) at 50 cm depth for 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3) (circles), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) (diamonds) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) (squares) at Mer Bleue during the 2014 measurement period.



#### 4.5 CH<sub>4</sub> production pathway

CH<sub>4</sub> production pathways were investigated using the Keeling Plot approach to discern the dominant CH<sub>4</sub> production pathway (Figure 4-18). The  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signatures were restricted to a narrow range of depleted values ( $\sim -81$  to  $-71$  ‰) for all chambers. Taking into account the random effect of chamber (as a repeated measure), there was no statistically significant effect of community on  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signatures ( $F(2, 5.668)=0.2107, p=0.8161$ ). The  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signatures for all three communities provided evidence for the CO<sub>2</sub> reduction pathway of CH<sub>4</sub> production, which produces a relatively  $^{13}\text{C}$ -depleted signature ( $-110$  to  $-60$  ‰) compared to the acetate fermentation ( $-65$  to  $-50$  ‰) pathway (Whiticar *et al.* 1986) (Figure 4-18).

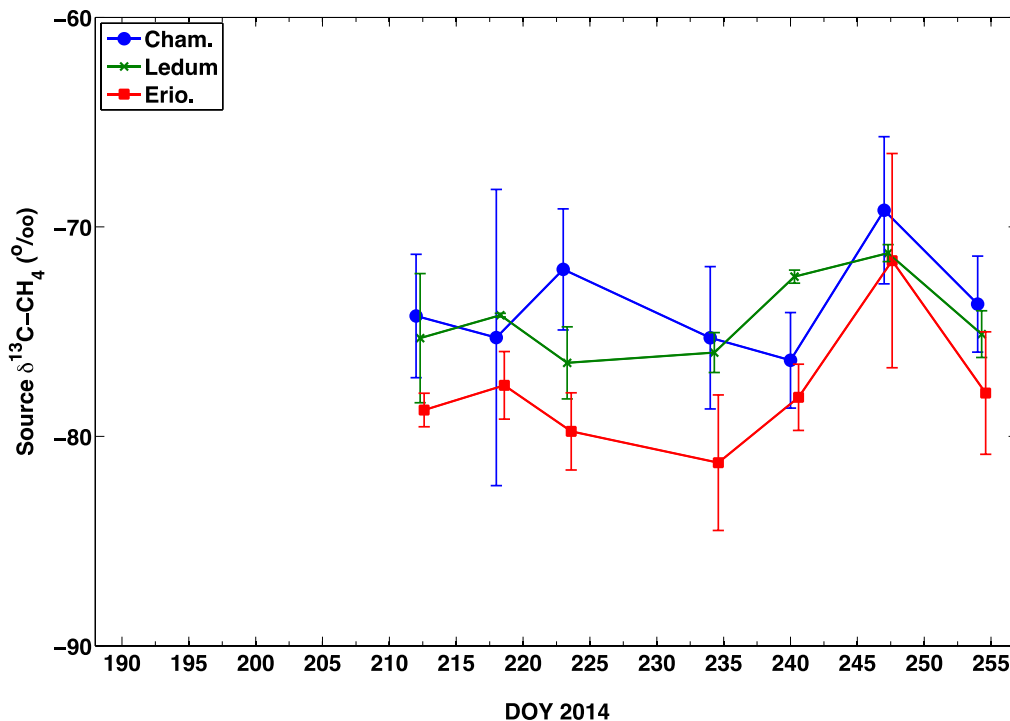


Figure 4-18: Mean ( $\pm$  SD)  $\delta^{13}\text{C}$ -CH<sub>4</sub> isotopic signatures (‰) for CH<sub>4</sub> emitted from *Chamaedaphne*, *Ledum* and *Eriophorum*-dominated communities at Mer Bleue during the mid- to late-summer 2014 measurement period.

#### 4.6 Estimates of seasonal CH<sub>4</sub> storage change

The change in pore water CH<sub>4</sub> storage over time was explored with a focus on multiple peat depths and combinations of sampling depths along with different time periods. Week-to-week analyses were confounded by high variability in pore water CH<sub>4</sub> concentrations, while the temporal variations in pore water CH<sub>4</sub> at 50 cm dominated the storage term. Changes in storage of CH<sub>4</sub> computed using only the 50 cm depth (over a 15 cm layer) over the entire pore water sampling period revealed the most distinct differences among communities. C-A 3 (*Chamaedaphne*) had the largest increase in CH<sub>4</sub> storage ( $0.71 \text{ mg m}^{-2} \text{ h}^{-1}$ ) at the 50 cm depth (Figure 4-19). C-A 1 (*Chamaedaphne*) also had a relatively large increase in CH<sub>4</sub> storage ( $0.44 \text{ mg m}^{-2} \text{ h}^{-1}$ ). However, the remaining *Chamaedaphne* autochamber, C-A 2, had only moderate CH<sub>4</sub> storage change ( $0.34 \text{ mg m}^{-2} \text{ h}^{-1}$ ) in comparison to the other two *Chamaedaphne* chambers over the season. This chamber's deep WT position (~ 45 cm below peat surface) and the corresponding thick aerobic peat layer, which provides increased potential for CH<sub>4</sub> consumption, likely contributed to this difference. E-M 2 demonstrated the smallest change in CH<sub>4</sub> storage ( $0.16 \text{ mg m}^{-2} \text{ h}^{-1}$ ). The remaining chambers showed moderate CH<sub>4</sub> storage increase over the season ranging from 0.20 to  $0.33 \text{ mg m}^{-2} \text{ h}^{-1}$ .

A linear relationship between CH<sub>4</sub> storage change at 50 cm and mean CH<sub>4</sub> flux for manual sampling days during the pore water collection period was not significant (Figure 4-20) ( $r = -0.4438$ ,  $N=9$ ,  $p=0.2315$ ). However, an exponential function fit the data (Figure 4-20) ( $p = 0.0024$ ), reflecting the pattern of greater fluxes with less change in CH<sub>4</sub> storage in the pore water. For example, the chamber with the greatest CH<sub>4</sub> storage change

had very low fluxes (C-A 3) and some of the highest emitters (L-A 2, 3) had low CH<sub>4</sub> storage change.

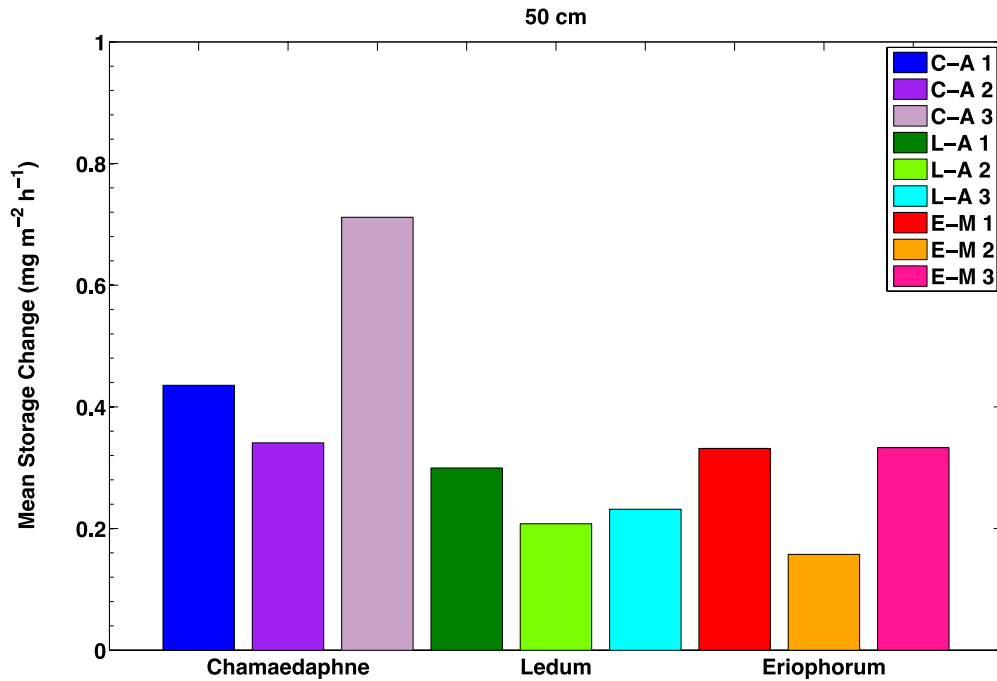


Figure 4-19: Mean CH<sub>4</sub> storage change (mg m<sup>-2</sup> h<sup>-1</sup>) for 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) at the 50 cm depth (15 cm thickness) during the pore water sampling period (DOY 155-255).

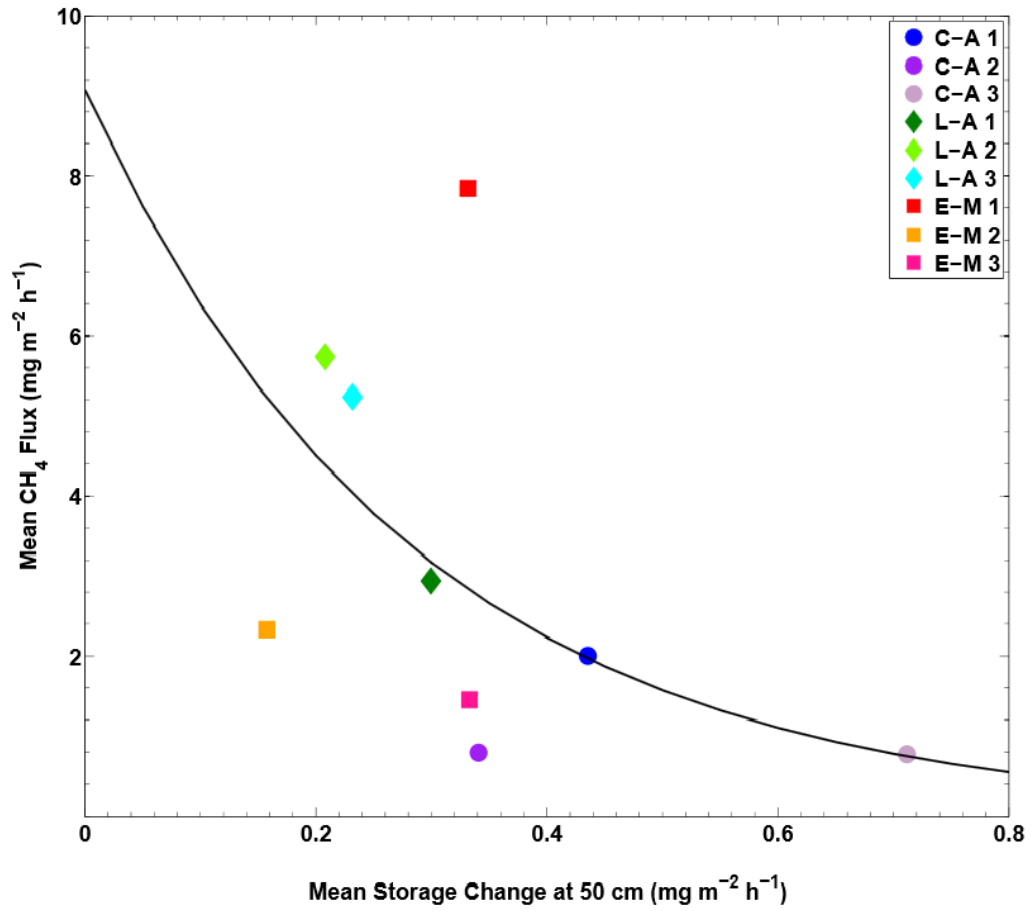


Figure 4-20: Relationship between mean storage change at 50 cm ( $\text{mg m}^{-2} \text{h}^{-1}$ ) and mean  $\text{CH}_4$  flux ( $\text{mg m}^{-2} \text{h}^{-1}$ ) for 3 *Chamaedaphne*-dominated autochambers (C-A 1, 2, 3) (circles), 3 *Ledum*-dominated autochambers (L-A 1, 2, 3) (diamonds) and 3 *Eriophorum*-dominated manual chambers (E-M 1, 2, 3) (squares) at Mer Bleue for manual sampling days during the pore water collection period 2014 (DOY 155-255). The solid black line shows the nonlinear regression calculated to predict mean  $\text{CH}_4$  flux based on storage change at 50 cm, where predicted mean  $\text{CH}_4$  flux ( $\text{mg m}^{-2} \text{h}^{-1}$ ) is equal to  $9.08 \times 0.03^{S_{\text{CH}_4}}$  where  $S_{\text{CH}_4}$  is storage change measured in  $\text{mg m}^{-2} \text{h}^{-1}$ .

#### 4.7 Relationships among indicators of CH<sub>4</sub> production, storage and transport

Overall, on a plot scale during the mid- to late-summer periods when all variables were available for analysis, vegetation community type can be related to a number of variables associated with the production, storage and transport of CH<sub>4</sub> (Figure 4-21 and Table 4-2). Although vegetation classification is subject to the user's classification scheme, *Eriophorum vaginatum*, as an important species with mechanistic implications to CH<sub>4</sub> transport and production, correlated strongly to community type (where *Chamaedaphne* = community 1, *Ledum* = community 2, and *Eriophorum* = community 3). More *Eriophorum* was associated with shallower WT, greater CH<sub>4</sub> emissions and more depleted  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signature (Figure 4-21 and Table 4-2). These variables tended to be associated with the first principle component, which explained 38.1% of the variation in the data (Figure 4-21). However, neither community nor % LAI attributed to *Eriophorum vaginatum* (EV%) correlated with 50 cm pore water characteristics (Figure 4-21 and Table 4-2). The second principle component explained 28.1% of the variance and tended to associate with below ground characteristics. For this set of variables, a measure of biomass (LAI-2000) correlated with some of these pore water characteristics (positive correlation with TDN and negative correlation with C/N). Many of these pore water characteristics also closely related to each other. For example, greater TDN was associated with lower SUVA<sub>254</sub> and C/N, as was also demonstrated with the temporal and spatial (depth) trends in Figure 4-13. Only WT weakly correlated with pore water CH<sub>4</sub> concentrations (p-value = 0.07) with deeper WT having greater CH<sub>4</sub> concentrations (characteristic of the non-*Eriophorum* plots although not correlating to any vegetation metrics). As with previous analyses, there was no significant correlation ( $p < 0.05$ )

between CH<sub>4</sub> storage change (calculated from pore water CH<sub>4</sub> from DOY 155 to 255) and any measure of community type or any other variable. However, there was a weak negative relationship ( $p = 0.09$ ) between storage change and community type, where more sedge has less storage change and a negative relationship between storage change and flux ( $p = 0.09$ ). Similarly, weak relationships between total LAI (perhaps as a proxy for plant productivity) and pore water characteristics included a negative correlation with DOC ( $p = 0.08$ ) and a positive correlation with SUVA ( $p = 0.09$ ).

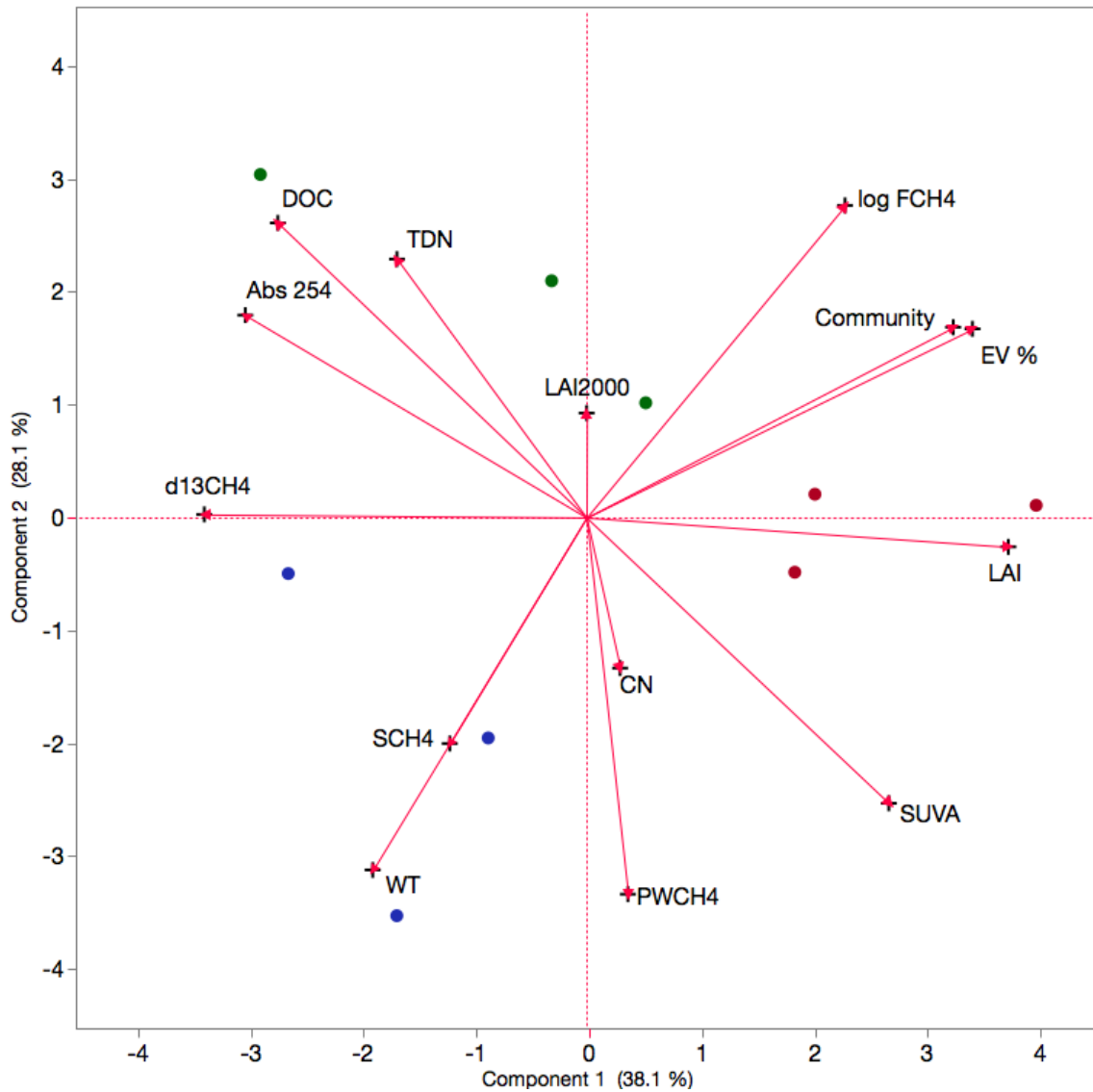


Figure 4-21: Principal component analysis for averaged chamber-specific variables between DOY 212 and 255. N = 9. All pore water and temperature variables are from the 50 cm depth. Variables include: community type, total green hits LAI (LAI), percent LAI attributed to *Eriophorum vaginatum* (EV %), total biomass LAI as measured with the LAI-2000 (LAI2000), CH<sub>4</sub> flux (F<sub>CH4</sub>), storage change of pore water CH<sub>4</sub> (SCH<sub>4</sub>), pore water CH<sub>4</sub> concentration (PW<sub>CH4</sub>),  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signature ( $\delta^{13}\text{C}$ -CH<sub>4</sub>), water table depth (WT), dissolved organic carbon (DOC), total dissolved nitrogen (TDN), carbon to nitrogen ratio (CN), absorbance at 254 nm (Abs<sub>254</sub>), and specific ultraviolet absorbance at 254 nm (SUVA<sub>254</sub>). Chambers are dots and grouped according to community type: *Chamaedaphne* = blue, *Ledum* = green, and *Eriophorum* = red.

Table 4-2: Pearson correlations for averaged chamber-specific variables between DOY 212 and 255. All pore water and temperature variables are from the 50 cm depth. Correlations in bold ( $p < 0.05$ ) and italics ( $p < 0.10$ ) are significant. N = 9. Variables include: community type, total green hits LAI (LAI), percent LAI attributed to *Eriophorum vaginatum* (EV %), total biomass LAI as measured with the LAI-2000 (LAI2000), CH<sub>4</sub> flux (F<sub>CH4</sub>), storage change of pore water CH<sub>4</sub> (S<sub>CH4</sub>), pore water CH<sub>4</sub> concentration (PW<sub>CH4</sub>),  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signature ( $\delta^{13}\text{C}$ -CH<sub>4</sub>), water table depth (WT), dissolved organic carbon (DOC), total dissolved nitrogen (TDN), carbon to nitrogen ratio (CN), absorbance at 254 nm (Abs<sub>254</sub>), and specific ultraviolet absorbance at 254 nm (SUVA<sub>254</sub>).

	Community	LAI- 2000	LAI	EV %	WT	DOC	TDN	CN	Abs <sub>254</sub>	SUVA <sub>2</sub> 54	$\delta^{13}\text{C}$ - CH <sub>4</sub>	PW <sub>CH4</sub>	S <sub>CH4</sub>	log F <sub>CH4</sub>
Community	<b>1.00</b>													
LAI2000	0.00	<b>1.00</b>												
LAI	<b>0.68</b>	0.16	<b>1.00</b>											
EV %	<b>0.87</b>	0.18	<b>0.77</b>	<b>1.00</b>										
WT	<b>-0.70</b>	-0.03	-0.39	<b>-0.81</b>	<b>1.00</b>									
DOC	-0.27	0.18	<i>-0.61</i>	-0.30	-0.23	<b>1.00</b>								
TDN	-0.21	<b>0.68</b>	-0.33	-0.12	-0.16	<i>0.64</i>	<b>1.00</b>							
CN	0.04	<b>-0.74</b>	-0.01	-0.04	0.05	-0.19	<b>-0.86</b>	<b>1.00</b>						
Abs <sub>254</sub>	-0.46	0.02	<b>-0.75</b>	-0.49	0.05	<b>0.79</b>	0.40	-0.02	<b>1.00</b>					
SUVA <sub>254</sub>	0.26	-0.18	<i>0.59</i>	0.28	0.25	<b>-0.99</b>	<i>-0.66</i>	0.21	<b>-0.71</b>	<b>1.00</b>				
$\delta^{13}\text{C}$ -CH <sub>4</sub>	<i>-0.59</i>	-0.45	<b>-0.94</b>	<b>-0.72</b>	0.30	0.55	0.09	0.28	<i>0.63</i>	-0.56	<b>1.00</b>			
PW <sub>CH4</sub>	-0.38	0.12	0.30	-0.28	<i>0.63</i>	-0.52	-0.24	-0.03	-0.56	0.48	-0.28	<b>1.00</b>		
S <sub>CH4</sub>	<i>-0.59</i>	0.42	-0.06	-0.33	0.42	0.01	0.16	-0.15	-0.12	-0.05	-0.01	<b>0.69</b>	<b>1.00</b>	
log F <sub>CH4</sub>	<i>0.65</i>	0.05	0.54	<b>0.72</b>	<b>-0.80</b>	0.06	0.14	-0.17	-0.05	-0.04	-0.51	-0.46	<i>-0.60</i>	<b>1.00</b>



## 5.0 DISCUSSION

Vegetation community characteristics are often used as a predictor for CH<sub>4</sub> emissions in peatlands, with sedge-dominated vegetation expected to have greater emissions than shrub, herb or moss dominated plots (Ward *et al.* 2013). At Mer Bleue, community type or % LAI attributed to *Eriophorum* (*EV%*), more specifically, was an important variable associated with greater CH<sub>4</sub> emissions, more depleted  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signature, and less storage change of CH<sub>4</sub> although perhaps surprisingly, there was no direct relationship between *EV%* and pore water characteristics. WT correlated well with community type and the *EV%* (Figure 4-21). WT at Mer Bleue is a good indicator of the potential magnitude of CH<sub>4</sub> emissions (Moore *et al.* 2011, Brown *et al.* 2014, Lai *et al.* 2014a). WT is also a good indicator of the vegetation community composition, where sedge cover (*Eriophorum vaginatum*) increased as the average WT depth approached the surface, and dwarf shrubs (*Chamaedaphne calyculata*) dominated in drier hummock positions (Bubier *et al.* 2006).

The  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signatures for all nine chambers ranged from -81 to -71 ‰, all well within the -110 to -60‰ range, suggesting the dominance of the CO<sub>2</sub> reduction pathway for CH<sub>4</sub> production at Mer Bleue (Whiticar *et al.* 1986). Although the vegetation community composition did not have a significant effect on the source signature when accounting for the random effects of the repeated measurements on a chamber (Figure 4-18), a simple negative correlation with the *EV%* suggested the influence of transport effects on source signature, as increased sedge cover allows for increased plant-mediated CH<sub>4</sub> transport. Plant-mediated transport processes, as well as diffusion processes,

preferentially remove the lighter  $^{12}\text{C}$ - $\text{CH}_4$  (Chanton 2005). However, the distance  $\text{CH}_4$  diffuses through the oxidation zone will also impact the  $\delta^{13}\text{C}$ - $\text{CH}_4$  value due to the substantial fractionation (5-30‰) of C by methanotrophs (i.e. microbes preferentially consume  $^{12}\text{C}$ - $\text{CH}_4$ ) (Whiticar 1999). In a study investigating C fractionation by methanotrophs, Coleman *et al.* (1981) found evidence for a microbial preference for the consumption of lighter  $^{12}\text{C}$ - $\text{CH}_4$ , leaving residual  $\text{CH}_4$  enriched with heavier  $^{13}\text{C}$  after  $\text{CH}_4$  oxidation. Thus, Figure 4-18 suggests that the presence of *Eriophorum* allows the direct transport of lighter, more depleted  $\text{CH}_4$  to the atmosphere by bypassing the  $\text{CH}_4$  consumption zone and associated fractionation by methanotrophs.

Optimal  $\text{CH}_4$  production conditions are likely related to input and output dynamics of potential substrates (Kalbitz *et al.* 2000, Preston *et al.* 2011), as organic C substrate and a source of N are necessary for methanogenic activity. The instantaneous DOC and TDN concentrations obtained from pore water are a function of production processes involving multiple sources (e.g. litter and humus, root exudates, microbial biomass), consumption through microbial activity, and transport dynamics (e.g. flushing, adsorption) (Kalbitz *et al.* 2000, Preston *et al.* 2011). While TDN comprises ammonium ( $\text{NH}_4^+$ ), nitrate ( $\text{NO}_3^-$ ) and dissolved organic nitrogen (DON), a previous study showed that the majority of TDN in Mer Bleue pore water is in the organic DON form (Rattle 2006). As such, there is an interaction between C and N cycling, and patterns in TDN are related to DOC dynamics (Rattle 2006). The current study demonstrated increasing pore water concentrations of DOC and TDN over the season (Figure 4-13). During the spring, snowmelt at Mer Bleue flushes the peat with small concentrations of DOC and TDN. As peat temperature increases throughout the growing season, decomposition processes

accelerate and both DOC and TDN are released into pore water (Moore & Dalva 1993, Kalbitz *et al.* 2000, Moore & Dalva 2000, Lai 2009). This likely contributes to the increasing seasonal pattern in DOC and TDN observed at all depths over 2014 at Mer Bleue.

The C/N ratio was assumed to be equal to the DOC/TDN ratio since TDN in pore water is dominated by DON at Mer Bleue (Rattle 2006). The pore water C/N ratio demonstrated limited temporal variation, which reflects the seasonal increases in both DOC and TDN (Figure 4-13). The pore water C/N ratio decreased with depth, similar to patterns in peat C/N ratio found in previous studies (Hornibrook *et al.* 2000b, Malmer & Wallén 2004, Rydin & Jeglum 2013, Wang *et al.* 2015). This can be attributed to a loss of C (i.e. consumption of carbohydrates) and fixation of N by microorganisms with depth (Hornibrook *et al.* 2000b, Rydin & Jeglum 2013). The peak TDN, and corresponding trough in C/N ratio, that occurred in June may have been a function of spring thawing and microbial turnover (Moore & Dalva 2000).

The quality of the DOC substrate was assessed using SUVA<sub>254</sub> measurements, which have a positive linear relation with DOC aromaticity (Weishaar *et al.* 2003, Olefeldt & Roulet 2012). Thus, greater SUVA<sub>254</sub> for pore water DOC indicates greater aromaticity of the DOC molecule (Weishaar *et al.* 2003). Measured SUVA<sub>254</sub> values fell within the commonly reported range of 0.5-6.0 L m<sup>-1</sup> mg<sup>-1</sup> DOC, corresponding to a percent aromaticity of 5-45% (Weishaar *et al.* 2003). The current results indicate an inverse relation between DOC and SUVA<sub>254</sub> measurements, with SUVA<sub>254</sub> (and thus DOC aromaticity) decreasing as the season progressed (Figure 4-13). This suggests that at low initial DOC concentrations at the beginning of the season, DOC is more aromatic

in nature. Since methanogens favour simple, relatively labile DOC molecules, this could limit CH<sub>4</sub> production during this early summer period (Blodau 2001). The increasing DOC concentration and lability over the 2014 summer at Mer Bleue supports findings by Kane *et al.* (2014), where increased temperature ( $1.9 \pm 0.4^{\circ}\text{C}$ ) in poor fen plots produced increased concentrations of more labile DOC in pore water.

One reason neither community type nor *EV%* directly related to pore water characteristics may have been due to the significant variability in pore water characteristics for *Eriophorum* community plots (Figure 4-14). This could be the result of the deeper rooting zone of sedges, which add labile C and other organic substrates to the anaerobic zone, potentially contributing to the increased variability in TDN and DOC quantity and quality.

The increasing seasonal DOC and TDN pore water concentrations, in conjunction with the presence of more labile DOC as the summer progressed, likely contributed to the spatial and temporal variability of pore water CH<sub>4</sub> concentrations at depth. The pore water CH<sub>4</sub> concentrations measured in this study were in range with those measured by Blodau *et al.* (2007) during 2003-2004, which yielded CH<sub>4</sub> concentrations in the range of 0-600  $\mu\text{mol L}^{-1}$ . In this study, pore water CH<sub>4</sub> concentrations peaked at 50 cm by late summer (Figure 4-15). Similarly Blodau *et al.* (2007) also found increasing concentrations over the summer months, with a peak CH<sub>4</sub> concentration at 40 - 55 cm by September/October in 2003 but a general trend of increasing concentrations with depth in 2004 (to 60 cm). This could be attributed to differences in moisture conditions where 2003 and 2004 WT reached depths 10-14 cm deeper than in 2014 and may have resulted in a deeper zone of CH<sub>4</sub> production. Also, sampling methods differed, where Blodau *et*

*al.* (2007) used the peeper technique and the current study used suction sampling and closed piezometer techniques. As previously detailed (Section 1.1.1), the peeper technique requires long equilibration times; from June 2003 to December 2004, Blodau *et al.* (2007)'s peepers were only sampled 12 times, for example. The shorter weekly interval between samples in the current study may have captured CH<sub>4</sub> concentration depth patterns that were not apparent with monthly sampling (i.e. spatial variability at a shorter time scale). Redox potential measurements obtained during the late-summer period supported the importance of the 50 cm peak in the CH<sub>4</sub> concentration profile, as they indicated strongly reducing conditions (<-100 mV) (Zhi-Guang 1985) at around this depth for most chambers.

In general, plots with smaller fluxes tended to have greater increases in CH<sub>4</sub> storage over the measurement period. For example, the *Chamaedaphne* community produced some of the largest pore water CH<sub>4</sub> concentrations at depth in late summer (Figure 4-15). Yet, this community consistently had the lowest CH<sub>4</sub> fluxes throughout the summer (Table 4-1, Figure 4-5). Since it is unlikely production rates were greater in *Chamaedaphne* plots, this suggests the influence of restrictive transport of CH<sub>4</sub> to the atmosphere and/or greater CH<sub>4</sub> consumption within the aerated peat at *Chamaedaphne* plots. As discussed earlier, there is evidence of *Eriophorum* providing a direct pathway to the atmosphere through aerenchymatous tissue such that more *Eriophorum* plants may enhance transport and limit the build-up of CH<sub>4</sub> in the pore water, with the opposite occurring in *Chamaedaphne*-dominated plots where diffusion through the peat (or ebullition) are the dominant pathways.

Although there is no direct measure of production rates in this study, production may have been lower in *Chamaedaphne*-dominated plots and greater in plots with more *Eriophorum* despite the greater pore water concentrations of CH<sub>4</sub> in *Chamaedaphne* plots. *Eriophorum* is usually associated with enhanced C and substrate addition leading to increased CH<sub>4</sub> production (Strack *et al.* 2008). The *Eriophorum* community produced a relatively restricted pore water CH<sub>4</sub> profile in early summer, with the late-summer profile demonstrating a more clear peak in CH<sub>4</sub> concentrations at 50 cm despite chamber E-M 2 having very low late-summer CH<sub>4</sub> concentrations at 50 cm and drawing the community average down. This chamber was also associated with very high 10 cm soil temperature (Figure 4-7), moderate fluxes and redox potential values in the -100 to 0 mV range during late summer, indicating a lack of strongly reducing conditions associated with CO<sub>2</sub> reduction to CH<sub>4</sub> (Figure 4-12). This implies that methanogenic activity was not optimized at the 40-50 cm production zone for E-M 2. While this redox status does not indicate the predominance of O<sub>2</sub> (>400 mV), it may provide evidence that CO<sub>2</sub> reduction processes are being suppressed by the consumption of more favourable terminal electron acceptors such as iron or sulfur (Table 2-1) (Deppe *et al.* 2010b, Bridgham *et al.* 2013). In contrast, strongly reducing conditions (~ -200 mV) were found at 40-50 cm depth for E-M 1 (Figure 4-11) where both CH<sub>4</sub> fluxes and 50 cm concentrations were high mid- to late-summer. Below this depth, decomposition is likely limited by substrate supply, possibly leading to a greater proportion of terminal electron acceptors higher up the redox pair cascade (i.e. SO<sub>4</sub><sup>2-</sup>, Fe<sup>3+</sup>) and redox potential measurements that are less negative. Although E-M 3 also showed evidence for strongly reducing conditions (~ -200 mV) and produced large 50 cm CH<sub>4</sub> concentrations late in the season, it consistently had the lowest

CH<sub>4</sub> fluxes of the *Eriophorum* chambers throughout the summer. This *Eriophorum* plant was sitting atop a hummock that was ~9 cm above the top of the collar and surrounding peat surface. This possibly exposed a greater surface area of peat to the atmosphere in the top 9 cm and contributed to increased CH<sub>4</sub> consumption potential as the CH<sub>4</sub> diffused up through the aerobic zone. C-A 1 and 3 both showed relatively large mean seasonal storage change, while C-A 2 had a value similar to other chambers in the *Ledum* and *Eriophorum* communities (Figure 4-19). As this *Chamaedaphne* chamber had the deepest WT of all chambers, it is likely that oxidation processes were again consuming CH<sub>4</sub> in the aerobic peat layer, yielding small CH<sub>4</sub> fluxes and a smaller than expected increase in CH<sub>4</sub> storage over the season.

The seasonal increase in 50 cm CH<sub>4</sub> concentrations is likely tied to increasing trends in DOC and TDN, which are required for methanogen metabolism. As 50 cm temperatures increased as the summer progressed, decomposition rates would have been expected to increase. Previous studies have addressed the importance of peat temperature for CH<sub>4</sub> dynamics, where CH<sub>4</sub> production rates are positively related to peat temperature (Dunfield *et al.* 1993, Valentine *et al.* 1994). CH<sub>4</sub> oxidation processes, however, are not as sensitive to temperature (Dunfield *et al.* 1993). In an examination of northern fens, a peatland hydrologic model by Roulet *et al.* (1992) predicted that an increase in air temperature of 3°C and associated increase in 10 cm peat temperature of 0.8°C could produce a CH<sub>4</sub> flux increase of 5-40%. In the current study, peat temperature at 10 cm was not a significant control on CH<sub>4</sub> fluxes over the 2014 sampling period (Figure 4-7). The pattern in 10 cm peat temperature was influenced by day-to-day variation, with patterns very similar to daily air temperature (Figure 4-3). Moore *et al.* (2011) found 40

cm peat temperature at Mer Bleue was significantly correlated with CH<sub>4</sub> flux, with WT representing a control on seasonal differences in CH<sub>4</sub> flux. The present study found 50 cm peat temperature to relate best with CH<sub>4</sub> flux (Figure 4-8), presumably because this depth is where CH<sub>4</sub> production is greatest (Blodau *et al.* 2007) and it is this mechanism that is likely most important in determining CH<sub>4</sub> emission rates. Although, warmer temperatures may also enhance CH<sub>4</sub> transport processes and reduce storage in pore water.

The statistically significant interaction effect of vegetation community and 50 cm peat temperature on CH<sub>4</sub> flux suggested that vegetation communities' CH<sub>4</sub> dynamics differed in their response to increased peat temperature. Previous studies have suggested that increased peat temperature could enhance ebullition CH<sub>4</sub> flux by increasing bubble volume (Fechner-Levy & Hemond 1996). It is possible that this increased ebullition flux response disproportionately impacts low-lying plots with WT close to the surface, as bubbles would have a thinner aerobic layer to traverse and would thereby have reduced potential for consumption through CH<sub>4</sub> oxidation. Increased peat temperature has also been suggested to enhance plant CH<sub>4</sub> transport through increased aerenchyma ventilation (Große 1996). This further underlines the influence of *Eriophorum* on CH<sub>4</sub> transport and emission rates.



## 6.0 CONCLUSION

Upon examination of a variety of potential mechanisms associated with CH<sub>4</sub> dynamics and inferred differences in CH<sub>4</sub> production, storage and transport processes for dominant vegetative communities during a wet summer at Mer Bleue, the following conclusions can be drawn in reference to thesis hypotheses:

1. During the growing season, both spatial and temporal variability in CH<sub>4</sub> emissions were high but temporal variations were greatest in the plots dominated by *Eriophorum* plants likely due to the combined effects of temporal variability in substrate production for methanogens and the influence of aerenchymatous tissues on CH<sub>4</sub> transport to the atmosphere.
2. Pore water biogeochemistry suggested increasing substrate availability and CH<sub>4</sub> production rates as peat temperature and plant growth increased throughout the growing season. Conditions for CH<sub>4</sub> production via CO<sub>2</sub> reduction were optimized near the 50 cm depth, as this depth is characterized by strongly reducing conditions and is close enough to the long-term average WT at Mer Bleue for sufficient organic C substrate supply.
3. The nonlinear relationship between mean 50 cm CH<sub>4</sub> storage change and mean CH<sub>4</sub> flux provided support for the hypothesis that plots with smaller storage increase are associated with larger CH<sub>4</sub> emissions. However, the lack of a significant linear relationship highlights the complexity of CH<sub>4</sub> dynamics in peatlands, and suggests a varying influence of CH<sub>4</sub> transport and consumption processes among chambers and communities at Mer Bleue.

Considering the uncertainty surrounding the global CH<sub>4</sub> budget, and the significance of CH<sub>4</sub> emissions from northern peatland ecosystems in particular, this study contributes a site-specific examination of CH<sub>4</sub> dynamics at the plant community scale for a temperate ombrotrophic bog that confirms the importance of plant community characteristics on CH<sub>4</sub> emissions through factors influencing production, storage and transport. Future studies could employ isotopic analyses of pore water samples from key depths to track the movement and transformations of CH<sub>4</sub> through the peat profile over the growing season, and enhance understanding of the relative importance of CH<sub>4</sub> production, consumption, and transport processes at Mer Bleue. Depth profiles could also be sampled at a higher temporal and spatial resolution in conjunction with full-season redox potential measurements to confirm the presence/importance, and spatial and seasonal variability of a critical zone below the water table for CH<sub>4</sub> production through methanogenic activity. If CH<sub>4</sub> production processes can be primarily related to a particular depth/zone in the peat profile regardless of the peatland, this would simplify modeling efforts to improve our understanding of the role peatlands play in the global CH<sub>4</sub> budget. Similarly, there is a need to focus not only on the influence of warmer peat on CH<sub>4</sub> production potential, but also on the response of CH<sub>4</sub> transport pathways to warmer peat in the context of global climate change.

## 7.0 APPENDICES

### Appendix A Autochamber photographs



*Chamaedaphne*-dominated autochamber (C-A 1) on June 18, 2014.



*Chamaedaphne*-dominated autochamber (C-A 2) on June 18, 2014.





*Chamaedaphne*-dominated autochamber (C-A 3) on June 18, 2014.



*Ledum*-dominated autochamber (L-A 1) on June 18, 2014.





*Ledum*-dominated autochamber (L-A 2) on June 18, 2014.



*Ledum*-dominated autochamber (L-A 3) on June 18, 2014.

## Appendix B Calculating concentrations of pore water constituents

### B.1 Units, conversions and constants

Relevant base and supplementary SI units (Brock & Richardson 2001).

<i>Physical Quantity</i>	<i>Unit Name</i>	<i>Symbol</i>
Length	meter	m
Mass	kilogram	kg
Time	second	s
Thermodynamic temperature	kelvin	K
Amount of substance	mole	mol
Plane angle	radian	rad

Relevant derived units (Brock & Richardson 2001).

<i>Derived Quantity</i>	<i>Unit Name</i>	<i>Symbol</i>	<i>SI Base Unit</i>	<i>Derived Unit</i>
Pressure, stress	pascal	Pa	$\text{kg m}^{-1} \text{s}^{-2}$	$\text{N m}^{-2}$
Energy, work, heat	joule	J	$\text{kg m}^2 \text{s}^{-2}$	N m
Force	newton	N	$\text{kg m s}^{-2}$	
Power	watt	W	$\text{kg m}^2 \text{s}^{-3}$	$\text{J s}^{-1}$
Area	square meter	A	$\text{m}^2$	
Volume	cubic meter	V	$\text{m}^3$	

Relevant units in continuous use with SI system of units (Brock & Richardson 2001).

<i>Unit Name</i>	<i>Symbol</i>	=
Minute	min	60 s
Hour	h	3600 s
Day	d	86 400 s
Degree	°	$(\pi/180) \text{ rad}$
Minute	'	$(1/60)^\circ$
Second	"	$(1/60)'$
Liter	L	$10^{-3} \text{ m}^3$

Relevant universal constants.

<i>Universal Constant</i>	<i>Symbol</i>	<i>Value</i>	<i>Reference</i>
Air density (0°C, 100 kPa)	$\rho_a$	$1.275 \text{ kg m}^{-3}$	Brock & Richardson 2001
Water density (0°C, liquid)	$\rho_w$	$1000 \text{ kg m}^{-3}$	Brock & Richardson 2001
Ideal gas constant	$R^*$	$8.3144261 \text{ J mol}^{-1} \text{ K}^{-1}$	Brock & Richardson 2001
Standard atmospheric pressure		101.33 kPa	Oke 1987

Pressure unit conversion table.

	<i>mbar</i>	<i>bar</i>	<i>Torr (mm Hg)</i>	<i>Pa</i>	<i>kPa</i>	<i>atm</i>
<i>1 mbar =</i>	1					
<i>1 bar =</i>	10 <sup>3</sup>	1				
<i>1 Torr =</i>	1.333	1.333 x 10 <sup>-3</sup>	1			
<i>1 Pa =</i>	10 <sup>-2</sup>	10 <sup>-5</sup>	7.5006 x 10 <sup>-3</sup>	1		
<i>1 kPa =</i>	10 <sup>1</sup>	10 <sup>-2</sup>	7.5006	10 <sup>3</sup>	1	
<i>1 atm =</i>	1.01325 x 10 <sup>3</sup>	1.01325	7.60 x 10 <sup>2</sup>	1.01325 x 10 <sup>5</sup>	1.01325 x 10 <sup>2</sup>	1

Other common conversions

<i>Unit</i>	<i>Gas</i>	<i>Solution</i>
Part per million (ppm)	μmol mol <sup>-1</sup>	mg/L
Part per billion (ppb)	nmol mol <sup>-1</sup>	μg/L

Constants for the calculation of CO<sub>2</sub> solubility (mol L<sup>-1</sup> atm<sup>-1</sup>) from moist air at a total pressure of 1 atm (Weiss & Price 1980).

<i>Constants</i>	<i>Value (mol L<sup>-1</sup> atm<sup>-1</sup>)</i>
A <sub>1</sub>	-160.7333
A <sub>2</sub>	215.4152
A <sub>3</sub>	89.8920
A <sub>4</sub>	-1.47759
B <sub>1</sub>	0.029941
B <sub>2</sub>	-0.027455
B <sub>3</sub>	0.0053407

Constants for the calculation of temperature (°K) and salinity (‰) dependent CH<sub>4</sub> Bunsen solubility coefficient (L L<sup>-1</sup> atm<sup>-1</sup>) (Wanninkhof 1992).

<i>Constants</i>	<i>Value (L L<sup>-1</sup> atm<sup>-1</sup>)</i>
A <sub>1</sub>	-68.8862
A <sub>2</sub>	101.4956
A <sub>3</sub>	28.7314
B <sub>1</sub>	-0.076146
B <sub>2</sub>	0.043970
B <sub>3</sub>	-0.0068672

## B.2 Calculating CO<sub>2</sub> concentration in pore water

### B.2.1 CO<sub>2</sub> solubility

As defined by Weiss & Price (1980), the solubility of CO<sub>2</sub> in water ( $F$ , mol L<sup>-1</sup> atm<sup>-1</sup>) as a function of temperature and salinity, including all nonideality effects, is calculated as:

$$(1) \quad \ln F = A_1 + A_2(100/T) + A_3 \ln(T/100) + A_4(T/100)^2 \\ + S[B_1 + B_2(T/100) + B_3(T/100)^2]$$

$$(2) \quad F = e^{\ln F}$$

where  $A$  and  $B$  values are constants (mol L<sup>-1</sup> atm<sup>-1</sup>) defined in Section A.1,  $T$  is absolute temperature (K) and  $S$  is salinity in parts per thousand (‰).

### B.2.2 CO<sub>2</sub> in headspace

The amount of CO<sub>2</sub> in the headspace ( $CO_{2h}$ , μmol) after equilibration is calculated as:

$$(3) \quad [CO_2]_{RES} = [CO_2]_{GC} - [CO_2]_{AMB}$$

$$(4) \quad CO_{2h} = \left( [CO_2]_{RES} \frac{PV_h}{R * T} \right) \frac{1m^3}{1000L}$$

where  $[CO_2]_{GC}$  is the headspace concentration of CO<sub>2</sub> (μmol CO<sub>2</sub> mol<sup>-1</sup>), as measured on the gas chromatograph,  $[CO_2]_{AMB}$  is the ambient concentration of CO<sub>2</sub> (μmol CO<sub>2</sub> mol<sup>-1</sup>) at the time of sampling,  $[CO_2]_{RES}$  is the residual concentration of CO<sub>2</sub> in the headspace (μmol CO<sub>2</sub> mol<sup>-1</sup>) after subtracting the ambient influence,  $P$  is the pressure at



equilibration (Pa),  $V_h$  is the headspace volume after equilibration (L),  $R^*$  is the ideal gas constant 8.314472 J mol<sup>-1</sup> K<sup>-1</sup> as given in Section A.1, and  $T$  is the absolute air temperature at equilibration (K).

### ***B.2.3 CO<sub>2</sub> in water***

The amount of CO<sub>2</sub> in the water ( $CO_{2_w}$ , μmol) after equilibration is calculated as:

$$(5) \quad CO_{2_w} = F \times [CO_2]_{RES} \times P \times V_w$$

where  $F$  is the CO<sub>2</sub> solubility (mol L<sup>-1</sup> atm<sup>-1</sup>, Equation 1 and 2),  $[CO_2]_{RES}$  is the residual concentration of CO<sub>2</sub> in the headspace (μmol CO<sub>2</sub> mol<sup>-1</sup>) as per Equation 3,  $P$  is the pressure at equilibration (atm), and  $V_w$  is the water volume after equilibration (L).

### ***B.2.4 Total CO<sub>2</sub> concentration in pore water***

The total concentration of CO<sub>2</sub> in the pore water ( $[CO_2]_{PW}$ , μmol CO<sub>2</sub> L<sup>-1</sup>) can thus be calculated as:

$$(6) \quad [CO_2]_{PW} = \frac{CO_{2_h} + CO_{2_w}}{V_w}$$

where  $CO_{2_h}$  and  $CO_{2_w}$  are the amount of CO<sub>2</sub> in the headspace and water respectively after equilibration (μmol CO<sub>2</sub>), and  $V_w$  is the water volume after equilibration (L).

### ***B.2.5 Sample CO<sub>2</sub> pore water concentration calculations***

The above procedure was applied for sample input variables, and the calculated concentration values of CO<sub>2</sub> in pore water are provided in the table below.

Sample calculations for determining the concentration of CO<sub>2</sub> in pore water according to the procedure detailed in Sections A.2.1 through A.2.4.

<i>Variable</i>	<i>Units</i>	<i>Sample #1</i>	<i>Sample #2</i>	<i>Sample #3</i>
<i>T</i>	K	283.15	293.15	303.15
<i>S</i>	‰	0	0	0
<i>F</i>	mol L <sup>-1</sup> atm <sup>-1</sup>	5.28 × 10 <sup>-2</sup>	3.81 × 10 <sup>-2</sup>	2.85 × 10 <sup>-2</sup>
<i>[CO<sub>2</sub>]<sub>GC</sub></i>	μmol CO <sub>2</sub> mol <sup>-1</sup>	8.15 × 10 <sup>2</sup>	1.01 × 10 <sup>4</sup>	3.13 × 10 <sup>2</sup>
<i>[CO<sub>2</sub>]<sub>AMB</sub></i>	μmol CO <sub>2</sub> mol <sup>-1</sup>	4.02 × 10 <sup>2</sup>	4.02 × 10 <sup>2</sup>	4.02 × 10 <sup>2</sup>
<i>[CO<sub>2</sub>]<sub>RES</sub></i>	μmol CO <sub>2</sub> mol <sup>-1</sup>	4.13 × 10 <sup>2</sup>	9.65 × 10 <sup>3</sup>	3.09 × 10 <sup>4</sup>
<i>P</i>	Pa	1.0076 × 10 <sup>5</sup>	1.0076 × 10 <sup>5</sup>	1.0076 × 10 <sup>5</sup>
<i>P</i>	atm	9.94423 × 10 <sup>-1</sup>	9.94423 × 10 <sup>-1</sup>	9.94423 × 10 <sup>-1</sup>
<i>V<sub>h</sub></i>	L	3.0 × 10 <sup>-2</sup>	3.0 × 10 <sup>-2</sup>	3.0 × 10 <sup>-2</sup>
<i>V<sub>w</sub></i>	L	3.0 × 10 <sup>-2</sup>	3.0 × 10 <sup>-2</sup>	3.0 × 10 <sup>-2</sup>
<i>CO<sub>2h</sub></i>	μmol CO <sub>2</sub>	5.31 × 10 <sup>-1</sup>	1.20 × 10 <sup>1</sup>	3.70 × 10 <sup>1</sup>
<i>CO<sub>2w</sub></i>	μmol CO <sub>2</sub>	6.51 × 10 <sup>-1</sup>	1.10 × 10 <sup>1</sup>	2.62 × 10 <sup>1</sup>
<i>[CO<sub>2</sub>]<sub>PW</sub></i>	μmol CO <sub>2</sub> L <sup>-1</sup>	3.94 × 10 <sup>1</sup>	7.66 × 10 <sup>2</sup>	2.11 × 10 <sup>3</sup>

### B.3 Calculating CH<sub>4</sub> concentration in pore water

#### B.3.1 CH<sub>4</sub> solubility

The Bunsen solubility coefficient of CH<sub>4</sub> ( $\beta$ , mol L<sup>-1</sup> atm<sup>-1</sup>) can be calculated as a function of temperature and salinity (Wanninkhof 1992), where:

$$(7) \quad \ln \beta = A_1 + A_2(100/T) + A_3 \ln(T/100) + S[B_1 + B_2(T/100) + B_3(T/100)^2]$$

$$(8) \quad \beta = e^{\ln \beta} \frac{101325}{R^*(298.15)} \square 0.001$$

where  $A$  and  $B$  values are constants (L L<sup>-1</sup> atm<sup>-1</sup>) defined in Section A.1,  $T$  is absolute temperature (K) and  $S$  is salinity in parts per thousand (‰). Equation 8 converts this solubility value from L CH<sub>4</sub> L<sup>-1</sup> water atm<sup>-1</sup> to units of mol L<sup>-1</sup> atm<sup>-1</sup>.

### B.3.2 CH<sub>4</sub> in headspace

The amount of CH<sub>4</sub> in the headspace ( $CH_{4_h}$ ,  $\mu\text{mol}$ ) after equilibration is calculated as:

$$(9) \quad [CH_4]_{RES} = [CH_4]_{GC} - [CH_4]_{AMB}$$

$$(10) \quad CH_{4_h} = \left( [CH_4]_{RES} \frac{PV_h}{R^*T} \right) \frac{1m^3}{1000L}$$

where  $[CH_4]_{GC}$  is the headspace concentration of CH<sub>4</sub> ( $\mu\text{mol CH}_4 \text{ mol}^{-1}$ ), as measured on the gas chromatograph,  $[CH_4]_A$  is the ambient concentration of CH<sub>4</sub> ( $\mu\text{mol CH}_4 \text{ mol}^{-1}$ ) at the time of sampling,  $[CH_4]_{RES}$  is the residual concentration of CH<sub>4</sub> in the headspace ( $\mu\text{mol CH}_4 \text{ mol}^{-1}$ ) after subtracting the ambient influence,  $P$  is the pressure at equilibration (Pa),  $V_h$  is the headspace volume after equilibration (L),  $R^*$  is the ideal gas constant 8.314 4261 (J K<sup>-1</sup> mol<sup>-1</sup>) as given in Section A.1, and  $T$  is the absolute air temperature at equilibration (K).

### B.3.3 CH<sub>4</sub> in water

The amount of CH<sub>4</sub> in the water ( $CH_{4_w}$ ,  $\mu\text{mol}$ ) after equilibration is calculated as:

$$(11) \quad CH_{4_w} = \beta [CH_4]_{RES} P V_w$$

where  $\beta$  is the Bunsen solubility coefficient of CH<sub>4</sub> ( $\text{mol L}^{-1} \text{ atm}^{-1}$ , Equation 6 and 7),  $[CH_4]_{RES}$  is the residual concentration of CH<sub>4</sub> in the headspace ( $\mu\text{mol CH}_4 \text{ mol}^{-1}$ ),  $P$  is the pressure at equilibration (atm), and  $V_w$  is the water volume after equilibration (L).

### B.3.4 Total CH<sub>4</sub> concentration in pore water

The total concentration of CH<sub>4</sub> in the pore water ( $[CH_4]_{PW}$ ,  $\mu\text{mol CH}_4 \text{ L}^{-1}$ ) can thus be calculated as:

$$(12) \quad [CH_4]_{PW} = \frac{CH_{4h} + CH_{4w}}{V_w}$$

where  $CH_{4h}$  and  $CH_{4w}$  are the amount of CH<sub>4</sub> in the headspace and water respectively after equilibration ( $\mu\text{mol CH}_4$ ), and  $V_w$  is the water volume after equilibration (L).

### B.3.5 Sample CH<sub>4</sub> pore water concentration calculations

The above procedure was applied for sample input variables, and the calculated concentration values of CH<sub>4</sub> in pore water are provided in the table below.

Sample calculations for determining the concentration of CH<sub>4</sub> in pore water according to the procedure detailed in sections A.3.1 through A.3.4.

Variable	Units	Sample #1	Sample #2	Sample #3
$T$	K	283.15	293.15	303.15
$S$	‰	0	0	0
$\beta$	$\text{mol L}^{-1} \text{ atm}^{-1}$	$1.80 \times 10^{-3}$	$1.40 \times 10^{-3}$	$1.20 \times 10^{-3}$
$[CH_4]_{GC}$	$\mu\text{mol CH}_4 \text{ mol}^{-1}$	$1.15 \times 10^2$	$5.43 \times 10^3$	$1.99 \times 10^4$
$[CH_4]_{AMB}$	$\mu\text{mol CH}_4 \text{ mol}^{-1}$	$1.90 \times 10^0$	$1.90 \times 10^0$	$1.90 \times 10^0$
$[CH_4]_{RES}$	$\mu\text{mol CH}_4 \text{ mol}^{-1}$	$1.13 \times 10^2$	$5.42 \times 10^3$	$1.99 \times 10^4$
$P$	Pa	$1.0076 \times 10^5$	$1.0076 \times 10^5$	$1.0076 \times 10^5$
$P$	atm	$9.94423 \times 10^{-1}$	$9.94423 \times 10^{-1}$	$9.94423 \times 10^{-1}$
$V_h$	L	$3.0 \times 10^{-2}$	$3.0 \times 10^{-2}$	$3.0 \times 10^{-2}$
$V_w$	L	$3.0 \times 10^{-2}$	$3.0 \times 10^{-2}$	$3.0 \times 10^{-2}$
$CH_{4h}$	$\mu\text{mol CH}_4$	$1.45 \times 10^{-1}$	$6.73 \times 10^0$	$2.38 \times 10^1$
$CH_{4w}$	$\mu\text{mol CH}_4$	$6.07 \times 10^{-3}$	$2.26 \times 10^{-1}$	$7.12 \times 10^{-1}$
$[CH_4]_{PW}$	$\mu\text{mol CH}_4 \text{ L}^{-1}$	<b><math>5.04 \times 10^0</math></b>	<b><math>2.32 \times 10^2</math></b>	<b><math>8.17 \times 10^2</math></b>

## Appendix C Seasonal CH<sub>4</sub> Fluxes

### C.1 Comparison of CH<sub>4</sub> fluxes among seasons

Comparison of mean daily CH<sub>4</sub> flux (mg CH<sub>4</sub> m<sup>-2</sup> h<sup>-1</sup>), WT (cm) and 50 cm peat temperature (T<sub>50cm</sub>, °C) from *Chamaedaphne* (C-A 1,2,3), *Ledum* (L-A 1,2,3) and *Eriophorum* (E-M 1,2,3)-dominated communities at Mer Bleue, 2014. Standard deviation is given in parentheses. Note: WP, whole period (DOY 120-155); LSp, late spring (DOY 120-154); ES, early summer (DOY 155-189); MS, mid-summer (DOY 190-221); LS, late summer (DOY 222-255) for the 2014 measurement period.

Whole Period (DOY 120-255)			
	CH <sub>4</sub> Flux (mg m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	T <sub>50cm</sub> (°C)
<i>Chamaedaphne</i>	1.03 (0.65)	36.4 (3.2)	11.1 (3.0)
C-A 1	1.68 (1.14)	29.1 (2.4)	
C-A 2	0.75 (0.55)	43.8 (3.1)	
C-A 3	0.65 (0.54)	34.6 (3.2)	
<i>Ledum</i>	3.81 (1.98)	28.5 (3.2)	
L-A 1	2.55 (0.97)	28.8 (2.9)	
L-A 2	4.56 (2.78)	31.1 (2.2)	
L-A 3	4.25 (2.27)	24.6 (2.8)	
<i>Eriophorum</i>	3.27 (2.42)	28.4 (3.1)	
E-M 1	6.16 (4.88)	25.8 (2.8)	
E-M 2	2.13 (1.56)	30.1 (2.7)	
E-M 3	1.27(1.22)	28.2 (2.6)	
Late Spring (DOY 120-154)			
<i>Chamaedaphne</i>	0.31 (0.10)	34.9 (3.2)	6.6 (1.8)
C-A 1	0.68 (0.16)	27.9 (2.5)	
C-A 2	0.22 (0.14)	43.4 (3.3)	
C-A 3	0.01 (0.03)	32.9 (3.4)	
<i>Ledum</i>	0.99 (0.33)	26.9 (2.9)	
L-A 1	1.12 (0.38)	27.3 (3.3)	
L-A 2	0.59 (0.27)	29.8 (2.4)	
L-A 3	1.20 (0.30)	23.1 (2.9)	
<i>Eriophorum</i>	0.42 (0.34)	N/A	
E-M 1	0.33 (0.32)	30.2 (1.6)	
E-M 2	0.68 (0.46)	34.0 (1.5)	
E-M 3	0.40 (0.44)	31.9 (1.4)	

Early Summer (DOY 155-189)			
	CH <sub>4</sub> Flux (mg m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	T <sub>50cm</sub> (°C)
<i>Chamaedaphne</i>	0.88 (0.38)	37.8 (3.4)	11.3 (1.2)
C-A 1	1.60 (0.80)	29.1 (1.7)	
C-A 2	0.62 (0.21)	43.7 (3.5)	
C-A 3	0.38 (0.25)	34.3 (2.9)	
<i>Ledum</i>	3.41 (0.90)	27.9 (2.5)	
L-A 1	2.60 (0.42)	28.6 (2.6)	
L-A 2	4.04 (1.39)	31.2 (2.3)	
L-A 3	3.36 (1.02)	24.4 (2.5)	
<i>Eriophorum</i>	1.58 (0.72)	27.8 (3.6)	
E-M 1	2.57 (1.71)	25.0 (2.9)	
E-M 2	1.22 (0.91)	29.4 (2.8)	
E-M 3	0.49 (0.57)	27.5 (2.7)	
Mid Summer (DOY 190-221)			
<i>Chamaedaphne</i>	1.44(0.50)	37.4 (2.6)	13.2 (0.2)
C-A 1	2.29 (1.29)	30.4 (2.3)	
C-A 2	0.99 (0.53)	45.1 (2.6)	
C-A 3	1.03 (0.20)	36.7 (2.8)	
<i>Ledum</i>	5.33 (0.23)	30.5 (3.2)	
L-A 1	3.22 (0.19)	30.6 (2.2)	
L-A 2	6.56 (0.62)	32.5 (1.6)	
L-A 3	6.11 (0.37)	26.6 (2.6)	
<i>Eriophorum</i>	5.27 (1.83)	28.8 (3.4)	
E-M 1	9.57 (3.77)	27.0 (2.6)	
E-M 2	2.80 (1.66)	31.4 (2.6)	
E-M 3	1.88 (1.52)	29.3 (2.5)	
Late Summer (DOY 222-255)			
<i>Chamaedaphne</i>	1.58 (0.57)	35.6 (2.5)	13.6 (0.4)
C-A 1	2.23 (1.16)	28.9 (2.3)	
C-A 2	1.22 (0.61)	43.0 (2.7)	
C-A 3	1.27 (0.32)	34.7 (2.5)	
<i>Ledum</i>	5.83(0.35)	28.7 (3.1)	
L-A 1	3.35 (0.55)	28.6 (2.4)	
L-A 2	7.41 (0.65)	31.1 (1.8)	
L-A 3	6.64 (0.22)	24.3 (2.0)	
<i>Eriophorum</i>	5.20(1.41)	28.9 (2.4)	
E-M 1	10.04 (3.32)	25.1 (2.4)	
E-M 2	3.43 (1.15)	29.2 (2.2)	
E-M 3	2.01 (0.89)	27.5 (2.2)	

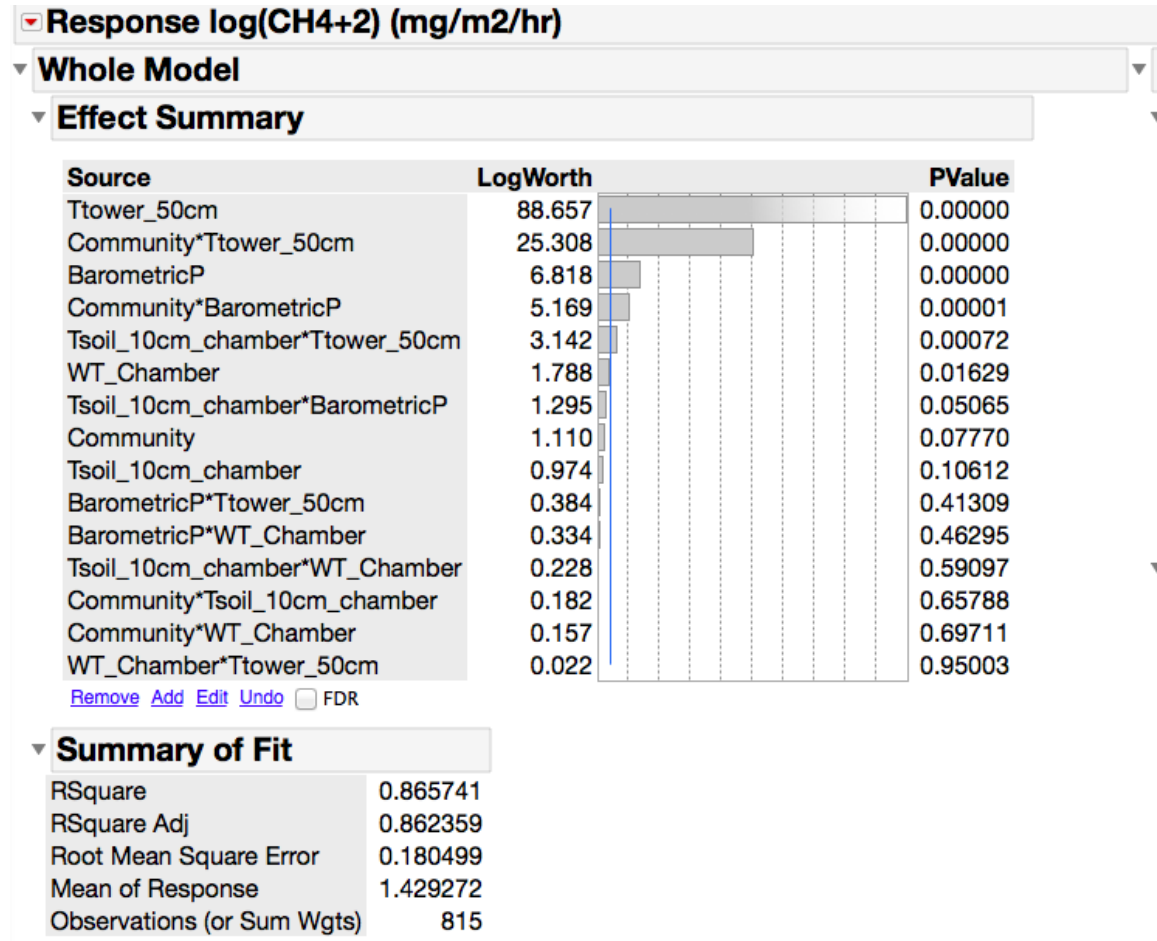
Comparison of mean daily CH<sub>4</sub> flux (mg CH<sub>4</sub> m<sup>-2</sup> h<sup>-1</sup>), WT (cm) and 50 cm peat temperature (T<sub>50cm</sub>, °C) from *Chamaedaphne* (C-A 1,2,3), *Ledum* (L-A 1,2,3) and *Eriophorum* (E-M 1,2,3)-dominated communities at Mer Bleue, 2014. Standard deviation is given in parentheses. Note: WP, whole period (DOY 120-155); LSp, late spring (DOY 120-154); ES, early summer (DOY 155-189); MS, mid-summer (DOY 190-221); LS, late summer (DOY 222-255) for the 2014 measurement period.

<i>Chamaedaphne</i>		<i>Ledum</i>		<i>Eriophorum</i>		T <sub>50cm</sub> (°C)
CH <sub>4</sub> Flux (mg CH <sub>4</sub> m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	CH <sub>4</sub> Flux (mg CH <sub>4</sub> m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	CH <sub>4</sub> Flux (mg CH <sub>4</sub> m <sup>-2</sup> h <sup>-1</sup> )	WT (cm)	
Whole Period (DOY 120-255, n=1116)						
1.03 <sup>b</sup> (0.65)	36.4 <sup>a</sup> (3.2)	3.81 <sup>a</sup> (1.98)	28.5 <sup>ab</sup> (3.2)	3.26 <sup>ab</sup> (2.42)	28.0 <sup>b</sup> (2.7)	11.1 (3.0)
Late Spring (DOY 120-154, n=207)						
0.31 <sup>b</sup> (0.09)	34.9 <sup>a</sup> (3.2)	0.99 <sup>a</sup> (0.33)	26.9 <sup>a</sup> (2.9)	0.42 <sup>NA</sup> (0.34)	N/A	6.6 (1.8)
Early Summer (DOY 155-189, n=207)						
0.88 <sup>b</sup> (0.38)	37.8 <sup>a</sup> (3.4)	3.41 <sup>a</sup> (0.90)	27.9 <sup>a</sup> (2.5)	1.58 <sup>b</sup> (0.72)	27.3 <sup>a</sup> (2.8)	11.3 (1.2)
Mid Summer (DOY 190-221, n=207)						
1.44 <sup>b</sup> (0.50)	37.4 <sup>a</sup> (2.6)	5.33 <sup>a</sup> (0.23)	30.5 <sup>a</sup> (3.2)	5.27 <sup>ab</sup> (1.82)	29.2 <sup>a</sup> (2.6)	13.2 (0.2)
Late Summer (DOY 222-255, n=207)						
1.58 <sup>b</sup> (0.57)	35.6 <sup>a</sup> (2.5)	5.83 <sup>a</sup> (0.35)	28.7 <sup>a</sup> (3.1)	5.20 <sup>a</sup> (1.40)	27.3 <sup>a</sup> (2.3)	13.6 (0.4)

## Appendix D Repeated measures analysis using mixed models in JMP

### D.1 Mixed model analysis of factors affecting CH<sub>4</sub> fluxes

Output of JMP mixed model analysis where the log transform of CH<sub>4</sub> fluxes for manually sampled days is treated as the response with community as a fixed effect, chamber as a random effect and covariates including: barometric pressure, 10 cm peat temperature and WT from each chamber, and 50 cm peat temperature from the tower.





## REML Variance Component Estimates

Random Effect	Var Ratio	Var Component	Std Error	95% Lower	95% Upper	Pct of Total
Chamber[Community]	2.9872185	0.0973232	0.0566209	-0.013652	0.2082981	74.920
Residual		0.0325799	0.0016413	0.0295879	0.0360522	25.080
Total		0.1299031	0.0566438	0.0643588	0.3860533	100.000

-2 LogLikelihood = -274.8156033

Note: Total is the sum of the positive variance components.

Total including negative estimates = 0.1299031

## Covariance Matrix of Variance Component Estimates

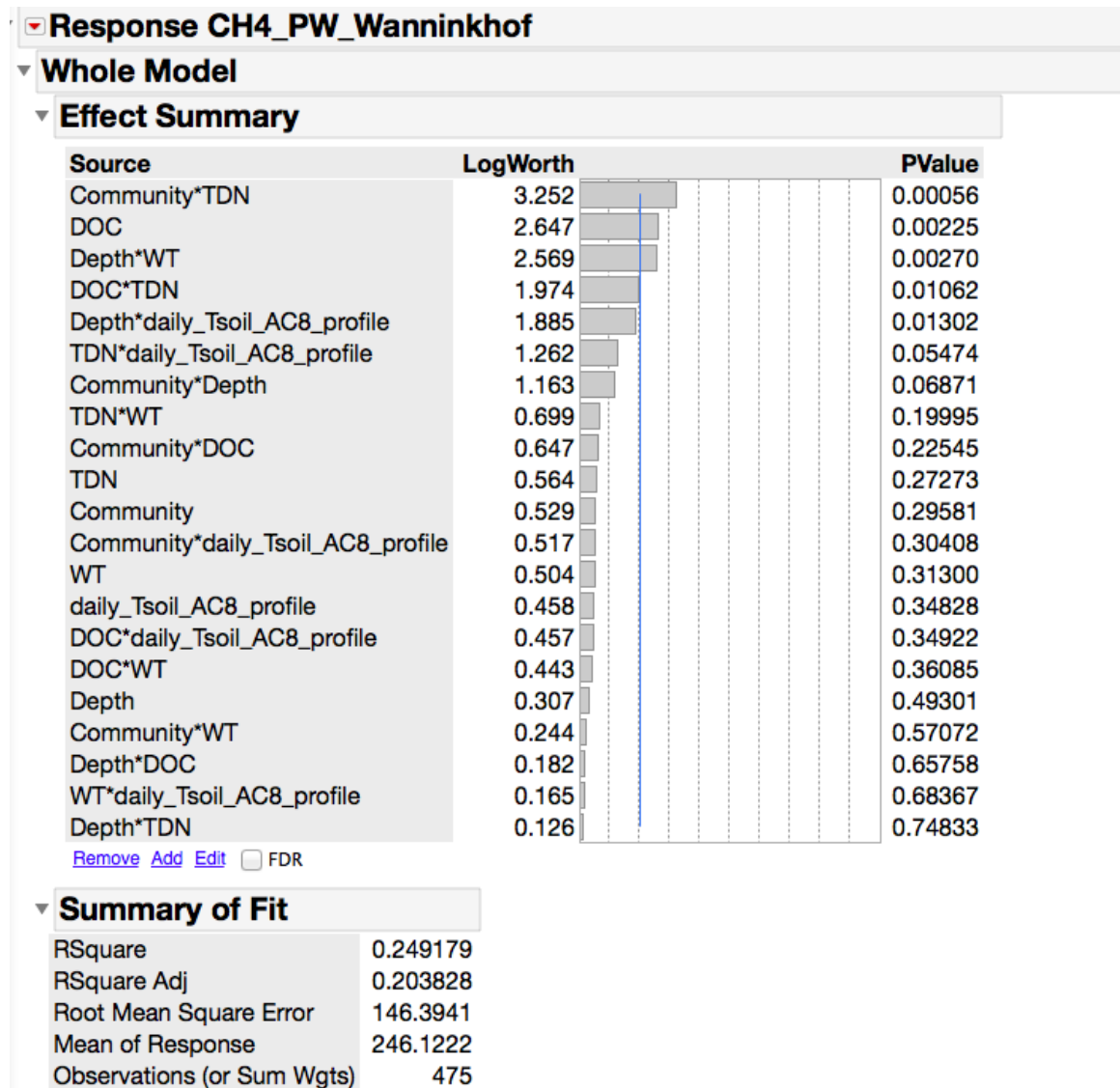
## Iterations

## Fixed Effect Tests

Source	Nparm	DF	DFDen	F Ratio	Prob > F
Community	2	2	6.052	4.0135	0.0777
Tsoil_10cm_chamber	1	1	794	2.6170	0.1061
BarometricP	1	1	788.1	28.0691	<.0001*
WT_Chamber	1	1	793.8	5.7958	0.0163*
Ttower_50cm	1	1	791.8	523.8658	<.0001*
Community*Tsoil_10cm_chamber	2	2	792.2	0.4189	0.6579
Community*BarometricP	2	2	788.1	12.0836	<.0001*
Community*WT_Chamber	2	2	793.1	0.3610	0.6971
Community*Ttower_50cm	2	2	791.1	62.7842	<.0001*
Tsoil_10cm_chamber*BarometricP	1	1	789.2	3.8315	0.0506
Tsoil_10cm_chamber*WT_Chamber	1	1	788.7	0.2891	0.5910
Tsoil_10cm_chamber*Ttower_50cm	1	1	788.3	11.5246	0.0007*
BarometricP*WT_Chamber	1	1	788.2	0.5393	0.4630
BarometricP*Ttower_50cm	1	1	788.9	0.6706	0.4131
WT_Chamber*Ttower_50cm	1	1	788.8	0.0039	0.9500

## D.2 Mixed model analysis of factors affecting pore water CH<sub>4</sub> concentrations

Output of JMP mixed model analysis where the pore water CH<sub>4</sub> concentration (DOY 155-255) is treated as the response with chamber as a random effect and DOC, TDN, depth-specific peat temperature and WT as covariates, and depth and community as fixed effects.



## ▼ REML Variance Component Estimates

Random Effect	Var Ratio	Var Component	Std Error	95% Lower	95% Upper	Pct of Total
Chamber[Community]	0.0227175	486.86377	595.98604	-681.2474	1654.9749	2.221
Residual		21431.232	1441.7402	18864.108	24563.439	97.779
Total		21918.096	1531.1358	19201.481	25258.201	100.000

-2 LogLikelihood = 5984.5265908

Note: Total is the sum of the positive variance components.

Total including negative estimates = 21918.096

## ► Covariance Matrix of Variance Component Estimates

## ► Iterations

## ▼ Fixed Effect Tests

Source	Nparm	DF	DFDen	F Ratio	Prob > F
Community	2	2	8.53	1.4098	0.2958
Depth	1	1	445.2	0.4707	0.4930
DOC	1	1	362.8	9.4639	0.0023*
TDN	1	1	446.9	1.2059	0.2727
WT	1	1	111.8	1.0272	0.3130
daily_Tsoil_AC8_profile	1	1	441.3	0.8816	0.3483
Community*Depth	2	2	445.7	2.6941	0.0687
Community*DOC	2	2	406.3	1.4951	0.2254
Community*TDN	2	2	446.2	7.6141	0.0006*
Community*WT	2	2	82.23	0.5647	0.5707
Community*daily_Tsoil_AC8_profile	2	2	446	1.1937	0.3041
Depth*DOC	1	1	435	0.1967	0.6576
Depth*TDN	1	1	447	0.1031	0.7483
Depth*WT	1	1	444.2	9.1017	0.0027*
Depth*daily_Tsoil_AC8_profile	1	1	446.2	6.2161	0.0130*
DOC*TDN	1	1	446.7	6.5832	0.0106*
DOC*WT	1	1	444.3	0.8367	0.3608
DOC*daily_Tsoil_AC8_profile	1	1	445.6	0.8781	0.3492
TDN*WT	1	1	446.6	1.6476	0.1999
TDN*daily_Tsoil_AC8_profile	1	1	446.3	3.7093	0.0547
WT*daily_Tsoil_AC8_profile	1	1	446.3	0.1662	0.6837

Output of JMP mixed model analysis where the pore water CH<sub>4</sub> concentration (DOY 212-255) is treated as the response with chamber as a random effect and DOC, TDN, depth-specific peat temperature and WT as covariates, and depth and community as fixed effects.

Response CH4_PW_Wanninkhof			
Whole Model			
Effect Summary			
Source	LogWorth		PValue
Community*TDN	4.304		0.00005
TDN	2.206		0.00623
TDN*WT	1.916		0.01213
DOC*TDN	1.804		0.01571
Community	1.439		0.03640
TDN*daily_Tsoil_AC8_profile	0.740		0.18212
Community*DOC	0.594		0.25490
DOC*daily_Tsoil_AC8_profile	0.441		0.36264
DOC*WT	0.427		0.37370
Depth	0.394		0.40343
Community*daily_Tsoil_AC8_profile	0.354		0.44308
Depth*daily_Tsoil_AC8_profile	0.351		0.44606
daily_Tsoil_AC8_profile	0.296		0.50595
WT*daily_Tsoil_AC8_profile	0.225		0.59555
Depth*WT	0.108		0.77935
WT	0.108		0.77985
Community*WT	0.081		0.83010
Depth*DOC	0.066		0.85868
DOC	0.031		0.93113
Depth*TDN	0.017		0.96260
Community*Depth	0.017		0.96260
<a href="#">Remove</a> <a href="#">Add</a> <a href="#">Edit</a> <input type="checkbox"/> FDR			
Summary of Fit			
RSquare	0.307449		
RSquare Adj	0.219661		
Root Mean Square Error	148.2043		
Mean of Response	261.9299		
Observations (or Sum Wgts)	241		

## REML Variance Component Estimates

Random Effect	Var Ratio	Var Component	Std Error	95% Lower	95% Upper	Pct of Total
Chamber[Community]	-0.016774	-368.4243	468.19292	-1286.066	549.21696	0.000
Residual		21964.523	2150.2659	18291.801	26872.615	100.000
Total		21964.523	2150.2659	18291.801	26872.615	100.000

-2 LogLikelihood = 2944.9847446

Note: Total is the sum of the positive variance components.

Total including negative estimates = 21596.098

## Covariance Matrix of Variance Component Estimates

## Iterations

## Fixed Effect Tests

Source	Nparm	DF	DFDen	F Ratio	Prob > F
Community	2	2	9.669	4.7589	0.0364*
Depth	1	1	211.9	0.7009	0.4034
DOC	1	1	148	0.0075	0.9311
TDN	1	1	209.3	7.6374	0.0062*
WT	1	1	18.74	0.0804	0.7798
daily_Tsoil_AC8_profile	1	1	211.8	0.4439	0.5059
Community*Depth	2	2	212.4	0.0381	0.9626
Community*DOC	2	2	75.69	1.3919	0.2549
Community*TDN	2	2	193.6	10.4347	<.0001*
Community*WT	2	2	15.81	0.1884	0.8301
Community*daily_Tsoil_AC8_profile	2	2	202.5	0.8173	0.4431
Depth*DOC	1	1	212.3	0.0318	0.8587
Depth*TDN	1	1	211.8	0.0022	0.9626
Depth*WT	1	1	212.2	0.0787	0.7793
Depth*daily_Tsoil_AC8_profile	1	1	191.3	0.5831	0.4461
DOC*TDN	1	1	212.9	5.9300	0.0157*
DOC*WT	1	1	209.8	0.7947	0.3737
DOC*daily_Tsoil_AC8_profile	1	1	212.5	0.8323	0.3626
TDN*WT	1	1	212.7	6.4007	0.0121*
TDN*daily_Tsoil_AC8_profile	1	1	213	1.7919	0.1821
WT*daily_Tsoil_AC8_profile	1	1	209.5	0.2826	0.5956

### D.3 Mixed model analysis of factors affecting $\delta^{13}\text{C}$ -CH<sub>4</sub> source signatures

Output of JMP mixed model analysis where the  $\delta^{13}\text{C}$ -CH<sub>4</sub> source signature is treated as the response with chamber as a random effect and mean pore water CH<sub>4</sub> concentration, log CH<sub>4</sub> flux, 10 cm peat temperature and WT as covariates, and community as a fixed effect.

Response d13Ch4		
Whole Model		
Effect Summary		
Source	LogWorth	PValue
mean WT*mean CH4 PW wanninkhof	0.575	0.26584
mean CH4 PW wanninkhof*mean Tsoil 10 cm	0.344	0.45292
Community*mean CH4 PW wanninkhof	0.342	0.45474
mean CH4 PW wanninkhof	0.326	0.47248
mean Tsoil 10 cm	0.323	0.47520
mean WT*log Ch4 flux	0.264	0.54406
mean CH4 PW wanninkhof*log Ch4 flux	0.245	0.56859
mean WT	0.175	0.66821
mean WT*mean Tsoil 10 cm	0.147	0.71365
Community*mean WT	0.138	0.72845
Community	0.088	0.81605
Community*mean Tsoil 10 cm	0.084	0.82327
log Ch4 flux*mean Tsoil 10 cm	0.058	0.87430
log Ch4 flux	0.015	0.96521
Community*log Ch4 flux	0.000	0.99985
<a href="#">Remove</a> <a href="#">Add</a> <a href="#">Edit</a> <input type="checkbox"/> FDR		
Summary of Fit		
RSquare	0.404831	
RSquare Adj	0.114505	
Root Mean Square Error	4.83897	
Mean of Response	-75.2857	
Observations (or Sum Wgts)	62	

### REML Variance Component Estimates

Random Effect	Var Ratio	Var Component	Std Error	95% Lower	95% Upper	Pct of Total
Chamber[Community]	0.3836872	8.984279	11.203353	-12.97389	30.942447	27.729
Residual		23.415634	5.4230701	15.585278	39.105004	72.271
Total		32.399913	11.8092	17.696678	77.458618	100.000

-2 LogLikelihood = 390.27492445

Note: Total is the sum of the positive variance components.

Total including negative estimates = 32.399913

### Covariance Matrix of Variance Component Estimates

### Iterations

### Fixed Effect Tests

Source	Nparm	DF	DFDen	F Ratio	Prob > F
Community	2	2	5.668	0.2107	0.8161
mean WT	1	1	34.36	0.1869	0.6682
mean CH4 PW wanninkhof	1	1	39.34	0.5263	0.4725
log Ch4 flux	1	1	9.521	0.0020	0.9652
mean Tsoil 10 cm	1	1	31.58	0.5223	0.4752
Community*mean WT	2	2	39.87	0.3194	0.7285
Community*mean CH4 PW wanninkhof	2	2	38.47	0.8044	0.4547
Community*log Ch4 flux	2	2	9.335	0.0002	0.9998
Community*mean Tsoil 10 cm	2	2	31.45	0.1957	0.8233
mean WT*mean CH4 PW wanninkhof	1	1	37.96	1.2754	0.2658
mean WT*log Ch4 flux	1	1	40.68	0.3743	0.5441
mean WT*mean Tsoil 10 cm	1	1	37.94	0.1367	0.7136
mean CH4 PW wanninkhof*log Ch4 flux	1	1	38.15	0.3307	0.5686
mean CH4 PW wanninkhof*mean Tsoil 10 cm	1	1	37.97	0.5751	0.4529
log Ch4 flux*mean Tsoil 10 cm	1	1	40.43	0.0253	0.8743



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